

# Potential for shoreline recession to accelerate discharge of groundwater pollutants to coastal waters

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## Abstract

Discharge of groundwater-derived pollutants to inland and marine coastal waters is influenced by the transport and reactive processes occurring in nearshore aquifers. The effect of shoreline change on these processes and subsequent discharge of pollutants to coastal waters is unclear. The objective of this study was to evaluate the impact of shoreline recession (landward movement of the mean shoreline) on the transport of nitrogen [N] and phosphorus [P] in a nearshore aquifer and their discharge to coastal waters. Field investigations were conducted on a permeable unconfined nearshore aquifer on Lake Huron, Canada, in years coinciding with historically low and high lake water levels. At the site, a septic system-derived nutrient-rich (N and P) groundwater plume is moving towards the lake and the mean shoreline position moved ~30 m landward between sampling years due coastal erosion and mean lake water level increase. Data indicate PO<sub>4</sub>-P fluxes to the lake were higher following shoreline recession due to shortened travel pathways. In contrast, NO<sub>3</sub>-N fluxes were governed by the specific geochemical conditions near the sediment-water interface, which are not only a function of the shoreline position. Further, findings show shoreline recession may modify mineral phases that tend to sequester pollutants (e.g., iron oxides) near the sediment-water interface and this may possibly mediate release of sediment-bound pollutants. The findings provide new insights into potential impacts of shoreline change on chemical discharge to coastal waters as needed to inform long-term water quality predictions and management.

1 **Potential for shoreline recession to accelerate discharge of groundwater**  
2 **pollutants to coastal waters**

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14 **Key points:**

- 15
- 16 • Increase in inland coastal water level resulted in ~ 30 m shoreline recession
  - 17 • Shoreline recession accelerated PO<sub>4</sub>-P discharge due to reduced travel paths
  - 18 • Geochemical conditions near the shoreline governed fate and discharge of NO<sub>3</sub>-N
  - Shoreline recession modified location of “iron curtain”

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36 mediate release of sediment-bound pollutants. The findings provide new insights into  
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38 inform long-term water quality predictions and management.

39 **Keywords:** climate change, shoreline erosion, coastal water level change, nutrients, iron curtain,  
40 septic system

## 41    **1    Introduction**

42    Coastal zones are the most populated and developed areas worldwide. Currently over half  
43    of the global population resides in coastal areas and this is expected to increase in the future  
44    based on current trends of coastal migration and population growth (Small and Nicholls,  
45    2003). Sandy beaches, which occur along marine coastlines and along inland coastlines  
46    that are adjacent to large inland lakes, make up more than one-third of the ice-free global  
47    coastline (Luijendijk et al., 2018). Many sandy coastlines worldwide are threatened by the  
48    landward movement of the shoreline, known as shoreline recession. Shoreline recession,  
49    which is caused by increases in coastal water levels (i.e. sea level rise or lake level changes)  
50    combined with shoreline erosion, is expected to increase as the climate changes (e.g.,  
51    higher frequency and more intense storm events) and in response to anthropogenic  
52    development (Bird, 2008). Luijendijk et al. (2018) found from satellite image analysis that  
53    24% of the world’s sandy beaches are eroding at rates exceeding 0.5 m/yr, with only 48%  
54    of sandy beaches considered stable. For the Laurentian Great Lakes, which have more than  
55    6000 km of permeable shoreline (Government of Canada - Environment and Climate  
56    Change Canada and U.S. EPA, 2009), shoreline recession caused by interannual lake water  
57    level fluctuations combined with erosive storm events is an increasing challenge. Lake  
58    water level fluctuations up to one to two meters occur along large lacustrine shorelines,  
59    such as the Great Lakes, approximately once a decade (Argyilan and Forman, 2003;  
60    Sellinger et al., 2008).

61    It is well established that shoreline recession along marine and inland coastlines has severe  
62    economic, social and environmental consequences due to impacts including land and  
63    habitat loss, property damage, and reduced revenue potential from tourism and industry

64 (Mukhopadhyay et al., 2018; Theuerkauf et al., 2019). As such, this is an active area of  
65 research, with studies generally focused on monitoring shoreline changes (e.g. Burningham  
66 and French, 2017; García-Rubio et al., 2015; Hagenaaars et al., 2018), fundamental  
67 understanding of underlying geomorphological and hydrological processes (e.g. Brooks et  
68 al., 2017; Swenson et al., 2006), and techniques for forecasting shoreline change over short  
69 and long time scales (e.g. Davidson et al., 2017; Mukhopadhyay et al., 2018). Few studies  
70 have examined the potential impacts of shoreline recession, including sediment erosion and  
71 changes in coastal water levels, on the flux of land-derived pollutants to coastal waters.  
72 Two prior studies examined on the impact of shoreline erosion on nutrient loading to Lake  
73 Tahoe, United States (Adams and Minor, 2002) and Lake Diefenbaker Reservoir, Canada  
74 (Hewlett et al., 2015). Both studies found that nutrient loading caused by shoreline erosion  
75 (due to release of sediment-bound nutrients) only contributed a fraction of the total nutrient  
76 load into the lakes (< 5% of total nutrient input). More recently, beach sand erosion was  
77 found to be the main mechanism of transferring *E.coli* to nearshore coastal waters during  
78 intensified wave conditions at a sandy beaches (Vogel et al., 2016). Finally, focused on a  
79 coastal wetlands rather than sandy shorelines, Braun et al. (2019) showed that shoreline  
80 recession may reduce the carbon storage potential of a coastal wetland and thereby impact  
81 carbon dynamics within the wetland. The above-mentioned studies all focus on sediment-  
82 bound pollutants that are released to coastal waters as sediment near the shoreline is eroded  
83 and released to the coastal waters. While this can be an important transport mechanism,  
84 shoreline recession may also affect the delivery of dissolved pollutants into coastal waters,  
85 particularly land-derived groundwater pollutants, whose delivery to coastal waters strongly  
86 depends on the specific subsurface transport pathways between the land source and coastal

87 waters, as well as the geochemical conditions along these pathways in nearshore aquifers  
88 (Beldowska et al., 2022; Destouni et al., 2010). For instance, shoreline recession brings  
89 the shoreline closer to areas of human development (e.g. infrastructure, landfills, and  
90 contaminated sites), which decreases the distance between a groundwater pollution source  
91 and the coastal water. This may in turn increase the potential for groundwater-derived  
92 pollutants reaching coastal waters.

93 Submarine groundwater discharge (SGD) and lacustrine groundwater discharge (LGD) are  
94 important pathways for land-derived pollutants such as nutrients, metals and organic  
95 contaminants to enter marine and inland coastal waters, respectively (e.g. Beck et al., 2016;  
96 Lewandowski et al., 2015; Moore, 2010). With increased urbanization, industrialization  
97 and intensification of agriculture in coastal areas, groundwater pollutant concentrations can  
98 be high in coastal areas and pose a risk to coastal water pollution. The importance of LGD  
99 has mainly been recognized for its contribution to lake nutrient loads and lake  
100 eutrophication (Robinson, 2015; Rosenberry et al., 2015). For instance, high nutrient  
101 concentrations, more than ten times higher than in the adjacent lake water, were observed  
102 in groundwater discharging into Lake Simcoe (Roy and Malenica, 2013). In addition to  
103 nutrients, studies have also shown that LGD may play an important role in delivering other  
104 chemical constituents, including metals (Lee et al., 2014; Rakhimbekova et al., 2018) and  
105 road salt (Howard and Livingstone, 2000; Roy, 2019) to lakes. The transport of pollutants  
106 via groundwater flow paths to coastal waters is generally slow, and transport can be  
107 retarded by various physical and chemical processes (e.g. filtration, sorption,  
108 precipitation). Therefore, changes in the shoreline position may impact the groundwater  
109 travel times, travel distances and the specific pathways, by which pollutant are transported

110 from their land source to coastal waters. In addition, it is well established that an important  
111 reaction zone can exist in nearshore aquifers close to the shoreline due to the mixing of  
112 groundwater and coastal water. This zone, commonly referred to as a subterranean estuary  
113 in marine environments, is known to strongly influence the discharge of groundwater-  
114 derived pollutants to coastal waters due to its high reactivity (Rakhimbekova et al., 2021b,  
115 2021c; Robinson et al., 2018). One important reactive phenomenon in this mixing zone is  
116 the sequestration (trapping) of pollutants to sediment mineral phases, particularly iron (Fe)  
117 oxides, which form under certain redox and pH conditions. This sediment trap mechanism,  
118 commonly referred to as the “iron curtain”, can act as a natural barrier that limits the release  
119 of a range of pollutants, including phosphorus (Charette and Sholkovitz, 2002; Spiteri et  
120 al., 2008), carbon (Linkhorst et al., 2017; Sirois et al., 2018), metals (McAllister et al.,  
121 2015), and trace metals (Beck et al., 2008) to coastal waters. It is possible shoreline  
122 recession may cause a shift in the location of the mixing zone, thereby impacting reactivity  
123 and geochemical conditions, including the function of sediment trap and its ability to  
124 sequester pollutants. This may in turn modify pollutant fluxes delivered to coastal waters  
125 via SGD and LGD.

126 To the best of our knowledge, there are no studies that have explored the impact of  
127 shoreline changes, including shoreline recession, on the discharge of groundwater-derived  
128 pollutants to coastal waters. The objective of this study is to address this knowledge gap.  
129 Field data collected downgradient of a large septic system over a six-year period is used to  
130 examine the influence of shoreline recession on the subsurface transport of a P- and N-rich  
131 wastewater plume and its potential discharge to the adjacent lake. In addition, the study  
132 investigates the way in which shoreline recession may alter geochemical conditions near

133 the sediment-water interface with a specific focus on the abundance and distribution of  
134 mineral phases that tend to sequester pollutants, including P, close to the groundwater-  
135 coastal water interface. Finally, this study explores the broader possible impacts of  
136 shoreline recession beyond our specific study site by discussion of potential sources of  
137 anthropogenic groundwater contaminant sources along shorelines and future projections  
138 on lake water level changes and expected shoreline changes. The findings of this study are  
139 needed given the prevalence of groundwater contamination near marine and inland  
140 coastlines, and the projected changes in coastal water levels and storm intensity and  
141 frequency as the climate changes.

## 142 **2 Methodology**

### 143 **2.1 Field investigation**

#### 144 ***2.1.1 Site description***

145 Field investigations were conducted at Ipperwash Beach on Lake Huron, Ontario, Canada  
146 in 2014, 2015, and 2020 (Figure 1a). Ipperwash Beach is a sandy beach that extends over  
147 20 km and has an extensive dune system behind the beach that rises several meters above  
148 the lake surface (Eyles and Meulendyk, 2012). Extensive shoreline recession occurred at  
149 Ipperwash Beach between 2015 and 2020 in response to an increase in the water level in  
150 Lake Huron combined with beach erosion. In 2020, the water level in Lake Huron reached  
151 177.45 m (IGLD 85 datum) following over a decade of low water levels during which the  
152 record low water level of 175.57 m occurred in 2013 (Gronewold et al., 2021; Gronewold  
153 and Stow, 2014). The extent of shoreline recession between 2015 and 2020 are illustrated  
154 in aerial photographs shown in Figure 1b.

155 Ipperwash Beach was selected for this study as a public comfort station with a large septic  
156 system is located in close vicinity to the beach (Figure 1b). The septic system has been  
157 operational since at least 1974 and is only used during the warmer months from May to  
158 September. Due to shoreline recession, the mean shoreline position was located ~120 m  
159 from the septic system tile bed in 2015, but only ~90 m from the septic tile bed in 2020  
160 (Figure 1b). Detailed (approximately monthly) field sampling was conducted between  
161 August and October in 2014, and between May and August in 2015, to evaluate the spatial  
162 and temporal variability of phosphorus (P) and nitrogen (N) transport from the septic  
163 system through the nearshore aquifer to the lake. Details of this previous study are provided  
164 in Rakhimbekova et al. (2021a). This current study expands upon this prior work with new  
165 sampling performed in 2020 to determine the impact of shoreline recession on the transport  
166 of P and N through the nearshore aquifer and their potential discharge to the lake.

## 167 ***2.1.2 Field monitoring approach***

### 168 ***2.1.2.1 Field equipment***

169 For all sampling periods in 2014-2015 and 2020, a cross-shore monitoring transect was  
170 installed that aligned with the center of the septic system-derived nutrient-rich plume that  
171 is moving towards the lake. The original cross-shore monitoring transect in 2014-2015  
172 consisted of ten multi-level samplers (MLS) and extended from ~71 m downgradient of  
173 the septic tile bed ( $x = 71$  m; landward extent of the beach dunes) to ~20 m offshore ( $x =$   
174 140 m; Figure 1c). For the 2020 sampling, the cross-shore monitoring transect consisted of  
175 six MLS, and extended from  $x = 71$  m downgradient of septic tile bed ~20 m offshore ( $x =$   
176 110 m) (Figure 1c). The  $x$ - $z$  coordinate system used herein is defined with  $x = 0$  m located  
177 at the landward edge of the septic tile bed and  $z = 0$  as the elevation of a local permanent

178 benchmark. All MLS were installed up to 3 m below ground surface, with sampling ports  
179 at 0.2 m depth intervals.

180 Two piezometers (PZ) were permanently installed at  $x = 71$  m (PZ-1) and  $x = 93$  m (PZ-2)  
181 in July 2014, with additional piezometers temporarily installed along the cross-shore  
182 monitoring transect at the start of each sampling event to measure the groundwater and  
183 lake water levels across the nearshore aquifer and offshore. The location of monitoring  
184 equipment and ground surface elevations were surveyed with a total station (Topcon GTS-  
185 239W) relative to a local permanent benchmark during each sampling event (Figure 1c).

#### 186 **2.1.2.2 Electrical resistivity imaging**

187 The septic system origin of the nutrient-rich plume observed in the nearshore aquifer was  
188 confirmed in 2015 by using artificial sweeteners as a chemical tracer for human wastewater  
189 (Rakhimbekova et al., 2021a). However, the artificial sweetener data were limited to point  
190 information obtained from samples collected at selected MLS located over 70 m  
191 downgradient of the septic tile bed (i.e., samples collected from  $x = 71$  m onwards). During  
192 the field sampling in 2020, a non-invasive electrical resistivity tomography (ERT) survey  
193 was performed to provide continuous imaging of electrical resistivity within the subsurface  
194 and determine if this approach would be able to trace the nutrient-rich plume back to the  
195 septic tile beds.

196 An ERT survey line consisting of 96 stainless-steel electrodes (0.3 m) with an inline  
197 spacing of 1 m was deployed. The location of the line coincided with the location of the  
198 cross-shore monitoring transect but it extended from the landward edge of the septic tile  
199 bed ( $x = 0$  m) to the shoreline ( $x = 95$  m) (Figure 1b). A multi-channel Syscal Pro Switch

200 48 resistivity system (IRIS Instruments, France) was used to complete the survey. As this  
201 system is limited to 48 electrodes at a time, the roll-along method was used to record the  
202 survey line through three successive segments. For this, 24 electrodes were moved for each  
203 segment to ensure 50% overlap (i.e., electrode numbers: 0-48, 24-72, 48-96) (Robinson et  
204 al., 2022).

205 The multi-gradient electrode array was used with a total of 4183 measurements. Due to the  
206 dry surface sand, saline water was added around each electrode to ensure strong ground  
207 coupling (resistance: <1 kilo-ohm). The recorded apparent resistivity data was inverted  
208 with DC\_2DPro, an iterative least-squares smoothness-constrained inversion program  
209 (Kim, 2016). The surface elevation at each electrode, which was surveyed using the total  
210 station, was incorporated into the inversion, which was then run to a maximum of seven  
211 iterations, showing good convergence with a low root mean square (RMS) error of 5.5%.

### 212 ***2.1.2.3 Porewater sampling and analysis***

213 For all sampling events, pore water samples were first collected from all MLS ports using  
214 syringes attached to the PVC sampling tubes ( $\phi = 1.35$  mm). Porewater samples were  
215 collected from each port, filtered using 0.45  $\mu\text{m}$  cellulose acetate filters, and stored in 60  
216 mL HDPE bottles. Samples were analyzed for phosphate ( $\text{PO}_4\text{-P}$ ), nitrate ( $\text{NO}_3\text{-N}$ ) and  
217 chloride (Cl) within 48 hours of collection. After samples were collected, a peristaltic pump  
218 was used to draw pore water directly into a flow cell where pH, electrical conductivity (EC)  
219 and temperature were measured with an In-Situ Multi-Parameter TROLL® 9500 and YSI  
220 6600 V2 Sonde. Multiple nearshore lake water samples were also collected during each  
221 sampling event. Collected pore water samples were analyzed for nutrients ( $\text{PO}_4\text{-P}$ ,  $\text{NO}_3\text{-N}$ )

222 using Lachat 750 Flow Injection Analysis (Lachat 750), and Cl using High Performance  
223 Liquid Chromatography.

#### 224 **2.1.2.4 Sediment sampling and analysis**

225 During MLS installation in May 2015 and August 2020, multiple sediment samples were  
226 collected at depths of 0.5 m to 2 m below the ground surface for sediment chemistry  
227 characterization (Figure 1c). Sediment samples were analyzed via total acid digestion using  
228 method EPA 200.8 (US EPA, 1996) to determine the total elemental solid phase  
229 concentrations. In addition, due to the potential importance of Fe-bearing solid phases in  
230 sequestering chemical including P in nearshore aquifers, select sediment samples were  
231 treated with (i) acid ammonium oxalate solution (McKeague and Day, 1966), and (ii)  
232 dithionite-citrate-bicarbonate solution (Mehra, 1958) to provide characterization of the Fe-  
233 bearing solid phases (e.g. degree of crystallization). Oxalate-extractable Fe ( $Fe_o$ ) is  
234 operationally thought to represent amorphous oxides such as ferrihydrite and small  
235 amounts of organically-bound Fe (Schwertmann, 1964). Dithionite-extractable Fe ( $Fe_d$ ) is  
236 operationally thought to represent organically-bound oxides, amorphous oxides, and finely  
237 crystalline oxides (Akinbola et al., 2013). The difference between  $Fe_d$  and  $Fe_o$  generally  
238 represents the amount of solid phase Fe present in crystalline oxide forms (e.g., goethite  
239 and hematite). Total dissolved phosphorus (TDP) and total iron (Fe) in extractant solutions  
240 were analyzed using HACH DR 6000 UV VIS Spectrophotometer and inductively-coupled  
241 plasma mass spectroscopy (ICP-MS), respectively. Additional details on sediment sample  
242 analysis are provided in the Supporting Information (SI Section 1).

243        **2.2    Numerical model**

244    A two-dimensional (2D) model was developed in MODFLOW-NWT (Niswonger et al.,  
245    2011) to simulate steady state groundwater flows and estimate NO<sub>3</sub>-N and PO<sub>4</sub>-P loads  
246    discharging to the lake along the vertical cross-shore section. The model domain and  
247    parameterization were based on a groundwater model of the study site developed and  
248    validated by Malott et al. (2016). The two-dimensional model assumes negligible  
249    groundwater flow in the alongshore direction. The model was 150 m long in the cross-  
250    shore direction and the aquifer thickness at the shoreline location was set to 13 m at the  
251    landward boundary due to presence of a clay layer at this depth according to local well  
252    records (Figure 2). The beach face and lake boundaries (AB and BC) were specified based  
253    on topographic sand level surveys conducted during each sampling period. The model was  
254    discretized into 71 layers and 128 columns with greater refinement around the shoreline  
255    and the beach face. Grid discretization tests were performed to ensure a converged solution.  
256    Particle size analysis performed on sediment samples ( $n = 7$ ) collected across the nearshore  
257    aquifer up to 2.5 m below the water table indicated that the nearshore aquifer is comprised  
258    of relatively homogeneous fine sand (Coefficient of Uniformity,  $C_u = 2.18$  and sand grain  
259    size,  $d_{50} = 0.16$  mm) with the saturated  $K$  estimated to range from 7 to 15 m/d using the  
260    method of Krumbein and Monk (1943). Based on this, the simulated aquifer was  
261    homogeneous and isotropic with the hydraulic conductivity ( $K$ ) set to 10 m/d and effective  
262    porosity  $n = 0.3$ .

263    The upper (AB) and bottom (ED) model boundaries were no-flow boundaries assuming  
264    negligible aerial recharge and evaporation, and an impermeable aquifer basement,  
265    respectively. The offshore boundary (CD) was specified as no-flow boundary and was

266 located sufficiently far from the shoreline (>50 m offshore) that it did not influence the  
267 groundwater flows in the nearshore aquifer (Figure 2). The vertical landward boundary  
268 (AE) was located at  $x = 60$  m, which is 10 m inland from the landward extent of the beach  
269 dunes (Figure 2) and was set to be a constant head boundary. Three simulations were run  
270 with the boundary conditions modified to represent the mean hydrological conditions at  
271 the study site during the 2014, 2015 and 2020 sampling periods. Based on the measured  
272 groundwater levels, the hydraulic head at the landward boundary was set to -2.95 m, -3.05  
273 m and -1.95 m, respectively, for the models simulating the groundwater flows over the  
274 sampling periods in 2014, 2015 and 2020. A constant head condition was used along the  
275 lake boundary (BC) with the specified head values set based on the mean measured lake  
276 water level during each of the sampling periods and considering the effect of steady waves  
277 acting on the boundary. To simulate the effect of steady waves, the approach of Anwar et  
278 al. (2014) was adopted whereby the time-averaged effect of waves as described by wave  
279 set-up was used (details are provided in SI Section 2). For this, the empirical equation of  
280 Nielsen (2009) was used to estimate the wave-set up profile and calculate head values that  
281 were then prescribed along the lake boundary (BC). Offshore root mean square wave  
282 heights ( $H_{rms}$ ) of 0.35 m, 0.25 m and 0.35 m were used in the equation of Nielsen (2009)  
283 to simulate the mean wave conditions over the sampling periods in 2014, 2015 and 2020,  
284 respectively. The mean  $H_{rms}$  values were calculated using monitoring data from a wave  
285 buoy located ~35 km from the study site ( C45149 Southern Lake Huron, Department of  
286 Fisheries and Oceans Canada Buoy, 2020). Based on the mean lake water levels measured  
287 at the study site during the sampling periods in 2014, 2015 and 2020, the still lake water  
288 level in the models were set to -3.64 m, -3.73 m and -2.21 m, respectively.

## 289 **3 Results and discussion**

### 290 **3.1 Field and modeling results**

#### 291 **3.1.1 *Change in hydrological conditions between sampling periods***

292 The hydrological conditions at the study site including the mean lake water level, mean  
293 wave conditions, and landward groundwater levels varied between the sampling periods.  
294 The mean lake water level was approximately 0.7 m higher in 2020 compared to 2014-  
295 2015 (Figure 3a). More intensified wave conditions were also observed in 2020 compared  
296 to 2014-2015 with maximum wave height ( $H_{rms}$ ) reaching 3.7 m in 2020 compared to 3.0  
297 m in 2014 and 2.1 m in 2015 (mean  $H_{rms}$  was 0.35 m, 0.25 m and 0.35 m in 2020, 2015 and  
298 2014, respectively; Figure 3b). The extent of shoreline recession between 2015 and 2020  
299 is evident by comparing the topographical surveys performed in 2015 and 2020, along with  
300 corresponding aerial photographs (Figures 1b and 1c). The shoreline was estimated to  
301 move 30 m landward from  $x = 120$ - $125$  m in 2014-2015, to  $x = 90$  m in 2020, in response  
302 to the lake water level increase and beach sand erosion.

303 Increase in lake water levels may lead to a corresponding increase in the landward  
304 groundwater table elevation, and thereby reduce the vertical separation between septic tile  
305 beds and groundwater table. This can be a concern as it then limits the efficiency of the  
306 septic tile bed in attenuating wastewater constituents including nutrients (Mitchell et al.,  
307 2021). Based on groundwater levels measured at PZ-1 ( $x = 71$  m), there was  $\sim 1$  m increase  
308 in the groundwater table between 2014-2015 and 2020 (Figure 1c). However, with high  
309 sand dunes behind the beach, there is a thick (3 - 4 m) vadose zone below the septic tile  
310 beds at this study site (see Section 3.1.2 for details), and therefore it is not expected that  
311 the higher groundwater table elevation in 2020 compared to 2014 – 2015 would have

312 adversely impacted the function of the vadose zone below the septic tile beds in attenuating  
313 wastewater constituents. The adverse impact of increasing coastal levels on the attenuation  
314 of wastewater constituents below the septic tile beds is expected to be a larger concern in  
315 coastal plain settings with flatter topography and shallow groundwater tables.

316 The mean hydraulic gradient across nearshore aquifer was negative in 2014-2015 ( $h = -$   
317  $0.015$ ) and 2020 ( $h = -0.017$ ), indicating lakeward groundwater flow. Based on the  
318 hydraulic gradient in 2014-2015 and 2020, mean hydraulic conductivity ( $K = 10$  m/d) and  
319 effective porosity ( $n = 0.3$ ) for the nearshore aquifer, the mean horizontal groundwater  
320 velocity towards the lake is estimated to be 0.50 m/d in 2014-2015 and 0.56 m/d in 2020.  
321 Numerical simulations show the change in groundwater flow patterns (Figure 4a-c) in the  
322 nearshore aquifer between the 2014-2015 and 2020 sampling periods with the groundwater  
323 discharge zone shifting from  $x = 120-125$  m in 2014-2015 to  $x = 85-90$  m in 2020. Based  
324 on the estimated overall mean horizontal velocities and shoreline positions, the travel time  
325 for conservative species to be transported from the septic tile bed to the shoreline is  
326 estimated to be around 240 days in 2014-2015 compared to only 160 days in 2020.

327 While the flow patterns simulated by the steady state groundwater models do not indicate  
328 recirculation of lake water through the nearshore aquifer (Figure 4a-c), recirculating flows  
329 driven mostly by changing lake water levels and wave conditions are expected to occur  
330 landward of the mean shoreline position and simulated groundwater discharge zone (Malott  
331 et al., 2016). With the mean shoreline position and zone of groundwater discharge shifting  
332 landward in 2020 compared to 2014-2015, the recirculation zone, which is generally  
333 associated with high reactivity including the presence of mineral phases (e.g., Fe oxides)  
334 that tend sequester pollutants (Lee et al., 2014; Lewandowski et al., 2015; Rakhimbekova

335 et al., 2018), would also have shifted landward. It is possible that this landward shift in the  
336 recirculation zone could have altered geochemical conditions in the nearshore aquifer and  
337 this is explored in the following sections.

### 338 **3.1.2 Characterization of septic-derived nutrient plume**

339 Elevated groundwater PO<sub>4</sub>-P and NO<sub>3</sub>-N concentrations were observed at the study site in  
340 2014, 2015 and 2020 (Figures 4d to 4i). Groundwater PO<sub>4</sub>-P concentrations ranged from 8  
341 to 166 µg/L in 2014-2015, and from 3 to 98 µg/L in 2020. Maximum PO<sub>4</sub>-P concentrations  
342 in the nearshore aquifer exceeded the upper limit of the Canadian Water Quality Guidelines  
343 for the Protection of Aquatic Life for Total Phosphorus to limit eutrophic conditions (35–  
344 100 µg P/L; Canadian Council of Ministers of the Environment, 2014), and were  
345 considerably higher than the mean measured lake water concentrations for all sampling  
346 periods (2-3 µg/L). Groundwater NO<sub>3</sub>-N concentrations ranged from 4 to 57 mg/L in 2014-  
347 2015 and from 1 to 26 mg/L in 2020. Maximum NO<sub>3</sub>-N concentrations in the nearshore  
348 aquifer exceeded the Canadian Water Quality Guidelines for the Protection of Aquatic Life  
349 for all sampling periods (3 mg/L; Canadian Council of Ministers of the Environment,  
350 2014), and were also higher than the mean measured lake water concentrations (0.2-0.8  
351 mg/L).

352 Figures 4j-1 to 4m-o presents the distributions of chloride (Cl) concentrations and electrical  
353 conductivity (EC) measured along the monitoring transect in 2014, 2015 and 2020. Cl and  
354 EC can be considered conservative wastewater tracers at the site(Rakhimbekova et al.,  
355 2021a), and their distributions generally coincide with the locations of elevated PO<sub>4</sub>-P and  
356 NO<sub>3</sub>-N in the nearshore aquifer Figure 4d-i. NO<sub>3</sub>-N with EC landward of the maximum  
357 shoreline position are correlated (Figure S1) indicating the suitability of using ERT for

358 characterizing the spatial extent of the high nutrient groundwater plume. Figure 5 shows a  
359 2D cross-sectional ERT image that extends from  $x = 0$  m (septic tile bed) to  $x = 95$  m (near  
360 the mean shoreline position in 2020). The location of the septic tile bed, PZ-1 and MLS are  
361 indicated in the figure, along with the lake water and groundwater levels measured during  
362 the survey. The highly resistive near surface layer between  $x = 0$  m and  $x = 78$  m (red-  
363 purple;  $>1000$  ohm-m) represents the vadose zone with a mean thickness of approximately  
364 4 m. The underlying layer of low resistivity (blue-green;  $<150$  ohm-m) represents the zone  
365 of saturated groundwater, with some areas having very low resistivity ( $<50$  ohm-m). The  
366 ERT image can be validated by comparing bulk resistivity values with the porewater EC  
367 measurements in Figure 4m-o. For example, the porewater EC at  $x = 71$  m and  $z = -3$  m  
368 (PZ-1) is  $1008 \mu\text{S}/\text{cm}$  which suggests a bulk EC of  $209 \mu\text{S}/\text{cm}$ , or bulk resistivity of  $48$   
369 ohm-m (Archie, 1942) – this closely matches the bulk resistivity inferred by ERT at that  
370 location ( $55$  ohm-m). Figure 5 indicates that the zone of very low bulk resistivity is  
371 connected to the base of the septic tile bed and extends lakeward. This zone exists through  
372 the shallow nearshore aquifer and becomes shallower near the shoreline meeting the  
373 sediment-water interface around  $x = 92$  m. This ERT image clearly illustrates that the  
374 elevated  $\text{NO}_3\text{-N}$  and  $\text{PO}_4\text{-P}$  concentrations observed near the shoreline originate from the  
375 septic tile bed.

### 376 **3.1.3 Impact of shoreline recession on $\text{PO}_4\text{-P}$**

377 As seen in Figure 4, the highest  $\text{PO}_4\text{-P}$  concentrations were observed 73-90 m lakeward of  
378 the septic tile bed (maximum  $\text{PO}_4\text{-P} = 166 \mu\text{g}/\text{L}$ ;  $x = 73\text{-}90$  m) in 2014-2015, with  
379 concentrations decreasing closer to the shoreline (particularly in 2015). However, elevated  
380 concentrations of the conservative wastewater tracers, Cl and EC, near the shoreline ( $x =$

381 115-125 m) for these sampling periods suggest that the septic-derived wastewater plume  
382 has reached the shoreline. Taking into account septic system operation time, groundwater  
383 velocity, and length of the PO<sub>4</sub>-P plume in 2015, Rakhimbekova et al. (2021) concluded  
384 that PO<sub>4</sub>-P transport in the nearshore aquifer was mainly governed by retardation. Based  
385 on the location of the PO<sub>4</sub>-P plume in 2015, it was predicted that PO<sub>4</sub>-P may possibly reach  
386 the shoreline in approximately 10 years. However, this prediction is incorrect due to  
387 shoreline recession at the site. Figure 4f shows that in 2020 (within <5 years), the PO<sub>4</sub>-P  
388 plume has reached the new shoreline location ( $x = 90$  m). To provide a quantitative estimate  
389 on the extent to which PO<sub>4</sub>-P fluxes to the lake may have changed, the measured PO<sub>4</sub>-P  
390 concentrations were combined with the simulated groundwater flows for each sampling  
391 period. PO<sub>4</sub>-P fluxes were calculated by multiplying PO<sub>4</sub>-P concentrations (interpolated to  
392 a depth of 3 m below ground surface) along vertical line near the shoreline ( $x = 120.3$  m in  
393 2014 and 2015 and  $x = 88.7$  m in 2020; Figure 4 a-c) by simulated groundwater fluxes  
394 directed towards the lake for each sampling year (Additional details provided in SM S3).  
395 From these calculations, it is estimated that the PO<sub>4</sub>-P flux to the lake may have increased  
396 nearly tenfold from 53 mg/d/m in 2015 to 482 mg/d/m in 2020 (Figure 6a) due higher PO<sub>4</sub>-  
397 P concentrations near the shoreline in 2020 compared to near the shoreline in 2015 as well  
398 as slightly higher lakeward-directed groundwater velocity (Figure 4e, f). This result  
399 illustrates that the shoreline recession can accelerate the discharge of groundwater-derived  
400 pollutants that are slowly advancing towards the shoreline by decreasing the subsurface  
401 travel distance and time.

#### 402        **3.1.4 Impact of shoreline recession on NO<sub>3</sub>-N**

403        The NO<sub>3</sub>-N distribution in the nearshore aquifer was influenced by not only its shorter  
404        discharge pathway through the nearshore aquifer due to shoreline recession but also by the  
405        specific geochemical conditions near the sediment-water interface that impact reactive  
406        transformations that NO<sub>3</sub>-N may undergo prior to its discharge. Rakhimbekova et al.  
407        (2021a) previously showed that the extent of NO<sub>3</sub>-N removal that occurs in the nearshore  
408        aquifer prior to discharge may vary over time as the extent of denitrification changes in  
409        response to changing geochemical conditions near the sediment-water interface (e.g., due  
410        to changes in organic matter availability). Rakhimbekova et al. (2021a) presented detailed  
411        data analysis showing changing organic matter availability was associated with the  
412        observed decrease in NO<sub>3</sub>-N concentrations near the shoreline in 2015 (0-2 mg/L)  
413        compared to 2014 (4-67 mg/L), and thus lower estimated NO<sub>3</sub>-N flux towards the lake in  
414        2015 (4 mg/d/m) compared to 2014 (124 mg/d/m) (Figure 6b).

415        NO<sub>3</sub>-N concentrations observed near the mean shoreline position in 2020 ( $x = 85-90$  m)  
416        ranged from 4 to 26 mg/L in 2020 (Figure 4i). The estimated NO<sub>3</sub>-N flux directed towards  
417        the lake near the shoreline in 2020 (79 mg/d/m) was larger than in 2015 (4 mg/d/m) but  
418        smaller than in 2014 (124 mg/d/m). NO<sub>3</sub>-N flux is affected by both groundwater velocity  
419        and NO<sub>3</sub>-N concentrations. Since groundwater velocity near the shoreline was slightly  
420        higher in 2020 compared to 2014 (difference in flow arrow sizes in Figure 4 a-c), the higher  
421        estimated NO<sub>3</sub>-N flux in 2014 is due to higher NO<sub>3</sub>-N concentrations near the shoreline in  
422        2014 compared to 2020. Therefore, even though shoreline recession reduced the travel  
423        distance and travel time for the septic-derived NO<sub>3</sub>-N to reach the lake, the specific  
424        geochemical conditions in the nearshore aquifer near the shoreline ultimately govern the

425 flux of NO<sub>3</sub>-N to the lake rather than the overall travel distance and time. This finding is  
426 in contrast to the highly retarded PO<sub>4</sub>-P plume for which shoreline recession was found to  
427 directly accelerate PO<sub>4</sub>-P discharge to the lake due to the overall reduced travel distance  
428 between the septic tile bed and shoreline.

429 It is important to note that while the geochemical conditions near the shoreline have a  
430 controlling influence on the fate of NO<sub>3</sub>-N in the nearshore aquifer and its flux to the lake  
431 at our study site, this may not be the case at other sites. The nearshore aquifer at our study  
432 site (from the septic tile bed to shoreline) is predominantly oxic and therefore limited  
433 denitrification occurs as NO<sub>3</sub>-N is transported towards the shoreline. At other sites where  
434 the nearshore aquifer is more reducing, it is possible that the decrease in overall travel  
435 distance and times could limit the extent of denitrification that occurs as NO<sub>3</sub>-N is  
436 transported from the source to the coastal water (Meile et al., 2010). Further, it is possible  
437 that shoreline recession could have a greater impact on the discharge of reactive pollutants  
438 in nearshore aquifers with sharper geochemical (redox and pH) gradients than observed at  
439 Ipperwash Beach (Rakhimbekova et al., 2021a, 2018).

#### 440 ***3.1.5 Impact of shoreline recession on Fe curtail processes***

441 In addition to modifying the physical transport pathways and travel times for the septic-  
442 derived nutrients being transported towards to the lake, the change in shoreline location  
443 may also modify the geochemical conditions near the groundwater-lake interface. In  
444 particular, it may modify the distribution, abundance and reactivity of mineral phases that  
445 can sequester pollutants in the nearshore aquifer. Rakhimbekova et al. (2021b) showed  
446 elevated solid phase Fe and P exists in the nearshore aquifer at the study site and suggested  
447 that PO<sub>4</sub>-P derived from the decomposition of lake-derived organic matter may be

448 accumulating on Fe oxide minerals that exist close to the sediment-water interface.  
449 Previous studies in marine environments have similarly found that Fe oxides that form near  
450 the sediment-water interface can act as a geochemical barrier (i.e. “iron curtain”) with these  
451 minerals sequestering pollutants such as PO<sub>4</sub>-P and limiting their release to coastal  
452 waters(Charette and Sholkovitz, 2002; Spiteri et al., 2006). While it is thought that  
453 sequestered pollutants may later be re-mobilized if geochemical or hydrological conditions  
454 change (Nisbeth et al., 2019; Spiteri et al., 2006), in this section we examine how the  
455 distribution and abundance of solid phase P and Fe, including Fe oxides, in the nearshore  
456 aquifer may have changed in response to shoreline recession.

457 Rakhimbekova et al. (201b) previously showed that nearshore aquifer at the study site is  
458 predominantly oxidizing, and there is no distinct redox gradient near the sediment-water  
459 interface. Further, Fe oxides are available through the nearshore aquifer and tend to  
460 accumulate PO<sub>4</sub>-P on their mineral surfaces. Characterization of the distribution of solid  
461 phase P and Fe in the nearshore aquifer at the study site in 2015 and 2020 indicate a  
462 decrease in solid phase P and Fe between these times (Figure 7a-d). Due to the landward  
463 movement of the shoreline, sediment samples were taken in different locations in 2015  
464 compared with 2020. To enable comparison of P and Fe solid phase concentrations  
465 between two times, comparison only considers samples collected over the same area,  
466 between  $x = 87$  and 117 m (marked with red square on Figure 7). Generally, there was  
467 higher abundance of solid phase P near the shoreline in 2015 compared to 2020 ( $151 \pm 51$   
468  $\mu\text{g/g}$  in 2015,  $n = 17$ ;  $82 \pm 55 \mu\text{g/g}$  in 2020,  $n = 23$ ; Figure 7a and b). Solid phase Fe was  
469 also slightly higher in 2015 compared to 2020 ( $3.1 \pm 0.6 \text{ mg/g}$  in 2015,  $n = 17$ ;  $2.7 \pm 0.7$   
470  $\text{mg/g}$  in 2020,  $n = 23$ ; Figure 7c and d).

471 Selective solid phase extractions were conducted to characterize Fe-bearing solid phases  
472 (amorphous, Fe<sub>o</sub> and more crystalline, Fe<sub>d</sub>) at the study site in 2015 and 2020. Again, only  
473 sediment samples collected between  $x = 87$  and 117 m were used to compare the  
474 distribution of amorphous and crystalline Fe phases in 2015 and 2020. The abundance of  
475 Fe<sub>o</sub> decreased between 2015 and 2020 ( $0.5 \pm 0.2$  mg/g in 2015,  $n = 10$ ;  $0.2 \pm 0.09$  mg/g in  
476 2020,  $n = 13$ ; Figure 7e and f). In contrast, Fe<sub>d</sub> was slightly higher in 2020 compared to  
477 2015 ( $0.5 \pm 0.07$  mg/g in 2015;  $0.7$  mg/g  $\pm 0.3$  in 2020; Figure 7g vs 7h). Lower Fe<sub>d</sub> in  
478 2015 compared to 2020 may possibly be due to the formation of more crystalline Fe phases  
479 over time. Larger difference in the abundance of Fe<sub>o</sub> compared to Fe<sub>d</sub> between 2015 and  
480 2020 suggest that shoreline recession may have larger impact on amorphous Fe oxides  
481 compared to crystalline Fe oxides. This is not unexpected given that amorphous Fe oxides  
482 are less stable and shoreline recession may have altered factors that affect Fe oxide stability  
483 (e.g. oxygen availability, presence of bacteria) (Charette and Sholkovitz, 2002; Dean et al.,  
484 2003). Following shoreline recession, particularly a large erosive event, it might take time  
485 for amorphous Fe oxides to reform around the new shoreline location. This may explain  
486 the lower Fe<sub>o</sub> near the new shoreline location in 2020 compared to higher Fe<sub>o</sub> near the  
487 location of shoreline in 2015, which was a more stable shoreline position over the  
488 preceding years.

489 Amorphous Fe (e.g. ferrihydrite) tend to have a higher surface area that is more reactive  
490 and therefore has a higher capacity to sequester pollutants, including PO<sub>4</sub>-P, compared to  
491 more crystalline Fe oxides fractions such as goethite and hematite (Fuller et al., 1993).  
492 Therefore, lower abundance of Fe<sub>o</sub> in the nearshore aquifer in 2020 (following the shoreline  
493 recession) suggests that mineral surfaces along the groundwater discharge pathway may

494 have less ability to sequester PO<sub>4</sub>-P, and this may possibly lead to higher PO<sub>4</sub>-P fluxes to  
495 lake. The impact of shoreline recession on the distribution and abundance of Fe oxide  
496 minerals, particularly amorphous Fe oxides, is important as it may affect the function of  
497 these minerals in sequestering and releasing pollutants thereby modifying pollutant fluxes  
498 to coastal waters.

### 499 **3.2 Broader implications of the impact of shoreline recession**

500 The study findings demonstrate the potential impacts of shoreline recession driven by  
501 coastal water level changes and/or erosion on the transport and fate of groundwater  
502 pollutants and their discharge to adjacent coastal waters. The findings show shoreline  
503 recession may impact pollutants moving slowly along groundwater pathways towards the  
504 coastal water, as well as pollutants for which their fate and ultimate delivery to coastal  
505 waters is governed strongly by the geochemical conditions in the reaction zone that exists  
506 near the groundwater-coastal water interface. This section explores the broader  
507 implications of the study findings on the discharge of groundwater pollutants to the  
508 Laurentian Great Lakes, as well as other inland and marine coastal waters.

509 While it is known that water levels in the Great Lakes vary over 20 – 30 year cycles  
510 (Gronewold and Rood, 2019), recent future projections estimate that by 2040-2049 average  
511 annual water levels of Lake Superior, Lake Michigan-Huron and Lake Erie are projected  
512 to increase by +0.19, +0.44, +0.28 m, respectively due to an increase in over-lake  
513 precipitation and basin runoff compared to relatively smaller increase in lake evaporation  
514 (Kayastha et al., 2022). While there are no studies predicting future erosion rates and  
515 shoreline recession in response to the projected water level changes for the Great Lakes,  
516 several studies show that high water levels correspond with accelerated shoreline erosion,

517 including bluff erosion, on short time scales (e.g. Kaczmarek et al., 2016; Krueger et al.,  
518 2020). For instance, Castedo et al. (2013) simulated erosion rates to range between 0.5-1.5  
519 m/yr in response to lake water level variations of 0.5-1 m. A more recent study by Volpano  
520 et al. (2020) on coastal bluff erosion along Western Lake Michigan's shoreline found that  
521 erosion rates increased from 0.18-0.43 m/yr between 2009 and 2014 (period with below  
522 long term average water levels) to 0.49-1.19 m/yr between 2014 and 2018 (period with  
523 above long-term average water levels).

524 The outlook for marine shorelines is also concerning. On average, global sea level has risen  
525 at a rate of 3-4 mm/yr in the last 20 years (Yi et al., 2015). Future projections estimate sea  
526 level rise from 0.1-0.2 m by 2050 and from 0.3-2 m by 2100 due to ocean warming and  
527 land-based ice melt (Kopp et al., 2014; Vitousek et al., 2017). While shoreline recession is  
528 generally the combined result of various contributing factors and may be complicated by  
529 local and regional differences, it is widely acknowledged that increasing sea levels will  
530 promote shoreline recession (e.g. Anderson et al., 2015; Bruun, 1962; Dean and Houston,  
531 2016). For instance, a recent study predicted that 15% of world's sandy beaches could face  
532 severe coastal erosion by 2050 and up to 50% by the end of the century (Vousdoukas et  
533 al., 2020). Taking into consideration the projected sea level changes and associated  
534 shoreline recession, together with the findings of our study, there is high potential that  
535 future shoreline recession will modify inputs of groundwater pollutants to coastal waters.

536 While this study focuses on the wastewater plume from one septic system, there are 1.2  
537 and 1.3 million septic systems in Ontario, Canada (DeRabbie, 2020) and Michigan, United  
538 States (Department of Environment, Great Lakes and Energy, 2022), respectively, with  
539 many of these septic systems located in close proximity to the shorelines of the Great Lake.

540 It is also important to note that in addition to nutrients, septic-derived wastewater effluent  
541 also contains other pollutants of concern including pharmaceuticals, personal care products  
542 and pathogens, all of which may pose a risk to coastal water pollution with associated  
543 human health and ecosystem impacts. Further, many other groundwater contaminant  
544 sources are often located in coastal areas whereby shoreline recession may modify and  
545 potentially accelerate pollutant inputs to coastal waters. This includes landfills (both  
546 historic and currently operating) (Brand et al., 2018; Stefania et al., 2019) as well as  
547 industrial sites including Superfund sites which are the most contaminated sites in the  
548 United States (Carter, 2020).

#### 549 **4 Conclusions**

550 This study showed the impact of shoreline recession in modifying the fate of groundwater  
551 pollutants discharging to coastal waters at a specific study site, and discussed the broader  
552 potential implications for inland and marine coastal waters. The study findings show that  
553 the shoreline recession may accelerate the discharge of groundwater pollutants to coastal  
554 waters by decreasing the subsurface travel distance and travel time. In addition, shoreline  
555 recession may modify reactive processes close to the sediment-water interface, which may  
556 in turn affect the fate of reactive pollutants and their discharge to coastal waters.  
557 Importantly, shoreline recession may result in a landward shift of the recirculation zone  
558 that commonly exists close to the shoreline, and in doing so, alter the stability and  
559 abundance of sediment phases (e.g., Fe oxides) that often act to sequester pollutants and  
560 potentially limit their release to coastal waters. In particular, the study findings show that  
561 amorphous Fe oxides may be particularly impacted by the shoreline recession and may

562 take time to reform and function to sequester pollutants along the modified groundwater  
563 discharge pathway.

564 As it is expected that climate change will accelerate shoreline recession on coastlines  
565 worldwide due to increase in the magnitude of lake water level fluctuations, sea level rise,  
566 as well as increased frequency and intensity of storm events, it is recommended that future  
567 work evaluate the impact of shoreline changes of different time scales (e.g., event based,  
568 yearly, decadal) on pollutant fate in nearshore aquifers. The nearshore aquifer at the study  
569 site was relatively homogeneous and predominantly oxidizing and it is recommended that  
570 future studies evaluate the impact of shoreline recession in different nearshore aquifer types  
571 including more heterogeneous aquifers and those with more reducing conditions. Further,  
572 while this study focused on a freshwater coastal aquifer, field investigations are required  
573 to determine the impact of shoreline recession on the functioning of the reaction zone in  
574 nearshore aquifers along marine shorelines (i.e., subterranean estuary) and the  
575 corresponding impact on pollutant fluxes to marine waters. As the study data also  
576 demonstrate that the abundance and form of solid phase Fe near the shoreline may vary in  
577 response to shoreline change, there is a need to better understand the mineralization process  
578 for Fe oxides to better predict the impact of future changes on their functioning especially  
579 with respect to sequestering and releasing pollutants of concern. While this study focused  
580 on the impact of shoreline recession on nutrients derived from a septic system, it is  
581 recommended that future studies explore the potential impacts of shoreline recession on  
582 the discharge of other anthropogenic pollutants of concern that are commonly elevated in  
583 coastal groundwater systems.

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588 **References**

- 589 Adams, K.D., Minor, T.B., 2002. Historic shoreline change at Lake Tahoe from 1938 to 1998  
590 and its impact on sediment and nutrient loading. *J. Coast. Res.* 18, 637–651.
- 591 Akinbola, G.E., Adigun, M.O., Aduroja, O., 2013. Dithionite and oxalate extraction of iron and  
592 manganese in some basement complex soils of southwestern Nigeria. *J. Exp. Sci.* 4, 22–26
- 593 Anderson, T.R., Fletcher, C.H., Barbee, M.M., Frazer, L.N., Romine, B.M., 2015. Doubling of  
594 coastal erosion under rising sea level by mid-century in Hawaii. *Nat. Hazards* 78, 75–103.  
595 <https://doi.org/10.1007/s11069-015-1698-6>
- 596 Anwar, N., Robinson, C., Barry, D.A., 2014. Influence of tides and waves on the fate of nutrients  
597 in a nearshore aquifer: Numerical simulations. *Adv. Water Resour.* 73, 203–213.  
598 <https://doi.org/10.1016/j.advwatres.2014.08.015>
- 599 Argyilan, E.P., Forman, S.L., 2003. Lake level response to seasonal climatic variability in the  
600 Lake Michigan-Huron system from 1920 to 1995. *J. Great Lakes Res.* 29, 488–500.  
601 [https://doi.org/10.1016/S0380-1330\(03\)70453-5](https://doi.org/10.1016/S0380-1330(03)70453-5)
- 602 Beck, A.J., Kellum, A.A., Luek, J.L., Cochran, M.A., 2016. Chemical flux associated with  
603 spatially and temporally variable submarine groundwater discharge, and chemical  
604 modification in the subterranean estuary at Gloucester Point, VA (USA). *Estuaries and*  
605 *Coasts* 39, 1–12. <https://doi.org/10.1007/s12237-015-9972-0>
- 606 Beck, M., Dellwig, O., Schnetger, B., Brumsack, H.J., 2008. Cycling of trace metals (Mn, Fe,  
607 Mo, U, V, Cr) in deep pore waters of intertidal flat sediments. *Geochim. Cosmochim. Acta*  
608 72, 2822–2840. <https://doi.org/10.1016/j.gca.2008.04.013>

609 Bełdowska, M., Bełdowski, J., Kwasigroch, U., Szubska, M., Jędruch, A., 2022. Coastal cliff  
610 erosion as a source of toxic, essential and nonessential metals in the marine environment.  
611 *Oceanologia*. <https://doi.org/10.1016/j.oceano.2022.04.001>

612 Bird, E.C., 2008. *Coastal geomorphology*, Second Edition. John Wiley & Sons Ltd. West  
613 Sussex. England.

614 Brand, J.H., Spencer, K.L., O'shea, F.T., Lindsay, J.E., 2018. Potential pollution risks of historic  
615 landfills on low-lying coasts and estuaries. *WIREs Water* 5, 1–12.  
616 <https://doi.org/10.1002/wat2.1264>

617 Braun, K.N., Theuerkauf, E.J., Masterson, A.L., Curry, B.B., Horton, D.E., 2019. Modeling  
618 organic carbon loss from a rapidly eroding freshwater coastal wetland. *Sci. Rep.* 9, 1–13.  
619 <https://doi.org/10.1038/s41598-019-40855-5>

620 Brooks, S.M., Spencer, T., Christie, E.K., 2017. Storm impacts and shoreline recovery:  
621 Mechanisms and controls in the southern North Sea. *Geomorphology* 283, 48–60.  
622 <https://doi.org/10.1016/j.geomorph.2017.01.007>

623 Bruun, P., 1962. Sea-level rise as a cause of shore erosion. *J. Waterw. Harb. Div.* 88.  
624 <https://doi.org/10.1061/JWHEAU.0000252>

625 Burningham, H., French, J., 2017. Understanding coastal change using shoreline trend analysis  
626 supported by cluster-based segmentation. *Geomorphology* 282, 131–149.  
627 <https://doi.org/10.1016/j.geomorph.2016.12.029>

628 Canadian Council of Ministers of the Environment, 2014. *Canadian Water Quality Guidelines*  
629 *for the Protection of Aquatic Life*.

630 Carter, Jacob, Casey Kalman. 2020. A toxic relationship: Extreme coastal flooding and  
631 Superfund sites. Cambridge, MA: Union of Concerned Scientists.  
632 <https://www.ucsusa.org/resources/toxic-relationship>

633 Castedo, R., Fernández, M., Trenhaile, A.S., Paredes, C., 2013. Modeling cyclic recession of  
634 cohesive clay coasts: Effects of wave erosion and bluff stability. *Mar. Geol.* 335, 162–176.  
635 <https://doi.org/10.1016/j.margeo.2012.11.001>

636 Charette, M.A., Sholkovitz, E.R., 2002. Oxidative precipitation of groundwater-derived ferrous  
637 iron in the subterranean estuary of a coastal bay. *Geophys. Res. Lett.* 29, 85-1-85–4.  
638 <https://doi.org/10.1029/2001gl014512>

639 Davidson, M.A., Turner, I.L., Splinter, K.D., Harley, M.D., 2017. Annual prediction of shoreline  
640 erosion and subsequent recovery. *Coast. Eng.* 130, 14–25.  
641 <https://doi.org/10.1016/j.coastaleng.2017.09.008>

642 Dean, R.G., Houston, J.R., 2016. Determining shoreline response to sea level rise. *Coast. Eng.*  
643 114, 1–8. <https://doi.org/10.1016/j.coastaleng.2016.03.009>

644 Dean, W.E., Neff, B.P., Rosenberry, D.O., Winter, T.C., Parkhurst, R., 2003. The significance of  
645 groundwater to the accumulation of iron and manganese in the sediments of two  
646 hydrologically distinct lakes in North-Central Minnesota: a geological perspective.  
647 *GroundWater* 41, 951–963. <https://doi.org/10.1111/j.1745-6584.2003.tb02437.x>

648 Department of Environment, Great Lakes and Energy, 2022. Septic Smart  
649 [https://www.michigan.gov/egle/about/organization/drinking-water-and-environmental-](https://www.michigan.gov/egle/about/organization/drinking-water-and-environmental-health/onsite-wastewater-management/septicmart)  
650 [health/onsite-wastewater-management/septicmart](https://www.michigan.gov/egle/about/organization/drinking-water-and-environmental-health/onsite-wastewater-management/septicmart)

651 Department of Fisheries and Oceans Canada Buoy, 2014. C45149 Southern Lake Huron  
652 <http://www.meds-sdmm.dfo-mpo.gc.ca/isdm-gdsi/waves-vagues/data-donnees/>  
653 [index-eng.asp](http://www.meds-sdmm.dfo-mpo.gc.ca/isdm-gdsi/waves-vagues/data-donnees/index-eng.asp)

654 DeRabbie, D., 2020. Septic Systems and Homeowner Insurance. Ontario Onsite Wastewater  
655 Association Newsletter, 15, 1.  
656 [https://www.oowa.org/wp-content/uploads/2020/05/R\\_Vol15\\_Iss1.pdf](https://www.oowa.org/wp-content/uploads/2020/05/R_Vol15_Iss1.pdf)

657 Destouni, G., Persson, K., Prieto, C., Jarsjö, J., 2010. General quantification of catchment-scale  
658 nutrient and pollutant transport through the subsurface to surface and coastal waters.  
659 *Environ. Sci. Technol.* 44, 2048–2055. <https://doi.org/10.1021/es902338y>

660 Eyles, N., Meulendyk, T., 2012. Ground-penetrating radar stratigraphy and depositional model  
661 for evolving Late Holocene aeolian dunes on the Lake Huron coast, Ontario. *J. Great Lakes*  
662 *Res.* 38, 708–719. <https://doi.org/10.1016/j.jglr.2012.09.003>

663 Fuller, C.C., Dadis, J.A., Waychunas, G.A., 1993. Surface chemistry of ferrihydrite: Part 2.  
664 Kinetics of arsenate adsorption and coprecipitation. *Geochim. Cosmochim. Acta* 57, 2271–  
665 2282. [https://doi.org/10.1016/0016-7037\(93\)90568-H](https://doi.org/10.1016/0016-7037(93)90568-H)

666 García-Rubio, G., Huntley, D., Russell, P., 2015. Evaluating shoreline identification using  
667 optical satellite images. *Mar. Geol.* 359, 96–105.  
668 <https://doi.org/10.1016/j.margeo.2014.11.002>

669 Government of Canada - Environment and Climate Change Canada, U.S. EPA, 2009. Nearshore  
670 areas of the Great Lakes. <https://doi.org/978-1-100-13563-2>

671 Gronewold, A.D., Do, H.X., Mei, Y., Stow, C.A., 2021. A tug-of-war within the hydrologic

672 cycle of a continental freshwater basin. *Geophys. Res. Lett.* 48, 1–8.  
673 <https://doi.org/10.1029/2020GL090374>

674 Gronewold, A.D., Rood, R.B., 2019. Recent water level changes across Earth’s largest lake  
675 system and implications for future variability. *J. Great Lakes Res.* 45, 1–3.  
676 <https://doi.org/10.1016/j.jglr.2018.10.012>

677 Gronewold, A.D., Stow, C.A., 2014. Water loss from the Great Lakes. *Science* (80-. ). 343,  
678 1084–1085. <https://doi.org/10.1126/science.1249978>

679 Hageaars, G., de Vries, S., Luijendijk, A.P., de Boer, W.P., Reniers, A.J.H.M., 2018. On the  
680 accuracy of automated shoreline detection derived from satellite imagery: A case study of  
681 the sand motor mega-scale nourishment. *Coast. Eng.* 133, 113–125.  
682 <https://doi.org/10.1016/j.coastaleng.2017.12.011>

683 Hewlett, C., North, R.L., Johansson, J., Vandergucht, D.M., Hudson, J.J., 2015. Contribution of  
684 shoreline erosion to nutrient loading of the Lake Diefenbaker reservoir, Saskatchewan,  
685 Canada. *J. Great Lakes Res.* 41, 110–117. <https://doi.org/10.1016/j.jglr.2014.11.020>

686 Howard, K.W.F., Livingstone, S., 2000. Transport of urban contaminants into Lake Ontario via  
687 sub-surface flow. *Urban Water* 2, 183–195.  
688 [https://doi.org/10.1016/S1462-0758\(00\)00058-3](https://doi.org/10.1016/S1462-0758(00)00058-3)

689 Kaczmarek, H., Mazaeva, O.A., Kozyreva, E.A., Babicheva, V.A., Tyszkowski, S., Rybchenko,  
690 A.A., Brykała, D., Bartczak, A., Słowiński, M., 2016. Impact of large water level  
691 fluctuations on geomorphological processes and their interactions in the shore zone of a  
692 dam reservoir. *J. Great Lakes Res.* 42, 926–941. <https://doi.org/10.1016/j.jglr.2016.07.024>

693 Kayastha, M.B., Ye, X., Huang, C., Xue, P., 2022. Future rise of the Great Lakes water levels  
694 under climate change. *J. Hydrol.* 612, 128205.  
695 <https://doi.org/10.1016/j.jhydrol.2022.128205>

696 Kopp, R.E., Horton, R.M., Little, C.M., Mitrovica, J.X., Oppenheimer, M., Rasmussen, D.J.,  
697 Strauss, B.H., Tebaldi, C., 2014. Probabilistic 21st and 22nd century sea-level projections at  
698 a global network of tide-gauge sites. *Earth's Futur.* 2, 383–406.  
699 <https://doi.org/10.1002/2014ef000239>

700 Krueger, R., Zoet, L.K., Rawling, J.E., 2020. Coastal Bluff Evolution in Response to a Rapid  
701 Rise in Surface Water Level. *J. Geophys. Res. Earth Surf.* 125, 1–16.  
702 <https://doi.org/10.1029/2019JF005428>

703 Lee, J., Robinson, C., Couture, R.M., 2014. Effect of groundwater-lake interactions on arsenic  
704 enrichment in freshwater beach aquifers. *Environ. Sci. Technol.* 48, 10174–10181.  
705 <https://doi.org/10.1021/es5020136>

706 Lewandowski, J., Meinikmann, K., Nuetzmann, G., Rosenberry, D.O., 2015. Groundwater - the  
707 disregarded component in lake water and nutrient budgets. Part 2: Effects of groundwater  
708 on nutrients. *Hydrol. Process.* 29, 2922–2955. <https://doi.org/10.1002/hyp.10384>

709 Linkhorst, A., Dittmar, T., Waska, H., 2017. Molecular Fractionation of Dissolved Organic  
710 Matter in a Shallow Subterranean Estuary: The Role of the Iron Curtain. *Environ. Sci.*  
711 *Technol.* 51, 1312–1320. <https://doi.org/10.1021/acs.est.6b03608>

712 Luijendijk, A., Hagenaars, G., Ranasinghe, R., Baart, F., Donchyts, G., Aarninkhof, S., 2018.  
713 The State of the World's Beaches. *Sci. Rep.* 8, 1–11. [https://doi.org/10.1038/s41598-018-](https://doi.org/10.1038/s41598-018-24630-6)  
714 [24630-6](https://doi.org/10.1038/s41598-018-24630-6)

715 Malott, S., O'Carroll, D.M., Robinson, C.E., 2016. Dynamic groundwater flows and  
716 geochemistry in a sandy nearshore aquifer over a wave event. *Water Resour. Res.* 52, 5248–  
717 5264. <https://doi.org/10.1002/2014WR015716>

718 McAllister, S.M., Barnett, J.M., Heiss, J.W., Findlay, A.J., MacDonald, D.J., Dow, C.L., Luther,  
719 G.W., Michael, H.A., Chan, C.S., 2015. Dynamic hydrologic and biogeochemical processes  
720 drive microbially enhanced iron and sulfur cycling within the intertidal mixing zone of a  
721 beach aquifer. *Limnol. Oceanogr.* 60, 329–345. <https://doi.org/10.1111/lno.10029>

722 McKeague, J.A., Day, J.H., 1966. Dithionite and oxalate extractable Fe and Al as aids in  
723 differentiating various classes of soil. *Can. J. Soil Sci.* 46, 13–22.  
724 <https://doi.org/10.4141/cjss66-003>

725 Mehra, O.P., 1958. Iron oxide removal from soils and clays by a dithionite-citrate system  
726 buffered with sodium bicarbonate. *Clays Clay Miner.* 7, 317–327.  
727 <https://doi.org/10.1346/ccmn.1958.0070122>

728 Meile, C., Porubsky, W.P., Walker, R.L., Payne, K., 2010. Natural attenuation of nitrogen  
729 loading from septic effluents: Spatial and environmental controls. *Water Res.* 44, 1399–  
730 1408. <https://doi.org/10.1016/j.watres.2009.11.019>

731 Mitchell, M., Isdell, R.E., Herman, J., Tombleson, C., 2021. Impact Assessment and  
732 Management Challenges of Key Rural Human Health Infrastructure Under Sea Level Rise.  
733 *Front. Mar. Sci.* 8, 1–8. <https://doi.org/10.3389/fmars.2021.631757>

734 Moore, W.S., 2010. The effect of submarine groundwater discharge on the ocean. *Ann. Rev.*  
735 *Mar. Sci.* 2, 59–88. <https://doi.org/10.1146/annurev-marine-120308-081019>

736 Mukhopadhyay, A., Ghosh, P., Chanda, A., Ghosh, A., Ghosh, S., Das, S., Ghosh, T., Hazra, S.,  
737 2018. Threats to coastal communities of Mahanadi delta due to imminent consequences of  
738 erosion – Present and near future. *Sci. Total Environ.* 637–638, 717–729.  
739 <https://doi.org/10.1016/j.scitotenv.2018.05.076>

740 Nielsen, P., 2009. *Coastal and Estuarine Processes*. World Scientific, Singapore.  
741 <https://doi.org/10.1142/7114>

742 Nisbeth, C.S., Jessen, S., Bennike, O., Kidmose, J., Reitzel, K., 2019. Role of groundwater-borne  
743 geogenic phosphorus for the internal P release in shallow lakes. *Water (Switzerland)* 11, 1–  
744 16. <https://doi.org/10.3390/w11091783>

745 Niswonger, R.G., Panday, S., Motomu, I., 2011. MODFLOW-NWT , A Newton formulation for  
746 MODFLOW-2005. USGS reports 44.

747 Rakhimbekova, S., O’Carroll, D.M., Andersen, M.S., Wu, M.Z., Robinson, C.E., 2018. Effect of  
748 transient wave forcing on the behavior of arsenic in a nearshore aquifer. *Environ. Sci.*  
749 *Technol.* 52, 12338–12348. <https://doi.org/10.1021/acs.est.8b03659>

750 Rakhimbekova, S., O’Carroll, D.M., Oldfield, L.E., Ptacek, C.J., Robinson, C.E., 2021a.  
751 Spatiotemporal controls on septic system derived nutrients in a nearshore aquifer and their  
752 discharge to a large lake. *Sci. Total Environ.* 752.  
753 <https://doi.org/10.1016/j.scitotenv.2020.141262>

754 Rakhimbekova, S., O’Carroll, D.M., Robinson, C.E., 2021b. Factors controlling phosphorus  
755 mobility in nearshore aquifers adjacent to large lakes. *Sci. Total Environ.* 799, 149443.  
756 <https://doi.org/10.1016/j.scitotenv.2021.149443>

757 Rakhimbekova, S., O'Carroll, D.M., Robinson, C.E., 2021c. Occurrence of arsenic in nearshore  
758 aquifers adjacent to large inland lakes. *Environ. Sci. Technol.*  
759 <https://doi.org/10.1021/acs.est.1c02326>

760 Robinson, C., 2015. Review on groundwater as a source of nutrients to the Great Lakes and their  
761 tributaries. *J. Great Lakes Res.* 41, 941–950. <https://doi.org/10.1016/j.jglr.2015.08.001>

762 Robinson, C.E., Xin, P., Santos, I.R., Charette, M.A., Li, L., Barry, D.A., 2018. Groundwater  
763 dynamics in subterranean estuaries of coastal unconfined aquifers: Controls on submarine  
764 groundwater discharge and chemical inputs to the ocean. *Adv. Water Resour.* 115, 315–331.  
765 <https://doi.org/10.1016/j.advwatres.2017.10.041>

766 Rosenberry, D.O., Lewandowski, J., Meinikmann, K., Nützmänn, G., 2015. Groundwater - the  
767 disregarded component in lake water and nutrient budgets . Part 1 : effects of groundwater  
768 on hydrology. *Hydrol. Process.* 2921, 2895–2921. <https://doi.org/10.1002/hyp.10403>

769 Roy, J.W., 2019. Endobenthic organisms exposed to chronically high chloride from groundwater  
770 discharging along freshwater urban streams and lakeshores. *Environ. Sci. Technol.* 53,  
771 9389–9397. <https://doi.org/10.1021/acs.est.9b02288>

772 Roy, J.W., Malenica, A., 2013. Nutrients and toxic contaminants in shallow groundwater along  
773 Lake Simcoe urban shorelines. *Inl. Waters.* <https://doi.org/10.5268/IW-3.2.521>

774 Schwertmann, U., 1964. Differenzierung der eisenoxide des bodens durch extraktion mit  
775 ammonium-losung. <https://doi.org/10.1002/jpln.3591050303>

776 Sellinger, C.E., Stow, C.A., Lamon, E.C., Qian, S.S., 2008. Recent water level declines in the  
777 Lake Michigan-Huron system. *Environ. Sci. Technol.* 42, 367–373.

778 <https://doi.org/10.1021/es070664+>

779 Sirois, M., Couturier, M., Barber, A., Gélinas, Y., Chaillou, G., 2018. Interactions between iron  
780 and organic carbon in a sandy beach subterranean estuary. *Mar. Chem.* 202, 86–96.  
781 <https://doi.org/10.1016/j.marchem.2018.02.004>

782 Small, C., Nicholls, R.J., 2003. A global analysis of human settlement in coastal zones. *J. Coast.*  
783 *Res.* 19, 584–599.

784 Spiteri, C., Regnier, P., Slomp, C.P., Charette, M.A., 2006. pH-Dependent iron oxide  
785 precipitation in a subterranean estuary. *J. Geochemical Explor.* 88, 399–403.  
786 <https://doi.org/10.1016/j.gexplo.2005.08.084>

787 Spiteri, C., Slomp, C.P., Charette, M.A., Tuncay, K., Meile, C., 2008. Flow and nutrient  
788 dynamics in a subterranean estuary (Waquoit Bay, MA, USA): Field data and reactive  
789 transport modeling. *Geochim. Cosmochim. Acta* 72, 3398–3412.  
790 <https://doi.org/10.1016/j.gca.2008.04.027>

791 Stefania, G.A., Rotiroti, M., Buerge, I.J., Zanotti, C., Nava, V., Leoni, B., Fumagalli, L.,  
792 Bonomi, T., 2019. Identification of groundwater pollution sources in a landfill site using  
793 artificial sweeteners, multivariate analysis and transport modeling. *Waste Manag.* 95, 116–  
794 128. <https://doi.org/10.1016/j.wasman.2019.06.010>

795 Swenson, M.J., Wu, C.H., Edil, T.B., Mickelson, D.M., 2006. Bluff recession rates and wave  
796 impact along the Wisconsin coast of Lake Superior. *J. Great Lakes Res.* 32, 512–530.  
797 [https://doi.org/10.3394/0380-1330\(2006\)32\[512:BRRAWI\]2.0.CO;2](https://doi.org/10.3394/0380-1330(2006)32[512:BRRAWI]2.0.CO;2)

798 Theuerkauf, E.J., Braun, K.N., Nelson, D.M., Kaplan, M., Vivirito, S., Williams, J.D., 2019.

799 Coastal geomorphic response to seasonal water-level rise in the Laurentian Great Lakes: An  
800 example from Illinois Beach State Park, USA. *J. Great Lakes Res.* 45, 1055–1068.  
801 <https://doi.org/10.1016/j.jglr.2019.09.012>

802 US EPA, 1996. Method 3050B Acid digestion of sediments, sludges and soils, Method 3050B  
803 Acid digestion of sediments, sludges and soils.

804 Vitousek, S., Barnar, P.L., Fletcher, C.H., Frazer, N., Storlazzi, C.D., 2017. Doubling of coastal  
805 flooding frequency within decades due to sea-level rise 1–9. [https://doi.org/10.1038/s41598-](https://doi.org/10.1038/s41598-017-01362-7)  
806 [017-01362-7](https://doi.org/10.1038/s41598-017-01362-7)

807 Vogel, L.J., O’Carroll, D.M., Edge, T.A., Robinson, C.E., 2016. Release of *Escherichia coli* from  
808 Foreshore Sand and Pore Water during Intensified Wave Conditions at a Recreational  
809 Beach. *Environ. Sci. Technol.* 50, 5676–5684. <https://doi.org/10.1021/acs.est.6b00707>

810 Vousdoukas, M.I., Ranasinghe, R., Mentaschi, L., Plomaritis, T.A., Athanasiou, P., Luijendijk,  
811 A., Feyen, L., 2020. Sandy coastlines under threat of erosion. *Nat. Clim. Chang.* 10, 260–  
812 263. <https://doi.org/10.1038/s41558-020-0697-0>

813 Yi, S., Sun, W., Heki, K., Qian, A., 2015. An increase in the rate of global mean sea level rise  
814 since 2010. *Geophys. Res. Lett.* 42, 3998–4006. <https://doi.org/10.1002/2015GL063902>  
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1     **Potential for shoreline recession to accelerate discharge of groundwater**  
2                                     **pollutants to coastal waters**

3

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14     **Key points:**

- 15             • Increase in inland coastal water level resulted in ~ 30 m shoreline recession
- 16             • Shoreline recession accelerated PO<sub>4</sub>-P discharge due to reduced travel paths
- 17             • Geochemical conditions near the shoreline governed fate and discharge of NO<sub>3</sub>-N
- 18             • Shoreline recession modified location of “iron curtain”

19 **Abstract**

20 Discharge of groundwater-derived pollutants to inland and marine coastal waters is  
21 influenced by the transport and reactive processes occurring in nearshore aquifers. The  
22 effect of shoreline change on these processes and subsequent discharge of pollutants to  
23 coastal waters is unclear. The objective of this study was to evaluate the impact of shoreline  
24 recession (landward movement of the mean shoreline) on the transport of nitrogen [N] and  
25 phosphorus [P] in a nearshore aquifer and their discharge to coastal waters. Field  
26 investigations were conducted on a permeable unconfined nearshore aquifer on Lake  
27 Huron, Canada, in years coinciding with historically low and high lake water levels. At the  
28 site, a septic system-derived nutrient-rich (N and P) groundwater plume is moving towards  
29 the lake and the mean shoreline position moved ~30 m landward between sampling years  
30 due coastal erosion and mean lake water level increase. Data indicate PO<sub>4</sub>-P fluxes to the  
31 lake were higher following shoreline recession due to shortened travel pathways. In  
32 contrast, NO<sub>3</sub>-N fluxes were governed by the specific geochemical conditions near the  
33 sediment-water interface, which are not only a function of the shoreline position. Further,  
34 findings show shoreline recession may modify mineral phases that tend to sequester  
35 pollutants (e.g., iron oxides) near the sediment-water interface and this may possibly  
36 mediate release of sediment-bound pollutants. The findings provide new insights into  
37 potential impacts of shoreline change on chemical discharge to coastal waters as needed to  
38 inform long-term water quality predictions and management.

39 **Keywords:** climate change, shoreline erosion, coastal water level change, nutrients, iron curtain,  
40 septic system

## 41    **1    Introduction**

42    Coastal zones are the most populated and developed areas worldwide. Currently over half  
43    of the global population resides in coastal areas and this is expected to increase in the future  
44    based on current trends of coastal migration and population growth (Small and Nicholls,  
45    2003). Sandy beaches, which occur along marine coastlines and along inland coastlines  
46    that are adjacent to large inland lakes, make up more than one-third of the ice-free global  
47    coastline (Luijendijk et al., 2018). Many sandy coastlines worldwide are threatened by the  
48    landward movement of the shoreline, known as shoreline recession. Shoreline recession,  
49    which is caused by increases in coastal water levels (i.e. sea level rise or lake level changes)  
50    combined with shoreline erosion, is expected to increase as the climate changes (e.g.,  
51    higher frequency and more intense storm events) and in response to anthropogenic  
52    development (Bird, 2008). Luijendijk et al. (2018) found from satellite image analysis that  
53    24% of the world’s sandy beaches are eroding at rates exceeding 0.5 m/yr, with only 48%  
54    of sandy beaches considered stable. For the Laurentian Great Lakes, which have more than  
55    6000 km of permeable shoreline (Government of Canada - Environment and Climate  
56    Change Canada and U.S. EPA, 2009), shoreline recession caused by interannual lake water  
57    level fluctuations combined with erosive storm events is an increasing challenge. Lake  
58    water level fluctuations up to one to two meters occur along large lacustrine shorelines,  
59    such as the Great Lakes, approximately once a decade (Argyilan and Forman, 2003;  
60    Sellinger et al., 2008).

61    It is well established that shoreline recession along marine and inland coastlines has severe  
62    economic, social and environmental consequences due to impacts including land and  
63    habitat loss, property damage, and reduced revenue potential from tourism and industry

64 (Mukhopadhyay et al., 2018; Theuerkauf et al., 2019). As such, this is an active area of  
65 research, with studies generally focused on monitoring shoreline changes (e.g. Burningham  
66 and French, 2017; García-Rubio et al., 2015; Hagenaaars et al., 2018), fundamental  
67 understanding of underlying geomorphological and hydrological processes (e.g. Brooks et  
68 al., 2017; Swenson et al., 2006), and techniques for forecasting shoreline change over short  
69 and long time scales (e.g. Davidson et al., 2017; Mukhopadhyay et al., 2018). Few studies  
70 have examined the potential impacts of shoreline recession, including sediment erosion and  
71 changes in coastal water levels, on the flux of land-derived pollutants to coastal waters.  
72 Two prior studies examined on the impact of shoreline erosion on nutrient loading to Lake  
73 Tahoe, United States (Adams and Minor, 2002) and Lake Diefenbaker Reservoir, Canada  
74 (Hewlett et al., 2015). Both studies found that nutrient loading caused by shoreline erosion  
75 (due to release of sediment-bound nutrients) only contributed a fraction of the total nutrient  
76 load into the lakes (< 5% of total nutrient input). More recently, beach sand erosion was  
77 found to be the main mechanism of transferring *E.coli* to nearshore coastal waters during  
78 intensified wave conditions at a sandy beaches (Vogel et al., 2016). Finally, focused on a  
79 coastal wetlands rather than sandy shorelines, Braun et al. (2019) showed that shoreline  
80 recession may reduce the carbon storage potential of a coastal wetland and thereby impact  
81 carbon dynamics within the wetland. The above-mentioned studies all focus on sediment-  
82 bound pollutants that are released to coastal waters as sediment near the shoreline is eroded  
83 and released to the coastal waters. While this can be an important transport mechanism,  
84 shoreline recession may also affect the delivery of dissolved pollutants into coastal waters,  
85 particularly land-derived groundwater pollutants, whose delivery to coastal waters strongly  
86 depends on the specific subsurface transport pathways between the land source and coastal

87 waters, as well as the geochemical conditions along these pathways in nearshore aquifers  
88 (Beldowska et al., 2022; Destouni et al., 2010). For instance, shoreline recession brings  
89 the shoreline closer to areas of human development (e.g. infrastructure, landfills, and  
90 contaminated sites), which decreases the distance between a groundwater pollution source  
91 and the coastal water. This may in turn increase the potential for groundwater-derived  
92 pollutants reaching coastal waters.

93 Submarine groundwater discharge (SGD) and lacustrine groundwater discharge (LGD) are  
94 important pathways for land-derived pollutants such as nutrients, metals and organic  
95 contaminants to enter marine and inland coastal waters, respectively (e.g. Beck et al., 2016;  
96 Lewandowski et al., 2015; Moore, 2010). With increased urbanization, industrialization  
97 and intensification of agriculture in coastal areas, groundwater pollutant concentrations can  
98 be high in coastal areas and pose a risk to coastal water pollution. The importance of LGD  
99 has mainly been recognized for its contribution to lake nutrient loads and lake  
100 eutrophication (Robinson, 2015; Rosenberry et al., 2015). For instance, high nutrient  
101 concentrations, more than ten times higher than in the adjacent lake water, were observed  
102 in groundwater discharging into Lake Simcoe (Roy and Malenica, 2013). In addition to  
103 nutrients, studies have also shown that LGD may play an important role in delivering other  
104 chemical constituents, including metals (Lee et al., 2014; Rakhimbekova et al., 2018) and  
105 road salt (Howard and Livingstone, 2000; Roy, 2019) to lakes. The transport of pollutants  
106 via groundwater flow paths to coastal waters is generally slow, and transport can be  
107 retarded by various physical and chemical processes (e.g. filtration, sorption,  
108 precipitation). Therefore, changes in the shoreline position may impact the groundwater  
109 travel times, travel distances and the specific pathways, by which pollutant are transported

110 from their land source to coastal waters. In addition, it is well established that an important  
111 reaction zone can exist in nearshore aquifers close to the shoreline due to the mixing of  
112 groundwater and coastal water. This zone, commonly referred to as a subterranean estuary  
113 in marine environments, is known to strongly influence the discharge of groundwater-  
114 derived pollutants to coastal waters due to its high reactivity (Rakhimbekova et al., 2021b,  
115 2021c; Robinson et al., 2018). One important reactive phenomenon in this mixing zone is  
116 the sequestration (trapping) of pollutants to sediment mineral phases, particularly iron (Fe)  
117 oxides, which form under certain redox and pH conditions. This sediment trap mechanism,  
118 commonly referred to as the “iron curtain”, can act as a natural barrier that limits the release  
119 of a range of pollutants, including phosphorus (Charette and Sholkovitz, 2002; Spiteri et  
120 al., 2008), carbon (Linkhorst et al., 2017; Sirois et al., 2018), metals (McAllister et al.,  
121 2015), and trace metals (Beck et al., 2008) to coastal waters. It is possible shoreline  
122 recession may cause a shift in the location of the mixing zone, thereby impacting reactivity  
123 and geochemical conditions, including the function of sediment trap and its ability to  
124 sequester pollutants. This may in turn modify pollutant fluxes delivered to coastal waters  
125 via SGD and LGD.

126 To the best of our knowledge, there are no studies that have explored the impact of  
127 shoreline changes, including shoreline recession, on the discharge of groundwater-derived  
128 pollutants to coastal waters. The objective of this study is to address this knowledge gap.  
129 Field data collected downgradient of a large septic system over a six-year period is used to  
130 examine the influence of shoreline recession on the subsurface transport of a P- and N-rich  
131 wastewater plume and its potential discharge to the adjacent lake. In addition, the study  
132 investigates the way in which shoreline recession may alter geochemical conditions near

133 the sediment-water interface with a specific focus on the abundance and distribution of  
134 mineral phases that tend to sequester pollutants, including P, close to the groundwater-  
135 coastal water interface. Finally, this study explores the broader possible impacts of  
136 shoreline recession beyond our specific study site by discussion of potential sources of  
137 anthropogenic groundwater contaminant sources along shorelines and future projections  
138 on lake water level changes and expected shoreline changes. The findings of this study are  
139 needed given the prevalence of groundwater contamination near marine and inland  
140 coastlines, and the projected changes in coastal water levels and storm intensity and  
141 frequency as the climate changes.

## 142 **2 Methodology**

### 143 **2.1 Field investigation**

#### 144 **2.1.1 Site description**

145 Field investigations were conducted at Ipperwash Beach on Lake Huron, Ontario, Canada  
146 in 2014, 2015, and 2020 (Figure 1a). Ipperwash Beach is a sandy beach that extends over  
147 20 km and has an extensive dune system behind the beach that rises several meters above  
148 the lake surface (Eyles and Meulendyk, 2012). Extensive shoreline recession occurred at  
149 Ipperwash Beach between 2015 and 2020 in response to an increase in the water level in  
150 Lake Huron combined with beach erosion. In 2020, the water level in Lake Huron reached  
151 177.45 m (IGLD 85 datum) following over a decade of low water levels during which the  
152 record low water level of 175.57 m occurred in 2013 (Gronewold et al., 2021; Gronewold  
153 and Stow, 2014). The extent of shoreline recession between 2015 and 2020 are illustrated  
154 in aerial photographs shown in Figure 1b.

155 Ipperwash Beach was selected for this study as a public comfort station with a large septic  
156 system is located in close vicinity to the beach (Figure 1b). The septic system has been  
157 operational since at least 1974 and is only used during the warmer months from May to  
158 September. Due to shoreline recession, the mean shoreline position was located ~120 m  
159 from the septic system tile bed in 2015, but only ~90 m from the septic tile bed in 2020  
160 (Figure 1b). Detailed (approximately monthly) field sampling was conducted between  
161 August and October in 2014, and between May and August in 2015, to evaluate the spatial  
162 and temporal variability of phosphorus (P) and nitrogen (N) transport from the septic  
163 system through the nearshore aquifer to the lake. Details of this previous study are provided  
164 in Rakhimbekova et al. (2021a). This current study expands upon this prior work with new  
165 sampling performed in 2020 to determine the impact of shoreline recession on the transport  
166 of P and N through the nearshore aquifer and their potential discharge to the lake.

## 167 ***2.1.2 Field monitoring approach***

### 168 ***2.1.2.1 Field equipment***

169 For all sampling periods in 2014-2015 and 2020, a cross-shore monitoring transect was  
170 installed that aligned with the center of the septic system-derived nutrient-rich plume that  
171 is moving towards the lake. The original cross-shore monitoring transect in 2014-2015  
172 consisted of ten multi-level samplers (MLS) and extended from ~71 m downgradient of  
173 the septic tile bed ( $x = 71$  m; landward extent of the beach dunes) to ~20 m offshore ( $x =$   
174 140 m; Figure 1c). For the 2020 sampling, the cross-shore monitoring transect consisted of  
175 six MLS, and extended from  $x = 71$  m downgradient of septic tile bed ~20 m offshore ( $x =$   
176 110 m) (Figure 1c). The  $x$ - $z$  coordinate system used herein is defined with  $x = 0$  m located  
177 at the landward edge of the septic tile bed and  $z = 0$  as the elevation of a local permanent

178 benchmark. All MLS were installed up to 3 m below ground surface, with sampling ports  
179 at 0.2 m depth intervals.

180 Two piezometers (PZ) were permanently installed at  $x = 71$  m (PZ-1) and  $x = 93$  m (PZ-2)  
181 in July 2014, with additional piezometers temporarily installed along the cross-shore  
182 monitoring transect at the start of each sampling event to measure the groundwater and  
183 lake water levels across the nearshore aquifer and offshore. The location of monitoring  
184 equipment and ground surface elevations were surveyed with a total station (Topcon GTS-  
185 239W) relative to a local permanent benchmark during each sampling event (Figure 1c).

#### 186 **2.1.2.2 Electrical resistivity imaging**

187 The septic system origin of the nutrient-rich plume observed in the nearshore aquifer was  
188 confirmed in 2015 by using artificial sweeteners as a chemical tracer for human wastewater  
189 (Rakhimbekova et al., 2021a). However, the artificial sweetener data were limited to point  
190 information obtained from samples collected at selected MLS located over 70 m  
191 downgradient of the septic tile bed (i.e., samples collected from  $x = 71$  m onwards). During  
192 the field sampling in 2020, a non-invasive electrical resistivity tomography (ERT) survey  
193 was performed to provide continuous imaging of electrical resistivity within the subsurface  
194 and determine if this approach would be able to trace the nutrient-rich plume back to the  
195 septic tile beds.

196 An ERT survey line consisting of 96 stainless-steel electrodes (0.3 m) with an inline  
197 spacing of 1 m was deployed. The location of the line coincided with the location of the  
198 cross-shore monitoring transect but it extended from the landward edge of the septic tile  
199 bed ( $x = 0$  m) to the shoreline ( $x = 95$  m) (Figure 1b). A multi-channel Syscal Pro Switch

200 48 resistivity system (IRIS Instruments, France) was used to complete the survey. As this  
201 system is limited to 48 electrodes at a time, the roll-along method was used to record the  
202 survey line through three successive segments. For this, 24 electrodes were moved for each  
203 segment to ensure 50% overlap (i.e., electrode numbers: 0-48, 24-72, 48-96) (Robinson et  
204 al., 2022).

205 The multi-gradient electrode array was used with a total of 4183 measurements. Due to the  
206 dry surface sand, saline water was added around each electrode to ensure strong ground  
207 coupling (resistance: <1 kilo-ohm). The recorded apparent resistivity data was inverted  
208 with DC\_2DPro, an iterative least-squares smoothness-constrained inversion program  
209 (Kim, 2016). The surface elevation at each electrode, which was surveyed using the total  
210 station, was incorporated into the inversion, which was then run to a maximum of seven  
211 iterations, showing good convergence with a low root mean square (RMS) error of 5.5%.

### 212 ***2.1.2.3 Porewater sampling and analysis***

213 For all sampling events, pore water samples were first collected from all MLS ports using  
214 syringes attached to the PVC sampling tubes ( $\phi = 1.35$  mm). Porewater samples were  
215 collected from each port, filtered using 0.45  $\mu\text{m}$  cellulose acetate filters, and stored in 60  
216 mL HDPE bottles. Samples were analyzed for phosphate ( $\text{PO}_4\text{-P}$ ), nitrate ( $\text{NO}_3\text{-N}$ ) and  
217 chloride (Cl) within 48 hours of collection. After samples were collected, a peristaltic pump  
218 was used to draw pore water directly into a flow cell where pH, electrical conductivity (EC)  
219 and temperature were measured with an In-Situ Multi-Parameter TROLL® 9500 and YSI  
220 6600 V2 Sonde. Multiple nearshore lake water samples were also collected during each  
221 sampling event. Collected pore water samples were analyzed for nutrients ( $\text{PO}_4\text{-P}$ ,  $\text{NO}_3\text{-N}$ )

222 using Lachat 750 Flow Injection Analysis (Lachat 750), and Cl using High Performance  
223 Liquid Chromatography.

#### 224 **2.1.2.4 Sediment sampling and analysis**

225 During MLS installation in May 2015 and August 2020, multiple sediment samples were  
226 collected at depths of 0.5 m to 2 m below the ground surface for sediment chemistry  
227 characterization (Figure 1c). Sediment samples were analyzed via total acid digestion using  
228 method EPA 200.8 (US EPA, 1996) to determine the total elemental solid phase  
229 concentrations. In addition, due to the potential importance of Fe-bearing solid phases in  
230 sequestering chemical including P in nearshore aquifers, select sediment samples were  
231 treated with (i) acid ammonium oxalate solution (McKeague and Day, 1966), and (ii)  
232 dithionite-citrate-bicarbonate solution (Mehra, 1958) to provide characterization of the Fe-  
233 bearing solid phases (e.g. degree of crystallization). Oxalate-extractable Fe ( $Fe_o$ ) is  
234 operationally thought to represent amorphous oxides such as ferrihydrite and small  
235 amounts of organically-bound Fe (Schwertmann, 1964). Dithionite-extractable Fe ( $Fe_d$ ) is  
236 operationally thought to represent organically-bound oxides, amorphous oxides, and finely  
237 crystalline oxides (Akinbola et al., 2013). The difference between  $Fe_d$  and  $Fe_o$  generally  
238 represents the amount of solid phase Fe present in crystalline oxide forms (e.g., goethite  
239 and hematite). Total dissolved phosphorus (TDP) and total iron (Fe) in extractant solutions  
240 were analyzed using HACH DR 6000 UV VIS Spectrophotometer and inductively-coupled  
241 plasma mass spectroscopy (ICP-MS), respectively. Additional details on sediment sample  
242 analysis are provided in the Supporting Information (SI Section 1).

243       **2.2 Numerical model**

244       A two-dimensional (2D) model was developed in MODFLOW-NWT (Niswonger et al.,  
245       2011) to simulate steady state groundwater flows and estimate NO<sub>3</sub>-N and PO<sub>4</sub>-P loads  
246       discharging to the lake along the vertical cross-shore section. The model domain and  
247       parameterization were based on a groundwater model of the study site developed and  
248       validated by Malott et al. (2016). The two-dimensional model assumes negligible  
249       groundwater flow in the alongshore direction. The model was 150 m long in the cross-  
250       shore direction and the aquifer thickness at the shoreline location was set to 13 m at the  
251       landward boundary due to presence of a clay layer at this depth according to local well  
252       records (Figure 2). The beach face and lake boundaries (AB and BC) were specified based  
253       on topographic sand level surveys conducted during each sampling period. The model was  
254       discretized into 71 layers and 128 columns with greater refinement around the shoreline  
255       and the beach face. Grid discretization tests were performed to ensure a converged solution.  
256       Particle size analysis performed on sediment samples ( $n = 7$ ) collected across the nearshore  
257       aquifer up to 2.5 m below the water table indicated that the nearshore aquifer is comprised  
258       of relatively homogeneous fine sand (Coefficient of Uniformity,  $C_u = 2.18$  and sand grain  
259       size,  $d_{50} = 0.16$  mm) with the saturated  $K$  estimated to range from 7 to 15 m/d using the  
260       method of Krumbein and Monk (1943). Based on this, the simulated aquifer was  
261       homogeneous and isotropic with the hydraulic conductivity ( $K$ ) set to 10 m/d and effective  
262       porosity  $n = 0.3$ .

263       The upper (AB) and bottom (ED) model boundaries were no-flow boundaries assuming  
264       negligible aerial recharge and evaporation, and an impermeable aquifer basement,  
265       respectively. The offshore boundary (CD) was specified as no-flow boundary and was

266 located sufficiently far from the shoreline (>50 m offshore) that it did not influence the  
267 groundwater flows in the nearshore aquifer (Figure 2). The vertical landward boundary  
268 (AE) was located at  $x = 60$  m, which is 10 m inland from the landward extent of the beach  
269 dunes (Figure 2) and was set to be a constant head boundary. Three simulations were run  
270 with the boundary conditions modified to represent the mean hydrological conditions at  
271 the study site during the 2014, 2015 and 2020 sampling periods. Based on the measured  
272 groundwater levels, the hydraulic head at the landward boundary was set to -2.95 m, -3.05  
273 m and -1.95 m, respectively, for the models simulating the groundwater flows over the  
274 sampling periods in 2014, 2015 and 2020. A constant head condition was used along the  
275 lake boundary (BC) with the specified head values set based on the mean measured lake  
276 water level during each of the sampling periods and considering the effect of steady waves  
277 acting on the boundary. To simulate the effect of steady waves, the approach of Anwar et  
278 al. (2014) was adopted whereby the time-averaged effect of waves as described by wave  
279 set-up was used (details are provided in SI Section 2). For this, the empirical equation of  
280 Nielsen (2009) was used to estimate the wave-set up profile and calculate head values that  
281 were then prescribed along the lake boundary (BC). Offshore root mean square wave  
282 heights ( $H_{rms}$ ) of 0.35 m, 0.25 m and 0.35 m were used in the equation of Nielsen (2009)  
283 to simulate the mean wave conditions over the sampling periods in 2014, 2015 and 2020,  
284 respectively. The mean  $H_{rms}$  values were calculated using monitoring data from a wave  
285 buoy located ~35 km from the study site ( C45149 Southern Lake Huron, Department of  
286 Fisheries and Oceans Canada Buoy, 2020). Based on the mean lake water levels measured  
287 at the study site during the sampling periods in 2014, 2015 and 2020, the still lake water  
288 level in the models were set to -3.64 m, -3.73 m and -2.21 m, respectively.

## 289 **3 Results and discussion**

### 290 **3.1 Field and modeling results**

#### 291 **3.1.1 *Change in hydrological conditions between sampling periods***

292 The hydrological conditions at the study site including the mean lake water level, mean  
293 wave conditions, and landward groundwater levels varied between the sampling periods.  
294 The mean lake water level was approximately 0.7 m higher in 2020 compared to 2014-  
295 2015 (Figure 3a). More intensified wave conditions were also observed in 2020 compared  
296 to 2014-2015 with maximum wave height ( $H_{rms}$ ) reaching 3.7 m in 2020 compared to 3.0  
297 m in 2014 and 2.1 m in 2015 (mean  $H_{rms}$  was 0.35 m, 0.25 m and 0.35 m in 2020, 2015 and  
298 2014, respectively; Figure 3b). The extent of shoreline recession between 2015 and 2020  
299 is evident by comparing the topographical surveys performed in 2015 and 2020, along with  
300 corresponding aerial photographs (Figures 1b and 1c). The shoreline was estimated to  
301 move 30 m landward from  $x = 120$ - $125$  m in 2014-2015, to  $x = 90$  m in 2020, in response  
302 to the lake water level increase and beach sand erosion.

303 Increase in lake water levels may lead to a corresponding increase in the landward  
304 groundwater table elevation, and thereby reduce the vertical separation between septic tile  
305 beds and groundwater table. This can be a concern as it then limits the efficiency of the  
306 septic tile bed in attenuating wastewater constituents including nutrients (Mitchell et al.,  
307 2021). Based on groundwater levels measured at PZ-1 ( $x = 71$  m), there was  $\sim 1$  m increase  
308 in the groundwater table between 2014-2015 and 2020 (Figure 1c). However, with high  
309 sand dunes behind the beach, there is a thick (3 - 4 m) vadose zone below the septic tile  
310 beds at this study site (see Section 3.1.2 for details), and therefore it is not expected that  
311 the higher groundwater table elevation in 2020 compared to 2014 – 2015 would have

312 adversely impacted the function of the vadose zone below the septic tile beds in attenuating  
313 wastewater constituents. The adverse impact of increasing coastal levels on the attenuation  
314 of wastewater constituents below the septic tile beds is expected to be a larger concern in  
315 coastal plain settings with flatter topography and shallow groundwater tables.

316 The mean hydraulic gradient across nearshore aquifer was negative in 2014-2015 ( $h = -$   
317  $0.015$ ) and 2020 ( $h = -0.017$ ), indicating lakeward groundwater flow. Based on the  
318 hydraulic gradient in 2014-2015 and 2020, mean hydraulic conductivity ( $K = 10$  m/d) and  
319 effective porosity ( $n = 0.3$ ) for the nearshore aquifer, the mean horizontal groundwater  
320 velocity towards the lake is estimated to be 0.50 m/d in 2014-2015 and 0.56 m/d in 2020.  
321 Numerical simulations show the change in groundwater flow patterns (Figure 4a-c) in the  
322 nearshore aquifer between the 2014-2015 and 2020 sampling periods with the groundwater  
323 discharge zone shifting from  $x = 120$ -125 m in 2014-2015 to  $x = 85$ -90 m in 2020. Based  
324 on the estimated overall mean horizontal velocities and shoreline positions, the travel time  
325 for conservative species to be transported from the septic tile bed to the shoreline is  
326 estimated to be around 240 days in 2014-2015 compared to only 160 days in 2020.

327 While the flow patterns simulated by the steady state groundwater models do not indicate  
328 recirculation of lake water through the nearshore aquifer (Figure 4a-c), recirculating flows  
329 driven mostly by changing lake water levels and wave conditions are expected to occur  
330 landward of the mean shoreline position and simulated groundwater discharge zone (Malott  
331 et al., 2016). With the mean shoreline position and zone of groundwater discharge shifting  
332 landward in 2020 compared to 2014-2015, the recirculation zone, which is generally  
333 associated with high reactivity including the presence of mineral phases (e.g., Fe oxides)  
334 that tend sequester pollutants (Lee et al., 2014; Lewandowski et al., 2015; Rakhimbekova

335 et al., 2018), would also have shifted landward. It is possible that this landward shift in the  
336 recirculation zone could have altered geochemical conditions in the nearshore aquifer and  
337 this is explored in the following sections.

### 338 **3.1.2 Characterization of septic-derived nutrient plume**

339 Elevated groundwater PO<sub>4</sub>-P and NO<sub>3</sub>-N concentrations were observed at the study site in  
340 2014, 2015 and 2020 (Figures 4d to 4i). Groundwater PO<sub>4</sub>-P concentrations ranged from 8  
341 to 166 µg/L in 2014-2015, and from 3 to 98 µg/L in 2020. Maximum PO<sub>4</sub>-P concentrations  
342 in the nearshore aquifer exceeded the upper limit of the Canadian Water Quality Guidelines  
343 for the Protection of Aquatic Life for Total Phosphorus to limit eutrophic conditions (35–  
344 100 µg P/L; Canadian Council of Ministers of the Environment, 2014), and were  
345 considerably higher than the mean measured lake water concentrations for all sampling  
346 periods (2-3 µg/L). Groundwater NO<sub>3</sub>-N concentrations ranged from 4 to 57 mg/L in 2014-  
347 2015 and from 1 to 26 mg/L in 2020. Maximum NO<sub>3</sub>-N concentrations in the nearshore  
348 aquifer exceeded the Canadian Water Quality Guidelines for the Protection of Aquatic Life  
349 for all sampling periods (3 mg/L; Canadian Council of Ministers of the Environment,  
350 2014), and were also higher than the mean measured lake water concentrations (0.2-0.8  
351 mg/L).

352 Figures 4j-l to 4m-o presents the distributions of chloride (Cl) concentrations and electrical  
353 conductivity (EC) measured along the monitoring transect in 2014, 2015 and 2020. Cl and  
354 EC can be considered conservative wastewater tracers at the site(Rakhimbekova et al.,  
355 2021a), and their distributions generally coincide with the locations of elevated PO<sub>4</sub>-P and  
356 NO<sub>3</sub>-N in the nearshore aquifer Figure 4d-i. NO<sub>3</sub>-N with EC landward of the maximum  
357 shoreline position are correlated (Figure S1) indicating the suitability of using ERT for

358 characterizing the spatial extent of the high nutrient groundwater plume. Figure 5 shows a  
359 2D cross-sectional ERT image that extends from  $x = 0$  m (septic tile bed) to  $x = 95$  m (near  
360 the mean shoreline position in 2020). The location of the septic tile bed, PZ-1 and MLS are  
361 indicated in the figure, along with the lake water and groundwater levels measured during  
362 the survey. The highly resistive near surface layer between  $x = 0$  m and  $x = 78$  m (red-  
363 purple;  $>1000$  ohm-m) represents the vadose zone with a mean thickness of approximately  
364 4 m. The underlying layer of low resistivity (blue-green;  $<150$  ohm-m) represents the zone  
365 of saturated groundwater, with some areas having very low resistivity ( $<50$  ohm-m). The  
366 ERT image can be validated by comparing bulk resistivity values with the porewater EC  
367 measurements in Figure 4m-o. For example, the porewater EC at  $x = 71$  m and  $z = -3$  m  
368 (PZ-1) is  $1008 \mu\text{S}/\text{cm}$  which suggests a bulk EC of  $209 \mu\text{S}/\text{cm}$ , or bulk resistivity of  $48$   
369 ohm-m (Archie, 1942) – this closely matches the bulk resistivity inferred by ERT at that  
370 location ( $55$  ohm-m). Figure 5 indicates that the zone of very low bulk resistivity is  
371 connected to the base of the septic tile bed and extends lakeward. This zone exists through  
372 the shallow nearshore aquifer and becomes shallower near the shoreline meeting the  
373 sediment-water interface around  $x = 92$  m. This ERT image clearly illustrates that the  
374 elevated  $\text{NO}_3\text{-N}$  and  $\text{PO}_4\text{-P}$  concentrations observed near the shoreline originate from the  
375 septic tile bed.

### 376 **3.1.3 Impact of shoreline recession on $\text{PO}_4\text{-P}$**

377 As seen in Figure 4, the highest  $\text{PO}_4\text{-P}$  concentrations were observed 73-90 m lakeward of  
378 the septic tile bed (maximum  $\text{PO}_4\text{-P} = 166 \mu\text{g}/\text{L}$ ;  $x = 73\text{-}90$  m) in 2014-2015, with  
379 concentrations decreasing closer to the shoreline (particularly in 2015). However, elevated  
380 concentrations of the conservative wastewater tracers, Cl and EC, near the shoreline ( $x =$

381 115-125 m) for these sampling periods suggest that the septic-derived wastewater plume  
382 has reached the shoreline. Taking into account septic system operation time, groundwater  
383 velocity, and length of the PO<sub>4</sub>-P plume in 2015, Rakhimbekova et al. (2021) concluded  
384 that PO<sub>4</sub>-P transport in the nearshore aquifer was mainly governed by retardation. Based  
385 on the location of the PO<sub>4</sub>-P plume in 2015, it was predicted that PO<sub>4</sub>-P may possibly reach  
386 the shoreline in approximately 10 years. However, this prediction is incorrect due to  
387 shoreline recession at the site. Figure 4f shows that in 2020 (within <5 years), the PO<sub>4</sub>-P  
388 plume has reached the new shoreline location ( $x = 90$  m). To provide a quantitative estimate  
389 on the extent to which PO<sub>4</sub>-P fluxes to the lake may have changed, the measured PO<sub>4</sub>-P  
390 concentrations were combined with the simulated groundwater flows for each sampling  
391 period. PO<sub>4</sub>-P fluxes were calculated by multiplying PO<sub>4</sub>-P concentrations (interpolated to  
392 a depth of 3 m below ground surface) along vertical line near the shoreline ( $x = 120.3$  m in  
393 2014 and 2015 and  $x = 88.7$  m in 2020; Figure 4 a-c) by simulated groundwater fluxes  
394 directed towards the lake for each sampling year (Additional details provided in SM S3).  
395 From these calculations, it is estimated that the PO<sub>4</sub>-P flux to the lake may have increased  
396 nearly tenfold from 53 mg/d/m in 2015 to 482 mg/d/m in 2020 (Figure 6a) due higher PO<sub>4</sub>-  
397 P concentrations near the shoreline in 2020 compared to near the shoreline in 2015 as well  
398 as slightly higher lakeward-directed groundwater velocity (Figure 4e, f). This result  
399 illustrates that the shoreline recession can accelerate the discharge of groundwater-derived  
400 pollutants that are slowly advancing towards the shoreline by decreasing the subsurface  
401 travel distance and time.

#### 402        **3.1.4 Impact of shoreline recession on NO<sub>3</sub>-N**

403        The NO<sub>3</sub>-N distribution in the nearshore aquifer was influenced by not only its shorter  
404        discharge pathway through the nearshore aquifer due to shoreline recession but also by the  
405        specific geochemical conditions near the sediment-water interface that impact reactive  
406        transformations that NO<sub>3</sub>-N may undergo prior to its discharge. Rakhimbekova et al.  
407        (2021a) previously showed that the extent of NO<sub>3</sub>-N removal that occurs in the nearshore  
408        aquifer prior to discharge may vary over time as the extent of denitrification changes in  
409        response to changing geochemical conditions near the sediment-water interface (e.g., due  
410        to changes in organic matter availability). Rakhimbekova et al. (2021a) presented detailed  
411        data analysis showing changing organic matter availability was associated with the  
412        observed decrease in NO<sub>3</sub>-N concentrations near the shoreline in 2015 (0-2 mg/L)  
413        compared to 2014 (4-67 mg/L), and thus lower estimated NO<sub>3</sub>-N flux towards the lake in  
414        2015 (4 mg/d/m) compared to 2014 (124 mg/d/m) (Figure 6b).

415        NO<sub>3</sub>-N concentrations observed near the mean shoreline position in 2020 ( $x = 85-90$  m)  
416        ranged from 4 to 26 mg/L in 2020 (Figure 4i). The estimated NO<sub>3</sub>-N flux directed towards  
417        the lake near the shoreline in 2020 (79 mg/d/m) was larger than in 2015 (4 mg/d/m) but  
418        smaller than in 2014 (124 mg/d/m). NO<sub>3</sub>-N flux is affected by both groundwater velocity  
419        and NO<sub>3</sub>-N concentrations. Since groundwater velocity near the shoreline was slightly  
420        higher in 2020 compared to 2014 (difference in flow arrow sizes in Figure 4 a-c), the higher  
421        estimated NO<sub>3</sub>-N flux in 2014 is due to higher NO<sub>3</sub>-N concentrations near the shoreline in  
422        2014 compared to 2020. Therefore, even though shoreline recession reduced the travel  
423        distance and travel time for the septic-derived NO<sub>3</sub>-N to reach the lake, the specific  
424        geochemical conditions in the nearshore aquifer near the shoreline ultimately govern the

425 flux of  $\text{NO}_3\text{-N}$  to the lake rather than the overall travel distance and time. This finding is  
426 in contrast to the highly retarded  $\text{PO}_4\text{-P}$  plume for which shoreline recession was found to  
427 directly accelerate  $\text{PO}_4\text{-P}$  discharge to the lake due to the overall reduced travel distance  
428 between the septic tile bed and shoreline.

429 It is important to note that while the geochemical conditions near the shoreline have a  
430 controlling influence on the fate of  $\text{NO}_3\text{-N}$  in the nearshore aquifer and its flux to the lake  
431 at our study site, this may not be the case at other sites. The nearshore aquifer at our study  
432 site (from the septic tile bed to shoreline) is predominantly oxic and therefore limited  
433 denitrification occurs as  $\text{NO}_3\text{-N}$  is transported towards the shoreline. At other sites where  
434 the nearshore aquifer is more reducing, it is possible that the decrease in overall travel  
435 distance and times could limit the extent of denitrification that occurs as  $\text{NO}_3\text{-N}$  is  
436 transported from the source to the coastal water (Meile et al., 2010). Further, it is possible  
437 that shoreline recession could have a greater impact on the discharge of reactive pollutants  
438 in nearshore aquifers with sharper geochemical (redox and pH) gradients than observed at  
439 Ipperwash Beach (Rakhimbekova et al., 2021a, 2018).

#### 440 ***3.1.5 Impact of shoreline recession on Fe curtail processes***

441 In addition to modifying the physical transport pathways and travel times for the septic-  
442 derived nutrients being transported towards to the lake, the change in shoreline location  
443 may also modify the geochemical conditions near the groundwater-lake interface. In  
444 particular, it may modify the distribution, abundance and reactivity of mineral phases that  
445 can sequester pollutants in the nearshore aquifer. Rakhimbekova et al. (2021b) showed  
446 elevated solid phase Fe and P exists in the nearshore aquifer at the study site and suggested  
447 that  $\text{PO}_4\text{-P}$  derived from the decomposition of lake-derived organic matter may be

448 accumulating on Fe oxide minerals that exist close to the sediment-water interface.  
449 Previous studies in marine environments have similarly found that Fe oxides that form near  
450 the sediment-water interface can act as a geochemical barrier (i.e. “iron curtain”) with these  
451 minerals sequestering pollutants such as PO<sub>4</sub>-P and limiting their release to coastal  
452 waters(Charette and Sholkovitz, 2002; Spiteri et al., 2006). While it is thought that  
453 sequestered pollutants may later be re-mobilized if geochemical or hydrological conditions  
454 change (Nisbeth et al., 2019; Spiteri et al., 2006), in this section we examine how the  
455 distribution and abundance of solid phase P and Fe, including Fe oxides, in the nearshore  
456 aquifer may have changed in response to shoreline recession.

457 Rakhimbekova et al. (201b) previously showed that nearshore aquifer at the study site is  
458 predominantly oxidizing, and there is no distinct redox gradient near the sediment-water  
459 interface. Further, Fe oxides are available through the nearshore aquifer and tend to  
460 accumulate PO<sub>4</sub>-P on their mineral surfaces. Characterization of the distribution of solid  
461 phase P and Fe in the nearshore aquifer at the study site in 2015 and 2020 indicate a  
462 decrease in solid phase P and Fe between these times (Figure 7a-d). Due to the landward  
463 movement of the shoreline, sediment samples were taken in different locations in 2015  
464 compared with 2020. To enable comparison of P and Fe solid phase concentrations  
465 between two times, comparison only considers samples collected over the same area,  
466 between  $x = 87$  and 117 m (marked with red square on Figure 7). Generally, there was  
467 higher abundance of solid phase P near the shoreline in 2015 compared to 2020 ( $151 \pm 51$   
468  $\mu\text{g/g}$  in 2015,  $n = 17$ ;  $82 \pm 55 \mu\text{g/g}$  in 2020,  $n = 23$ ; Figure 7a and b). Solid phase Fe was  
469 also slightly higher in 2015 compared to 2020 ( $3.1 \pm 0.6 \text{ mg/g}$  in 2015,  $n = 17$ ;  $2.7 \pm 0.7$   
470  $\text{mg/g}$  in 2020,  $n = 23$ ; Figure 7c and d).

471 Selective solid phase extractions were conducted to characterize Fe-bearing solid phases  
472 (amorphous, Fe<sub>o</sub> and more crystalline, Fe<sub>d</sub>) at the study site in 2015 and 2020. Again, only  
473 sediment samples collected between  $x = 87$  and 117 m were used to compare the  
474 distribution of amorphous and crystalline Fe phases in 2015 and 2020. The abundance of  
475 Fe<sub>o</sub> decreased between 2015 and 2020 ( $0.5 \pm 0.2$  mg/g in 2015,  $n = 10$ ;  $0.2 \pm 0.09$  mg/g in  
476 2020,  $n = 13$ ; Figure 7e and f). In contrast, Fe<sub>d</sub> was slightly higher in 2020 compared to  
477 2015 ( $0.5 \pm 0.07$  mg/g in 2015;  $0.7$  mg/g  $\pm 0.3$  in 2020; Figure 7g vs 7h). Lower Fe<sub>d</sub> in  
478 2015 compared to 2020 may possibly be due to the formation of more crystalline Fe phases  
479 over time. Larger difference in the abundance of Fe<sub>o</sub> compared to Fe<sub>d</sub> between 2015 and  
480 2020 suggest that shoreline recession may have larger impact on amorphous Fe oxides  
481 compared to crystalline Fe oxides. This is not unexpected given that amorphous Fe oxides  
482 are less stable and shoreline recession may have altered factors that affect Fe oxide stability  
483 (e.g. oxygen availability, presence of bacteria) (Charette and Sholkovitz, 2002; Dean et al.,  
484 2003). Following shoreline recession, particularly a large erosive event, it might take time  
485 for amorphous Fe oxides to reform around the new shoreline location. This may explain  
486 the lower Fe<sub>o</sub> near the new shoreline location in 2020 compared to higher Fe<sub>o</sub> near the  
487 location of shoreline in 2015, which was a more stable shoreline position over the  
488 preceding years.

489 Amorphous Fe (e.g. ferrihydrite) tend to have a higher surface area that is more reactive  
490 and therefore has a higher capacity to sequester pollutants, including PO<sub>4</sub>-P, compared to  
491 more crystalline Fe oxides fractions such as goethite and hematite (Fuller et al., 1993).  
492 Therefore, lower abundance of Fe<sub>o</sub> in the nearshore aquifer in 2020 (following the shoreline  
493 recession) suggests that mineral surfaces along the groundwater discharge pathway may

494 have less ability to sequester PO<sub>4</sub>-P, and this may possibly lead to higher PO<sub>4</sub>-P fluxes to  
495 lake. The impact of shoreline recession on the distribution and abundance of Fe oxide  
496 minerals, particularly amorphous Fe oxides, is important as it may affect the function of  
497 these minerals in sequestering and releasing pollutants thereby modifying pollutant fluxes  
498 to coastal waters.

### 499 **3.2 Broader implications of the impact of shoreline recession**

500 The study findings demonstrate the potential impacts of shoreline recession driven by  
501 coastal water level changes and/or erosion on the transport and fate of groundwater  
502 pollutants and their discharge to adjacent coastal waters. The findings show shoreline  
503 recession may impact pollutants moving slowly along groundwater pathways towards the  
504 coastal water, as well as pollutants for which their fate and ultimate delivery to coastal  
505 waters is governed strongly by the geochemical conditions in the reaction zone that exists  
506 near the groundwater-coastal water interface. This section explores the broader  
507 implications of the study findings on the discharge of groundwater pollutants to the  
508 Laurentian Great Lakes, as well as other inland and marine coastal waters.

509 While it is known that water levels in the Great Lakes vary over 20 – 30 year cycles  
510 (Gronewold and Rood, 2019), recent future projections estimate that by 2040-2049 average  
511 annual water levels of Lake Superior, Lake Michigan-Huron and Lake Erie are projected  
512 to increase by +0.19, +0.44, +0.28 m, respectively due to an increase in over-lake  
513 precipitation and basin runoff compared to relatively smaller increase in lake evaporation  
514 (Kayastha et al., 2022). While there are no studies predicting future erosion rates and  
515 shoreline recession in response to the projected water level changes for the Great Lakes,  
516 several studies show that high water levels correspond with accelerated shoreline erosion,

517 including bluff erosion, on short time scales (e.g. Kaczmarek et al., 2016; Krueger et al.,  
518 2020). For instance, Castedo et al. (2013) simulated erosion rates to range between 0.5-1.5  
519 m/yr in response to lake water level variations of 0.5-1 m. A more recent study by Volpano  
520 et al. (2020) on coastal bluff erosion along Western Lake Michigan's shoreline found that  
521 erosion rates increased from 0.18-0.43 m/yr between 2009 and 2014 (period with below  
522 long term average water levels) to 0.49-1.19 m/yr between 2014 and 2018 (period with  
523 above long-term average water levels).

524 The outlook for marine shorelines is also concerning. On average, global sea level has risen  
525 at a rate of 3-4 mm/yr in the last 20 years (Yi et al., 2015). Future projections estimate sea  
526 level rise from 0.1-0.2 m by 2050 and from 0.3-2 m by 2100 due to ocean warming and  
527 land-based ice melt (Kopp et al., 2014; Vitousek et al., 2017). While shoreline recession is  
528 generally the combined result of various contributing factors and may be complicated by  
529 local and regional differences, it is widely acknowledged that increasing sea levels will  
530 promote shoreline recession (e.g. Anderson et al., 2015; Bruun, 1962; Dean and Houston,  
531 2016). For instance, a recent study predicted that 15% of world's sandy beaches could face  
532 severe coastal erosion by 2050 and up to 50% by the end of the century (Vousdoukas et  
533 al., 2020). Taking into consideration the projected sea level changes and associated  
534 shoreline recession, together with the findings of our study, there is high potential that  
535 future shoreline recession will modify inputs of groundwater pollutants to coastal waters.

536 While this study focuses on the wastewater plume from one septic system, there are 1.2  
537 and 1.3 million septic systems in Ontario, Canada (DeRabbie, 2020) and Michigan, United  
538 States (Department of Environment, Great Lakes and Energy, 2022), respectively, with  
539 many of these septic systems located in close proximity to the shorelines of the Great Lake.

540 It is also important to note that in addition to nutrients, septic-derived wastewater effluent  
541 also contains other pollutants of concern including pharmaceuticals, personal care products  
542 and pathogens, all of which may pose a risk to coastal water pollution with associated  
543 human health and ecosystem impacts. Further, many other groundwater contaminant  
544 sources are often located in coastal areas whereby shoreline recession may modify and  
545 potentially accelerate pollutant inputs to coastal waters. This includes landfills (both  
546 historic and currently operating) (Brand et al., 2018; Stefania et al., 2019) as well as  
547 industrial sites including Superfund sites which are the most contaminated sites in the  
548 United States (Carter, 2020).

#### 549 **4 Conclusions**

550 This study showed the impact of shoreline recession in modifying the fate of groundwater  
551 pollutants discharging to coastal waters at a specific study site, and discussed the broader  
552 potential implications for inland and marine coastal waters. The study findings show that  
553 the shoreline recession may accelerate the discharge of groundwater pollutants to coastal  
554 waters by decreasing the subsurface travel distance and travel time. In addition, shoreline  
555 recession may modify reactive processes close to the sediment-water interface, which may  
556 in turn affect the fate of reactive pollutants and their discharge to coastal waters.  
557 Importantly, shoreline recession may result in a landward shift of the recirculation zone  
558 that commonly exists close to the shoreline, and in doing so, alter the stability and  
559 abundance of sediment phases (e.g., Fe oxides) that often act to sequester pollutants and  
560 potentially limit their release to coastal waters. In particular, the study findings show that  
561 amorphous Fe oxides may be particularly impacted by the shoreline recession and may

562 take time to reform and function to sequester pollutants along the modified groundwater  
563 discharge pathway.

564 As it is expected that climate change will accelerate shoreline recession on coastlines  
565 worldwide due to increase in the magnitude of lake water level fluctuations, sea level rise,  
566 as well as increased frequency and intensity of storm events, it is recommended that future  
567 work evaluate the impact of shoreline changes of different time scales (e.g., event based,  
568 yearly, decadal) on pollutant fate in nearshore aquifers. The nearshore aquifer at the study  
569 site was relatively homogeneous and predominantly oxidizing and it is recommended that  
570 future studies evaluate the impact of shoreline recession in different nearshore aquifer types  
571 including more heterogeneous aquifers and those with more reducing conditions. Further,  
572 while this study focused on a freshwater coastal aquifer, field investigations are required  
573 to determine the impact of shoreline recession on the functioning of the reaction zone in  
574 nearshore aquifers along marine shorelines (i.e., subterranean estuary) and the  
575 corresponding impact on pollutant fluxes to marine waters. As the study data also  
576 demonstrate that the abundance and form of solid phase Fe near the shoreline may vary in  
577 response to shoreline change, there is a need to better understand the mineralization process  
578 for Fe oxides to better predict the impact of future changes on their functioning especially  
579 with respect to sequestering and releasing pollutants of concern. While this study focused  
580 on the impact of shoreline recession on nutrients derived from a septic system, it is  
581 recommended that future studies explore the potential impacts of shoreline recession on  
582 the discharge of other anthropogenic pollutants of concern that are commonly elevated in  
583 coastal groundwater systems.

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588 **References**

- 589 Adams, K.D., Minor, T.B., 2002. Historic shoreline change at Lake Tahoe from 1938 to 1998  
590 and its impact on sediment and nutrient loading. *J. Coast. Res.* 18, 637–651.
- 591 Akinbola, G.E., Adigun, M.O., Aduroja, O., 2013. Dithionite and oxalate extraction of iron and  
592 manganese in some basement complex soils of southwestern Nigeria. *J. Exp. Sci.* 4, 22–26
- 593 Anderson, T.R., Fletcher, C.H., Barbee, M.M., Frazer, L.N., Romine, B.M., 2015. Doubling of  
594 coastal erosion under rising sea level by mid-century in Hawaii. *Nat. Hazards* 78, 75–103.  
595 <https://doi.org/10.1007/s11069-015-1698-6>
- 596 Anwar, N., Robinson, C., Barry, D.A., 2014. Influence of tides and waves on the fate of nutrients  
597 in a nearshore aquifer: Numerical simulations. *Adv. Water Resour.* 73, 203–213.  
598 <https://doi.org/10.1016/j.advwatres.2014.08.015>
- 599 Argyilan, E.P., Forman, S.L., 2003. Lake level response to seasonal climatic variability in the  
600 Lake Michigan-Huron system from 1920 to 1995. *J. Great Lakes Res.* 29, 488–500.  
601 [https://doi.org/10.1016/S0380-1330\(03\)70453-5](https://doi.org/10.1016/S0380-1330(03)70453-5)
- 602 Beck, A.J., Kellum, A.A., Luek, J.L., Cochran, M.A., 2016. Chemical flux associated with  
603 spatially and temporally variable submarine groundwater discharge, and chemical  
604 modification in the subterranean estuary at Gloucester Point, VA (USA). *Estuaries and*  
605 *Coasts* 39, 1–12. <https://doi.org/10.1007/s12237-015-9972-0>
- 606 Beck, M., Dellwig, O., Schnetger, B., Brumsack, H.J., 2008. Cycling of trace metals (Mn, Fe,  
607 Mo, U, V, Cr) in deep pore waters of intertidal flat sediments. *Geochim. Cosmochim. Acta*  
608 72, 2822–2840. <https://doi.org/10.1016/j.gca.2008.04.013>

609 Bełdowska, M., Bełdowski, J., Kwasigroch, U., Szubska, M., Jędruch, A., 2022. Coastal cliff  
610 erosion as a source of toxic, essential and nonessential metals in the marine environment.  
611 *Oceanologia*. <https://doi.org/10.1016/j.oceano.2022.04.001>

612 Bird, E.C., 2008. *Coastal geomorphology*, Second Edition. John Wiley & Sons Ltd. West  
613 Sussex. England.

614 Brand, J.H., Spencer, K.L., O'shea, F.T., Lindsay, J.E., 2018. Potential pollution risks of historic  
615 landfills on low-lying coasts and estuaries. *WIREs Water* 5, 1–12.  
616 <https://doi.org/10.1002/wat2.1264>

617 Braun, K.N., Theuerkauf, E.J., Masterson, A.L., Curry, B.B., Horton, D.E., 2019. Modeling  
618 organic carbon loss from a rapidly eroding freshwater coastal wetland. *Sci. Rep.* 9, 1–13.  
619 <https://doi.org/10.1038/s41598-019-40855-5>

620 Brooks, S.M., Spencer, T., Christie, E.K., 2017. Storm impacts and shoreline recovery:  
621 Mechanisms and controls in the southern North Sea. *Geomorphology* 283, 48–60.  
622 <https://doi.org/10.1016/j.geomorph.2017.01.007>

623 Bruun, P., 1962. Sea-level rise as a cause of shore erosion. *J. Waterw. Harb. Div.* 88.  
624 <https://doi.org/10.1061/JWHEAU.0000252>

625 Burningham, H., French, J., 2017. Understanding coastal change using shoreline trend analysis  
626 supported by cluster-based segmentation. *Geomorphology* 282, 131–149.  
627 <https://doi.org/10.1016/j.geomorph.2016.12.029>

628 Canadian Council of Ministers of the Environment, 2014. *Canadian Water Quality Guidelines*  
629 *for the Protection of Aquatic Life*.

630 Carter, Jacob, Casey Kalman. 2020. A toxic relationship: Extreme coastal flooding and  
631 Superfund sites. Cambridge, MA: Union of Concerned Scientists.  
632 <https://www.ucsusa.org/resources/toxic-relationship>

633 Castedo, R., Fernández, M., Trenhaile, A.S., Paredes, C., 2013. Modeling cyclic recession of  
634 cohesive clay coasts: Effects of wave erosion and bluff stability. *Mar. Geol.* 335, 162–176.  
635 <https://doi.org/10.1016/j.margeo.2012.11.001>

636 Charette, M.A., Sholkovitz, E.R., 2002. Oxidative precipitation of groundwater-derived ferrous  
637 iron in the subterranean estuary of a coastal bay. *Geophys. Res. Lett.* 29, 85-1-85–4.  
638 <https://doi.org/10.1029/2001gl014512>

639 Davidson, M.A., Turner, I.L., Splinter, K.D., Harley, M.D., 2017. Annual prediction of shoreline  
640 erosion and subsequent recovery. *Coast. Eng.* 130, 14–25.  
641 <https://doi.org/10.1016/j.coastaleng.2017.09.008>

642 Dean, R.G., Houston, J.R., 2016. Determining shoreline response to sea level rise. *Coast. Eng.*  
643 114, 1–8. <https://doi.org/10.1016/j.coastaleng.2016.03.009>

644 Dean, W.E., Neff, B.P., Rosenberry, D.O., Winter, T.C., Parkhurst, R., 2003. The significance of  
645 groundwater to the accumulation of iron and manganese in the sediments of two  
646 hydrologically distinct lakes in North-Central Minnesota: a geological perspective.  
647 *GroundWater* 41, 951–963. <https://doi.org/10.1111/j.1745-6584.2003.tb02437.x>

648 Department of Environment, Great Lakes and Energy, 2022. Septic Smart  
649 [https://www.michigan.gov/egle/about/organization/drinking-water-and-environmental-](https://www.michigan.gov/egle/about/organization/drinking-water-and-environmental-health/onsite-wastewater-management/septicmart)  
650 [health/onsite-wastewater-management/septicmart](https://www.michigan.gov/egle/about/organization/drinking-water-and-environmental-health/onsite-wastewater-management/septicmart)

651 Department of Fisheries and Oceans Canada Buoy, 2014. C45149 Southern Lake Huron  
652 <http://www.meds-sdmm.dfo-mpo.gc.ca/isdm-gdsi/waves-vagues/data-donnees/>  
653 [index-eng.asp](http://www.meds-sdmm.dfo-mpo.gc.ca/isdm-gdsi/waves-vagues/data-donnees/index-eng.asp)

654 DeRabbie, D., 2020. Septic Systems and Homeowner Insurance. Ontario Onsite Wastewater  
655 Association Newsletter, 15, 1.  
656 [https://www.oowa.org/wp-content/uploads/2020/05/R\\_Vol15\\_Iss1.pdf](https://www.oowa.org/wp-content/uploads/2020/05/R_Vol15_Iss1.pdf)

657 Destouni, G., Persson, K., Prieto, C., Jarsjö, J., 2010. General quantification of catchment-scale  
658 nutrient and pollutant transport through the subsurface to surface and coastal waters.  
659 *Environ. Sci. Technol.* 44, 2048–2055. <https://doi.org/10.1021/es902338y>

660 Eyles, N., Meulendyk, T., 2012. Ground-penetrating radar stratigraphy and depositional model  
661 for evolving Late Holocene aeolian dunes on the Lake Huron coast, Ontario. *J. Great Lakes*  
662 *Res.* 38, 708–719. <https://doi.org/10.1016/j.jglr.2012.09.003>

663 Fuller, C.C., Dadis, J.A., Waychunas, G.A., 1993. Surface chemistry of ferrihydrite: Part 2.  
664 Kinetics of arsenate adsorption and coprecipitation. *Geochim. Cosmochim. Acta* 57, 2271–  
665 2282. [https://doi.org/10.1016/0016-7037\(93\)90568-H](https://doi.org/10.1016/0016-7037(93)90568-H)

666 García-Rubio, G., Huntley, D., Russell, P., 2015. Evaluating shoreline identification using  
667 optical satellite images. *Mar. Geol.* 359, 96–105.  
668 <https://doi.org/10.1016/j.margeo.2014.11.002>

669 Government of Canada - Environment and Climate Change Canada, U.S. EPA, 2009. Nearshore  
670 areas of the Great Lakes. <https://doi.org/978-1-100-13563-2>

671 Gronewold, A.D., Do, H.X., Mei, Y., Stow, C.A., 2021. A tug-of-war within the hydrologic

672 cycle of a continental freshwater basin. *Geophys. Res. Lett.* 48, 1–8.  
673 <https://doi.org/10.1029/2020GL090374>

674 Gronewold, A.D., Rood, R.B., 2019. Recent water level changes across Earth’s largest lake  
675 system and implications for future variability. *J. Great Lakes Res.* 45, 1–3.  
676 <https://doi.org/10.1016/j.jglr.2018.10.012>

677 Gronewold, A.D., Stow, C.A., 2014. Water loss from the Great Lakes. *Science* (80-. ). 343,  
678 1084–1085. <https://doi.org/10.1126/science.1249978>

679 Hagenars, G., de Vries, S., Luijendijk, A.P., de Boer, W.P., Reniers, A.J.H.M., 2018. On the  
680 accuracy of automated shoreline detection derived from satellite imagery: A case study of  
681 the sand motor mega-scale nourishment. *Coast. Eng.* 133, 113–125.  
682 <https://doi.org/10.1016/j.coastaleng.2017.12.011>

683 Hewlett, C., North, R.L., Johansson, J., Vandergucht, D.M., Hudson, J.J., 2015. Contribution of  
684 shoreline erosion to nutrient loading of the Lake Diefenbaker reservoir, Saskatchewan,  
685 Canada. *J. Great Lakes Res.* 41, 110–117. <https://doi.org/10.1016/j.jglr.2014.11.020>

686 Howard, K.W.F., Livingstone, S., 2000. Transport of urban contaminants into Lake Ontario via  
687 sub-surface flow. *Urban Water* 2, 183–195.  
688 [https://doi.org/10.1016/S1462-0758\(00\)00058-3](https://doi.org/10.1016/S1462-0758(00)00058-3)

689 Kaczmarek, H., Mazaeva, O.A., Kozyreva, E.A., Babicheva, V.A., Tyszkowski, S., Rybchenko,  
690 A.A., Brykała, D., Bartczak, A., Słowiński, M., 2016. Impact of large water level  
691 fluctuations on geomorphological processes and their interactions in the shore zone of a  
692 dam reservoir. *J. Great Lakes Res.* 42, 926–941. <https://doi.org/10.1016/j.jglr.2016.07.024>

693 Kayastha, M.B., Ye, X., Huang, C., Xue, P., 2022. Future rise of the Great Lakes water levels  
694 under climate change. *J. Hydrol.* 612, 128205.  
695 <https://doi.org/10.1016/j.jhydrol.2022.128205>

696 Kopp, R.E., Horton, R.M., Little, C.M., Mitrovica, J.X., Oppenheimer, M., Rasmussen, D.J.,  
697 Strauss, B.H., Tebaldi, C., 2014. Probabilistic 21st and 22nd century sea-level projections at  
698 a global network of tide-gauge sites. *Earth's Futur.* 2, 383–406.  
699 <https://doi.org/10.1002/2014ef000239>

700 Krueger, R., Zoet, L.K., Rawling, J.E., 2020. Coastal Bluff Evolution in Response to a Rapid  
701 Rise in Surface Water Level. *J. Geophys. Res. Earth Surf.* 125, 1–16.  
702 <https://doi.org/10.1029/2019JF005428>

703 Lee, J., Robinson, C., Couture, R.M., 2014. Effect of groundwater-lake interactions on arsenic  
704 enrichment in freshwater beach aquifers. *Environ. Sci. Technol.* 48, 10174–10181.  
705 <https://doi.org/10.1021/es5020136>

706 Lewandowski, J., Meinikmann, K., Nuetzmann, G., Rosenberry, D.O., 2015. Groundwater - the  
707 disregarded component in lake water and nutrient budgets. Part 2: Effects of groundwater  
708 on nutrients. *Hydrol. Process.* 29, 2922–2955. <https://doi.org/10.1002/hyp.10384>

709 Linkhorst, A., Dittmar, T., Waska, H., 2017. Molecular Fractionation of Dissolved Organic  
710 Matter in a Shallow Subterranean Estuary: The Role of the Iron Curtain. *Environ. Sci.*  
711 *Technol.* 51, 1312–1320. <https://doi.org/10.1021/acs.est.6b03608>

712 Luijendijk, A., Hagenaars, G., Ranasinghe, R., Baart, F., Donchyts, G., Aarninkhof, S., 2018.  
713 The State of the World's Beaches. *Sci. Rep.* 8, 1–11. [https://doi.org/10.1038/s41598-018-](https://doi.org/10.1038/s41598-018-24630-6)  
714 [24630-6](https://doi.org/10.1038/s41598-018-24630-6)

715 Malott, S., O'Carroll, D.M., Robinson, C.E., 2016. Dynamic groundwater flows and  
716 geochemistry in a sandy nearshore aquifer over a wave event. *Water Resour. Res.* 52, 5248–  
717 5264. <https://doi.org/10.1002/2014WR015716>

718 McAllister, S.M., Barnett, J.M., Heiss, J.W., Findlay, A.J., MacDonald, D.J., Dow, C.L., Luther,  
719 G.W., Michael, H.A., Chan, C.S., 2015. Dynamic hydrologic and biogeochemical processes  
720 drive microbially enhanced iron and sulfur cycling within the intertidal mixing zone of a  
721 beach aquifer. *Limnol. Oceanogr.* 60, 329–345. <https://doi.org/10.1111/lno.10029>

722 McKeague, J.A., Day, J.H., 1966. Dithionite and oxalate extractable Fe and Al as aids in  
723 differentiating various classes of soil. *Can. J. Soil Sci.* 46, 13–22.  
724 <https://doi.org/10.4141/cjss66-003>

725 Mehra, O.P., 1958. Iron oxide removal from soils and clays by a dithionite-citrate system  
726 buffered with sodium bicarbonate. *Clays Clay Miner.* 7, 317–327.  
727 <https://doi.org/10.1346/ccmn.1958.0070122>

728 Meile, C., Porubsky, W.P., Walker, R.L., Payne, K., 2010. Natural attenuation of nitrogen  
729 loading from septic effluents: Spatial and environmental controls. *Water Res.* 44, 1399–  
730 1408. <https://doi.org/10.1016/j.watres.2009.11.019>

731 Mitchell, M., Isdell, R.E., Herman, J., Tombleson, C., 2021. Impact Assessment and  
732 Management Challenges of Key Rural Human Health Infrastructure Under Sea Level Rise.  
733 *Front. Mar. Sci.* 8, 1–8. <https://doi.org/10.3389/fmars.2021.631757>

734 Moore, W.S., 2010. The effect of submarine groundwater discharge on the ocean. *Ann. Rev.*  
735 *Mar. Sci.* 2, 59–88. <https://doi.org/10.1146/annurev-marine-120308-081019>

736 Mukhopadhyay, A., Ghosh, P., Chanda, A., Ghosh, A., Ghosh, S., Das, S., Ghosh, T., Hazra, S.,  
737 2018. Threats to coastal communities of Mahanadi delta due to imminent consequences of  
738 erosion – Present and near future. *Sci. Total Environ.* 637–638, 717–729.  
739 <https://doi.org/10.1016/j.scitotenv.2018.05.076>

740 Nielsen, P., 2009. *Coastal and Estuarine Processes*. World Scientific, Singapore.  
741 <https://doi.org/10.1142/7114>

742 Nisbeth, C.S., Jessen, S., Bennike, O., Kidmose, J., Reitzel, K., 2019. Role of groundwater-borne  
743 geogenic phosphorus for the internal P release in shallow lakes. *Water (Switzerland)* 11, 1–  
744 16. <https://doi.org/10.3390/w11091783>

745 Niswonger, R.G., Panday, S., Motomu, I., 2011. MODFLOW-NWT , A Newton formulation for  
746 MODFLOW-2005. USGS reports 44.

747 Rakhimbekova, S., O’Carroll, D.M., Andersen, M.S., Wu, M.Z., Robinson, C.E., 2018. Effect of  
748 transient wave forcing on the behavior of arsenic in a nearshore aquifer. *Environ. Sci.*  
749 *Technol.* 52, 12338–12348. <https://doi.org/10.1021/acs.est.8b03659>

750 Rakhimbekova, S., O’Carroll, D.M., Oldfield, L.E., Ptacek, C.J., Robinson, C.E., 2021a.  
751 Spatiotemporal controls on septic system derived nutrients in a nearshore aquifer and their  
752 discharge to a large lake. *Sci. Total Environ.* 752.  
753 <https://doi.org/10.1016/j.scitotenv.2020.141262>

754 Rakhimbekova, S., O’Carroll, D.M., Robinson, C.E., 2021b. Factors controlling phosphorus  
755 mobility in nearshore aquifers adjacent to large lakes. *Sci. Total Environ.* 799, 149443.  
756 <https://doi.org/10.1016/j.scitotenv.2021.149443>

757 Rakhimbekova, S., O'Carroll, D.M., Robinson, C.E., 2021c. Occurrence of arsenic in nearshore  
758 aquifers adjacent to large inland lakes. *Environ. Sci. Technol.*  
759 <https://doi.org/10.1021/acs.est.1c02326>

760 Robinson, C., 2015. Review on groundwater as a source of nutrients to the Great Lakes and their  
761 tributaries. *J. Great Lakes Res.* 41, 941–950. <https://doi.org/10.1016/j.jglr.2015.08.001>

762 Robinson, C.E., Xin, P., Santos, I.R., Charette, M.A., Li, L., Barry, D.A., 2018. Groundwater  
763 dynamics in subterranean estuaries of coastal unconfined aquifers: Controls on submarine  
764 groundwater discharge and chemical inputs to the ocean. *Adv. Water Resour.* 115, 315–331.  
765 <https://doi.org/10.1016/j.advwatres.2017.10.041>

766 Rosenberry, D.O., Lewandowski, J., Meinikmann, K., Nützmänn, G., 2015. Groundwater - the  
767 disregarded component in lake water and nutrient budgets . Part 1 : effects of groundwater  
768 on hydrology. *Hydrol. Process.* 2921, 2895–2921. <https://doi.org/10.1002/hyp.10403>

769 Roy, J.W., 2019. Endobenthic organisms exposed to chronically high chloride from groundwater  
770 discharging along freshwater urban streams and lakeshores. *Environ. Sci. Technol.* 53,  
771 9389–9397. <https://doi.org/10.1021/acs.est.9b02288>

772 Roy, J.W., Malenica, A., 2013. Nutrients and toxic contaminants in shallow groundwater along  
773 Lake Simcoe urban shorelines. *Inl. Waters.* <https://doi.org/10.5268/IW-3.2.521>

774 Schwertmann, U., 1964. Differenzierung der eisenoxide des bodens durch extraktion mit  
775 ammonium-losung. <https://doi.org/10.1002/jpln.3591050303>

776 Sellinger, C.E., Stow, C.A., Lamon, E.C., Qian, S.S., 2008. Recent water level declines in the  
777 Lake Michigan-Huron system. *Environ. Sci. Technol.* 42, 367–373.

778 <https://doi.org/10.1021/es070664+>

779 Sirois, M., Couturier, M., Barber, A., G elinas, Y., Chaillou, G., 2018. Interactions between iron  
780 and organic carbon in a sandy beach subterranean estuary. *Mar. Chem.* 202, 86–96.  
781 <https://doi.org/10.1016/j.marchem.2018.02.004>

782 Small, C., Nicholls, R.J., 2003. A global analysis of human settlement in coastal zones. *J. Coast.*  
783 *Res.* 19, 584–599.

784 Spiteri, C., Regnier, P., Slomp, C.P., Charette, M.A., 2006. pH-Dependent iron oxide  
785 precipitation in a subterranean estuary. *J. Geochemical Explor.* 88, 399–403.  
786 <https://doi.org/10.1016/j.gexplo.2005.08.084>

787 Spiteri, C., Slomp, C.P., Charette, M.A., Tuncay, K., Meile, C., 2008. Flow and nutrient  
788 dynamics in a subterranean estuary (Waquoit Bay, MA, USA): Field data and reactive  
789 transport modeling. *Geochim. Cosmochim. Acta* 72, 3398–3412.  
790 <https://doi.org/10.1016/j.gca.2008.04.027>

791 Stefania, G.A., Rotiroti, M., Buerge, I.J., Zanotti, C., Nava, V., Leoni, B., Fumagalli, L.,  
792 Bonomi, T., 2019. Identification of groundwater pollution sources in a landfill site using  
793 artificial sweeteners, multivariate analysis and transport modeling. *Waste Manag.* 95, 116–  
794 128. <https://doi.org/10.1016/j.wasman.2019.06.010>

795 Swenson, M.J., Wu, C.H., Edil, T.B., Mickelson, D.M., 2006. Bluff recession rates and wave  
796 impact along the Wisconsin coast of Lake Superior. *J. Great Lakes Res.* 32, 512–530.  
797 [https://doi.org/10.3394/0380-1330\(2006\)32\[512:BRRAWI\]2.0.CO;2](https://doi.org/10.3394/0380-1330(2006)32[512:BRRAWI]2.0.CO;2)

798 Theuerkauf, E.J., Braun, K.N., Nelson, D.M., Kaplan, M., Vivirito, S., Williams, J.D., 2019.

799 Coastal geomorphic response to seasonal water-level rise in the Laurentian Great Lakes: An  
800 example from Illinois Beach State Park, USA. *J. Great Lakes Res.* 45, 1055–1068.  
801 <https://doi.org/10.1016/j.jglr.2019.09.012>

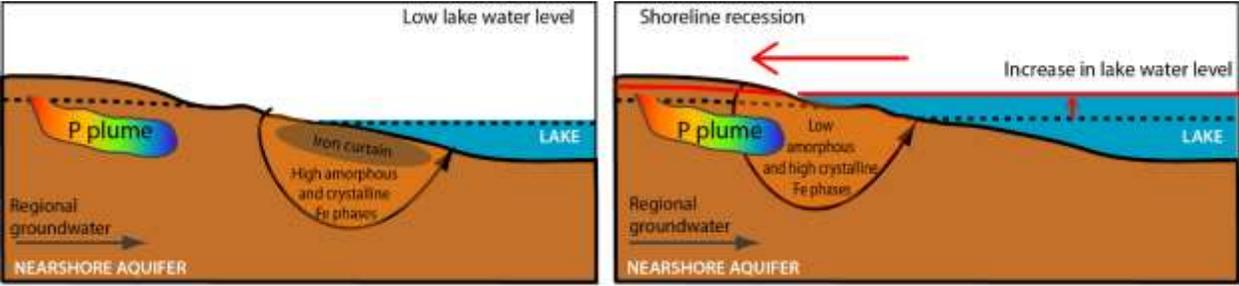
802 US EPA, 1996. Method 3050B Acid digestion of sediments, sludges and soils, Method 3050B  
803 Acid digestion of sediments, sludges and soils.

804 Vitousek, S., Barnar, P.L., Fletcher, C.H., Frazer, N., Storlazzi, C.D., 2017. Doubling of coastal  
805 flooding frequency within decades due to sea-level rise 1–9. [https://doi.org/10.1038/s41598-](https://doi.org/10.1038/s41598-017-01362-7)  
806 [017-01362-7](https://doi.org/10.1038/s41598-017-01362-7)

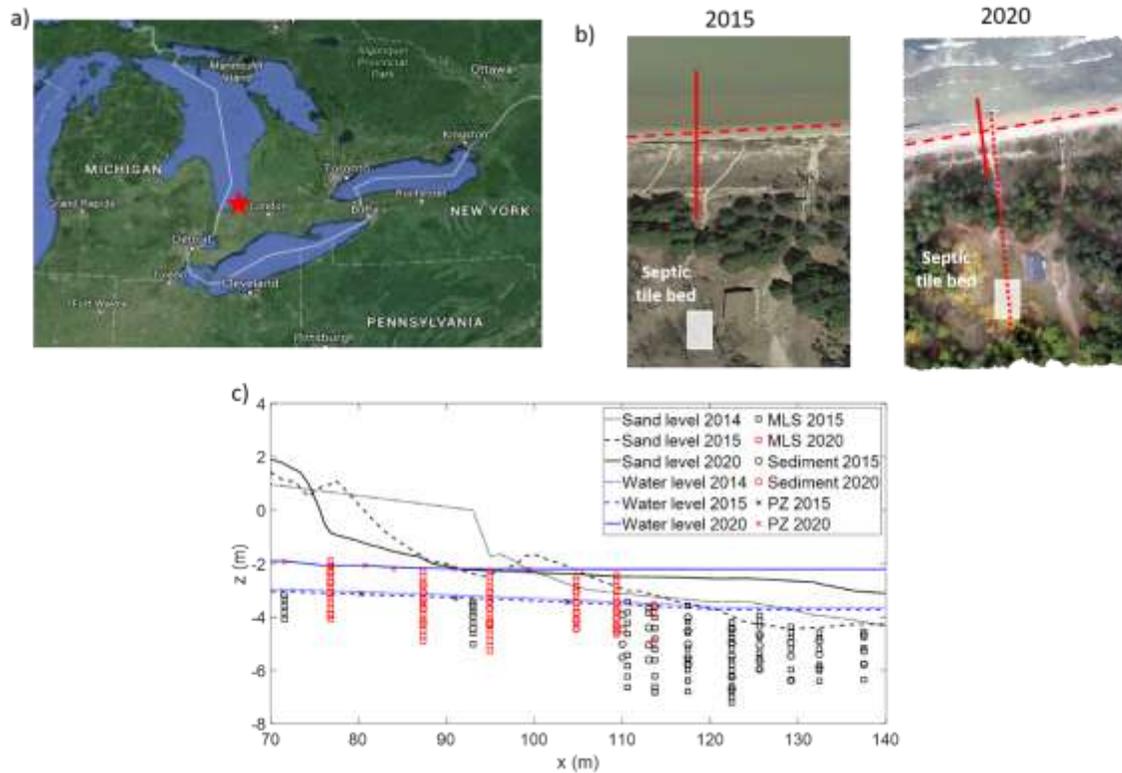
807 Vogel, L.J., O’Carroll, D.M., Edge, T.A., Robinson, C.E., 2016. Release of *Escherichia coli* from  
808 Foreshore Sand and Pore Water during Intensified Wave Conditions at a Recreational  
809 Beach. *Environ. Sci. Technol.* 50, 5676–5684. <https://doi.org/10.1021/acs.est.6b00707>

810 Vousdoukas, M.I., Ranasinghe, R., Mentaschi, L., Plomaritis, T.A., Athanasiou, P., Luijendijk,  
811 A., Feyen, L., 2020. Sandy coastlines under threat of erosion. *Nat. Clim. Chang.* 10, 260–  
812 263. <https://doi.org/10.1038/s41558-020-0697-0>

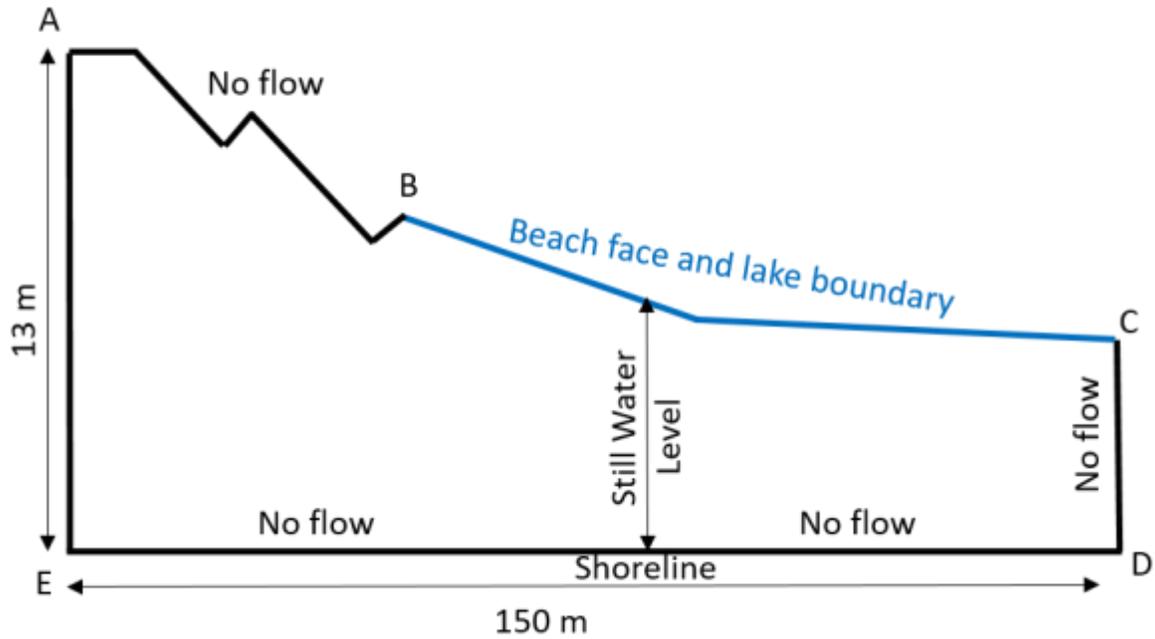
813 Yi, S., Sun, W., Heki, K., Qian, A., 2015. An increase in the rate of global mean sea level rise  
814 since 2010. *Geophys. Res. Lett.* 42, 3998–4006. <https://doi.org/10.1002/2015GL063902>  
815  
816



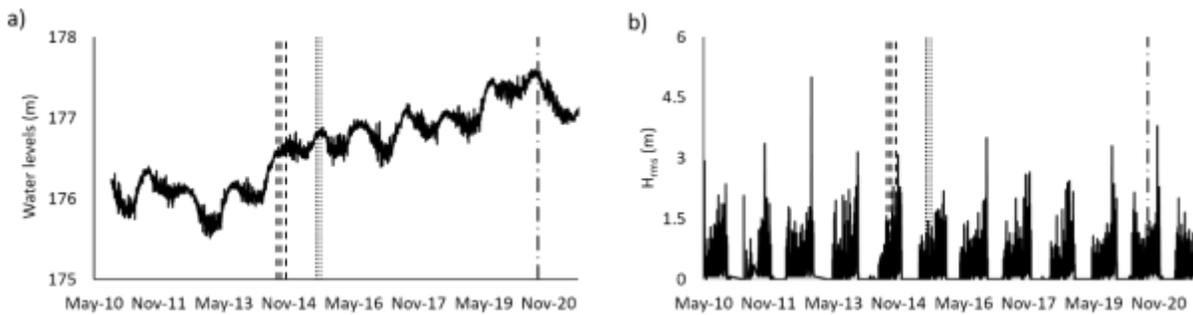
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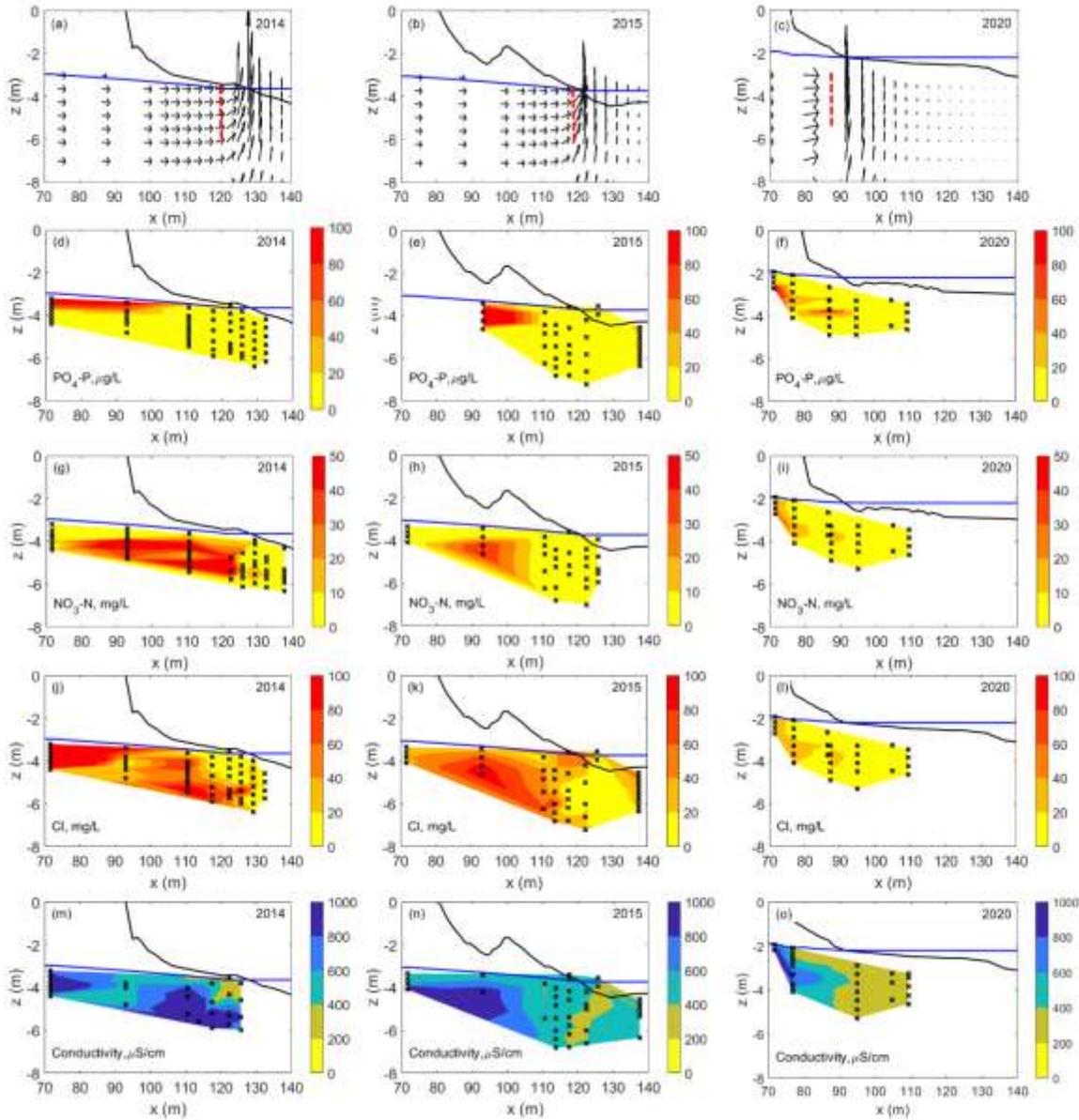
**Figure 1.** a) Map of Laurentian Great Lakes Basin with the red star indicating the location of the study site. b) Aerial image of the study site using Google Earth imagery in 2015 and Drone (DJI Mavic Air) image in 2020, red dashed lines indicate the shoreline location in each image, solid red lines indicate the cross-shore monitoring transect location, and red dotted line indicates the ERT survey line. The location of septic tile bed is marked with a white rectangle. c) Layout of monitoring equipment along the cross-shore transect for sampling events in 2014 and 2015, and in 2020. Sand and water levels measured in 2014, 2015 and 2020 are also shown in (c).



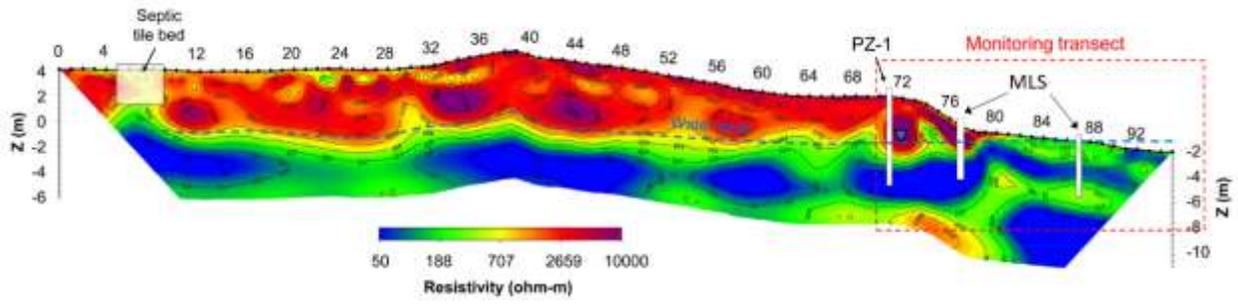
**Figure 2.** Numerical groundwater model domain including flow boundary conditions.



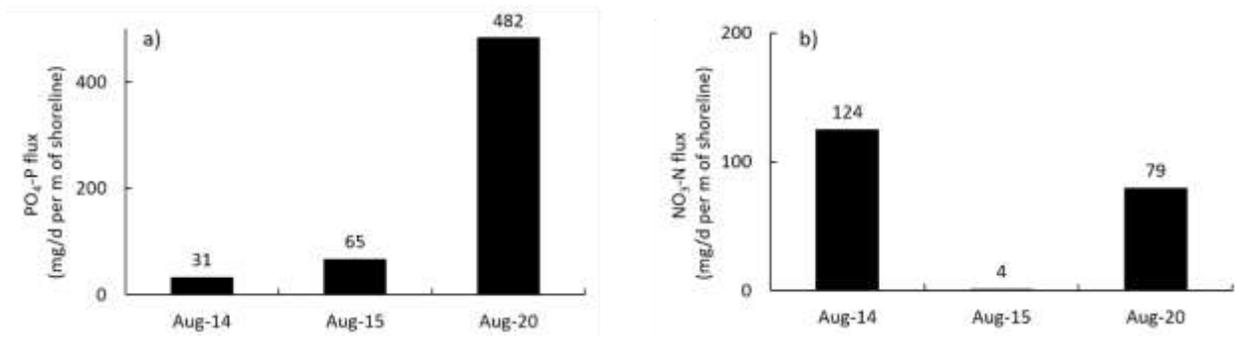
**Figure 3.** a) Average daily lake water levels measured at offshore buoy located 70 km north of the study site (Lake Huron at Goderich (02FE012) with elevation based on International Great Lakes Datum 1985 (Fisheries and Oceans Canada, 2020). b) Offshore root mean square wave height ( $H_{rms}$ ) measured at a buoy (C45149 Southern Lake H) located ~35 km offshore from the field site. The vertical black dashed lines in a) and b) indicate days of field sampling events.



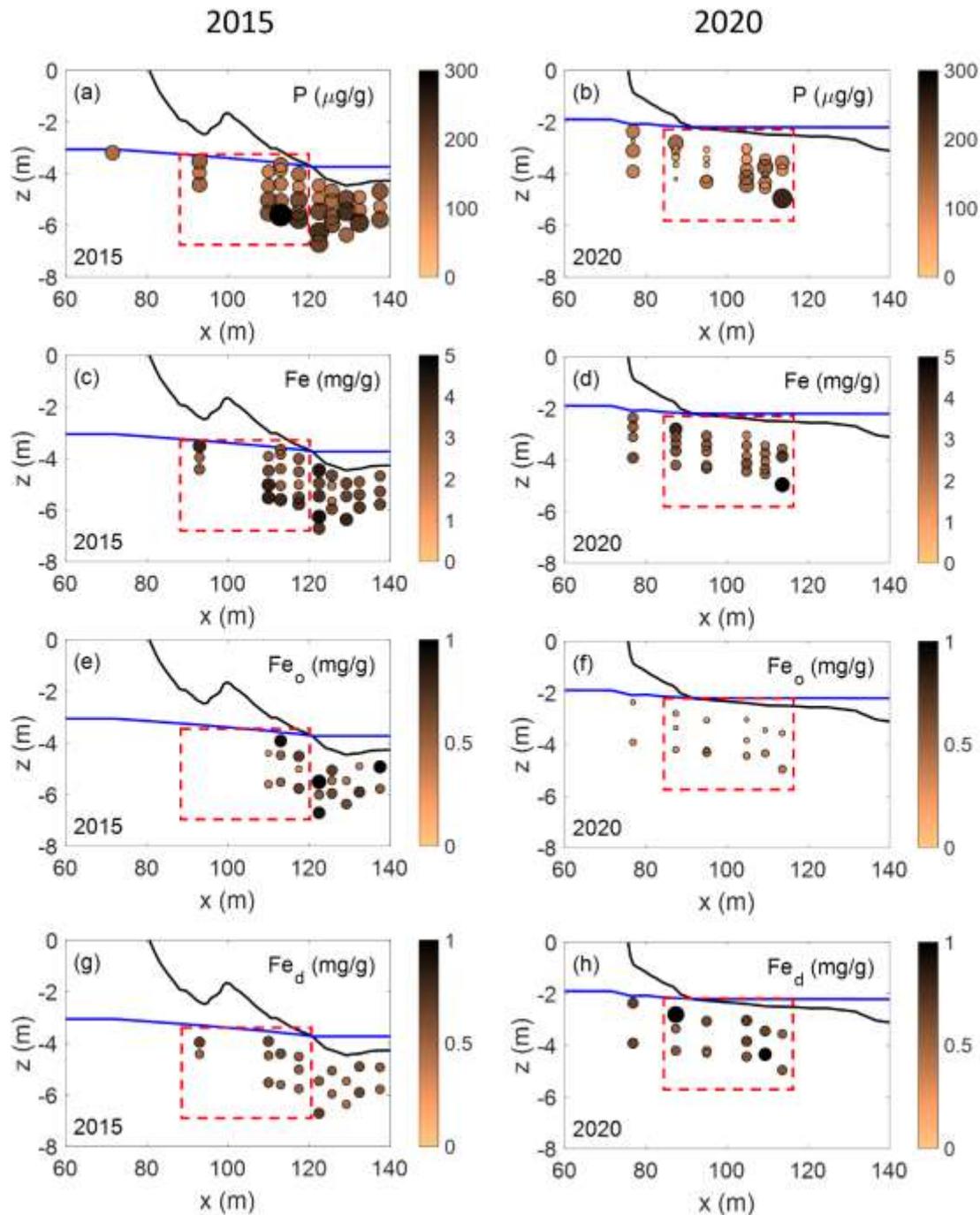
**Figure 4.** Simulated groundwater flow velocities based on the hydrological conditions during the a) 2014, b) 2015, and c) 2020 sampling periods. Distribution of d-f)  $\text{PO}_4\text{-P}$ , g-i)  $\text{NO}_3\text{-N}$ , j-l) chloride (Cl) and m-o) electrical conductivity (EC) in 2014, 2015 and 2020. For all subplots, the solid black line represents the sand surface elevation and the solid blue line represents the measured water levels for each sampling event. Red dashed lines in a-c) indicate lines across which  $\text{PO}_4\text{-P}$  and  $\text{NO}_3\text{-N}$  fluxes were calculated using the simulated groundwater fluxes and interpolated  $\text{PO}_4\text{-P}$  and  $\text{NO}_3\text{-N}$  concentrations ( $x = 120.3$  m in 2014 and 2015 and  $x = 88.7$  in 2020). Colored contours in d-o) represent the concentrations of  $\text{PO}_4\text{-P}$ ,  $\text{NO}_3\text{-N}$ , Cl and EC with the color bars shown on the right hand side, and black crosses show pore water sampling locations.



**Figure 5.** Cross-sectional resistivity image from the ERT survey. The septic drainage field, PZ-1 and MLS are indicated, along with the inferred water table (blue dashed line). The extent of the cross-shore monitoring transect is defined by the red dashed box.



**Figure 6.** Estimated a) PO<sub>4</sub>-P and b) NO<sub>3</sub>-N flux directed towards the sediment-water interface for August 2014, 2015 and 2020 sampling events.



**Figure 7.** Distribution of a, e) solid-phase P, b, f) solid phase Fe, c, g) oxalate-extractable, amorphous Fe phases and d, h) dithionite-extractable, more crystalline Fe phases in 2015 and 2020. Rows show the year of sampling event when sediment sample was collected and columns show the different P and Fe solid phase extraction data. The bubble sizes and color bar indicate the magnitude of the solid-phase concentration. The solid black line represents the sand surface elevation, and the solid blue line represents the measured water levels at each site. Only the samples within the red box were considered in comparing the abundance of solid-phase P and Fe between sampling years.

Supporting Information

**Potential for shoreline recession to accelerate pollutant fluxes across the groundwater-coastal water interface**

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**7 pages**

<b>Contents</b>	<b>Page</b>
1. Sediment extraction method	S3
2. Numerical model set up	S4
3. PO <sub>4</sub> -P and NO <sub>3</sub> -N flux calculations	S5
4. Correlation between NO <sub>3</sub> -N and Electrical Conductivity (EC)	S6
5. References	S7

## **S1. Sediment extraction methods**

Ammonium oxalate and sodium dithionite sediment extractions were conducted on select sediments samples collected in 2015 and 2020 to characterize different Fe-bearing solid phase pools. For the sodium dithionite extraction, 4 g of sediment samples was placed in a 250 mL centrifuge bottle and 40 mL of 0.3 M Na-citrate solution and 5 mL of 1 M NaHCO<sub>3</sub> solution were added (Mehra, 1958). The solution was treated in a water bath at 80°C and 1 g of solid Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub> was added to flocculate the sample, after which the solution was stirred continuously for 1 min and then occasionally for a total of 15 min. The solution was then centrifuged at 2000 rpm for 5 minutes, after which the clear supernatant was decanted, filtered and analyzed for dissolved Fe using flame atomic absorption spectroscopy (FAAS, Agilent 200 Series). For the ammonium oxalate extraction, 1 g of sediment sample was placed in a 250 mL centrifuge bottle, 40 mL of acid oxalate solution was added and then it was placed in mechanical shaker for 4 hour in the dark (McKeague and Day, 1966). The solution was then centrifuged at 2400 rpm for 10 minutes. The clear supernatant solution was then decanted, filtered and analyzed for dissolved Fe using FAAS.

## S2. Numerical model set up

Wave set-up approach has been shown to represent the main effects of waves on groundwater flow dynamics while reducing computational requirements (Robinson et al., 2014; Xin et al., 2010). Using this approach, the time-varying specified heads applied along the lake boundary (BC Figure 3) were calculated using an empirical wave setup formula developed by Nielsen (Nielsen, 2009) and given as:

$$\bar{\eta} = \frac{0.4H_{rms}}{1+10\frac{D+\bar{\eta}}{H_{rms}}} \quad (1)$$

where  $\eta$  (m) is the increase in water level above the still water level due to wave setup;  $H_{rms}$  (m) is the offshore root mean square wave height obtained from the Fisheries and Ocean Canada wave buoy (Figure 2b); and  $D$  (m) is the distance from the sediment water interface to the still water level and is a function of location across the beach face. Submerged nodes along the lake boundary were assigned a hydrostatic pressure corresponding to the wave setup profile. This was implemented using a modified form of the periodic boundary condition (PBC) package (Post, 2011) to allow a time-varying moving interface to be implemented. Nodes landward of the wave setup point along the lakeward boundary were represented as a no-flow boundary.

### **S3. PO<sub>4</sub>-P and NO<sub>3</sub>-N flux calculations**

A steady state groundwater flow model was used to simulate the water flux directed towards the lake. The PO<sub>4</sub>-P and NO<sub>3</sub>-N fluxes near the shoreline were estimated by multiplying the simulated groundwater flux across a vertical line near the shoreline by field measured PO<sub>4</sub>-P and NO<sub>3</sub>-N concentrations (interpolated to a depth of 3 m below beach face). This calculation was performed for the simulated groundwater flux in 2014, 2015 and 2015 and for field measured PO<sub>4</sub>-P and NO<sub>3</sub>-N concentrations in each of these years with the location of the vertical line at  $x = 120.3$  m in 2014 and 2015 and  $x = 88.7$  m in 2020. The limitation of this calculation is that it does not consider that the measured PO<sub>4</sub>-P and NO<sub>3</sub>-N concentrations may be modified by dilution by surface water at shallow depth below the groundwater-lake interface.

#### S4. Correlation between NO<sub>3</sub>-N and Electrical Conductivity (EC).

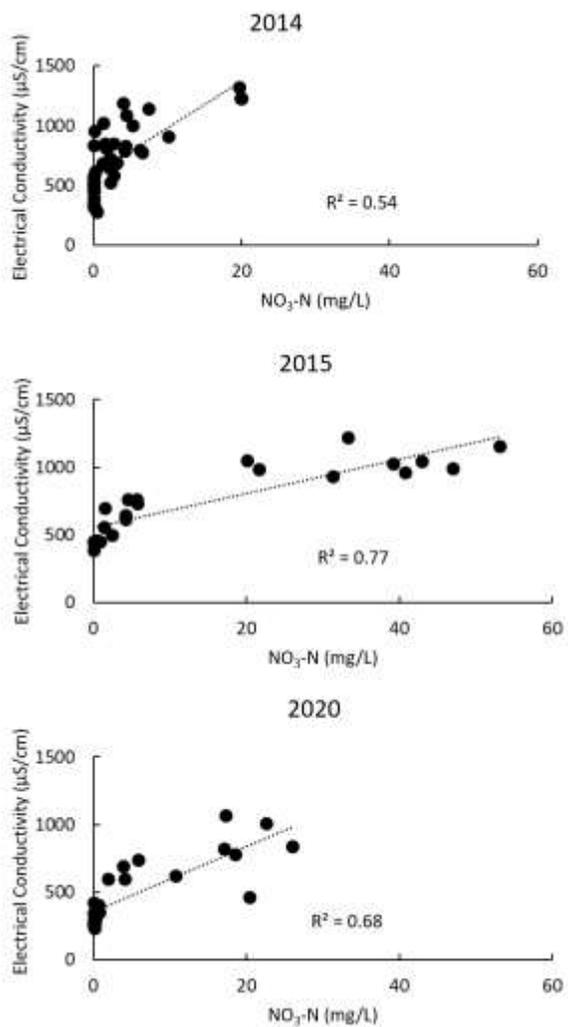


Figure S1. Correlation between NO<sub>3</sub>-N and electrical conductivity (EC) in 2014, 2015 and 2020.

## References

- McKeague, J.A., Day, J.H., 1966. Dithionite and oxalate extractable Fe and Al as aids in differentiating various classes of soil. *Can. J. Soil Sci.* 46, 13–22.  
<https://doi.org/10.4141/cjss66-003>
- Mehra, O.P., 1958. Iron oxide removal from soils and clays by a dithionite-citrate system buffered with sodium bicarbonate. *Clays Clay Miner.* 7, 317–327.  
<https://doi.org/10.1346/ccmn.1958.0070122>
- Nielsen, P., 2009. *Coastal and Estuarine Processes*. World Scientific Publishing Co Pte Ltd, Singapore.
- Post, V.E.A., 2011. A new package for simulating periodic boundary conditions in MODFLOW and SEAWAT. *Comput. Geosci.* 37, 1843–1849.  
<https://doi.org/10.1016/j.cageo.2011.01.012>
- Robinson, C., Xin, P., Li, L., Barry, D.A., 2014. Groundwater flow and salt transport in a subterranean estuary driven by intensified wave conditions. *Water Resour. Res.* 50, 165–181.  
<https://doi.org/10.1002/2013WR013813>
- Xin, P., Robinson, C., Li, L., Barry, D.A., Bakhtyar, R., 2010. Effect of wave forcing on a subterranean estuary. *Water Resour. Res.* 46, 1–17.  
<https://doi.org/10.1016/j.advwatres.2006.07.006>