

1 **Deciphering the significant role of biological ice nucleators in precipitation at the**
2 **organic molecular level**

3 **Mutong Niu¹, Wei Hu^{1,*}, Shu Huang¹, Jie Chen², Shujun Zhong¹, Ziye Huang¹, Peimin**

4 Duan¹, Xiangyu Pei³, Jing Duan⁴, Kai Bi⁵, Shuang Chen¹, Rui Jin¹, Ming Sheng¹, Ning

5 Yang¹, Libin Wu¹, Junjun Deng¹, Jialei Zhu¹, Fangxia Shen⁶, Zhijun Wu⁷, Daizhou Zhang⁸,

6 Pingqing Fu^{1,*}

7 ¹Institute of Surface-Earth System Science, School of Earth System Science, Tianjin University,
8 Tianjin 300072, China, ²Institute for Atmospheric and Climate Science, ETH Zürich, Zurich
9 8092, Switzerland, ³College of Environmental and Resource Sciences, Zhejiang Provincial Key
10 Laboratory of Organic Pollution Process and Control, Zhejiang University, Hangzhou 310058,
11 China, ⁴State Key Laboratory of Severe Weather & Key Laboratory for Cloud Physics, Chinese
12 Academy of Meteorological Sciences, Beijing 100081, China, ⁵Beijing Key Laboratory of
13 Cloud, Precipitation, and Atmospheric Water Resources, Beijing Weather Modification Office
14 (BWMO), Beijing 100089, China, ⁶School of Energy and Power Engineering, Beihang
15 University, Beijing 102206, China, ⁷State Key Joint Laboratory of Environmental Simulation
16 and Pollution Control, College of Environmental Sciences and Engineering, Peking University,
17 Beijing 100081, China, ⁸Faculty of Environmental and Symbiotic Sciences, Prefectural
18 University of Kumamoto, Kumamoto 862-8502, Japan

19

20 Corresponding author: W. Hu (huwei@tju.edu.cn), P. Fu (fupingqing@tju.edu.cn)

- 21
- 22 **Key Points:**
- 23 • Biological materials predominated in ice nucleating particles (INPs) identified in
24 precipitation.
- 25 • Sulfur-containing compounds, associated with specific microbial taxa, likely acted as
26 efficient INPs in precipitation.
- 27 • This study preliminarily established an understanding of the connections between INPs,
28 organic matter, and microorganisms.
- 29

30 Abstract

31 Biological particles, as a fraction of organic particles, potentially play a crucial role in ice
32 nucleation processes. However, the contributions and relationships of biological components and
33 organic matter (OM) to atmospheric ice nucleation are still largely unexplored. Here, droplet
34 freezing assays, high-throughput sequencing technology and ultrahigh-resolution mass
35 spectrometry were performed to detect the INPs, microorganisms and OM molecules in
36 precipitation collected at the summit of Mt. Lu, China, respectively. Results revealed a
37 predominant biological composition (71.7% and 93.2%) of total and nanoscale INPs (<0.22 µm)
38 at temperatures above -15°C. Specifically, bacterial INPs accounted for 36.1% of the biological
39 INPs at temperatures above -15°C. A notable correlation between sulfur-containing compounds,
40 mainly proteinaceous and lignin-like substances, and INPs was uncovered, with a co-occurrence
41 network linking these compounds to Gram-positive bacteria and Agaricomycetes. This study
42 underscored the possible significance of sulfur-containing compounds in biological INP
43 efficiency, which could further help shed light on the ice nucleation mechanisms and potential
44 sources of biological INPs.

45

46 Plain Language Summary

47 Ice nucleating particles (INPs) are particles that facilitate the freezing of water at temperatures
48 above the homogenous freezing point, impacting cloud formation and precipitation processes in
49 the atmosphere. This study identified different types of INPs, microbes and organic matter in
50 precipitation sampled from Mt. Lu in southeastern China and investigated the connections
51 between them. The findings suggested that a significant portion of INPs were of biological
52 origin. Sulfur-containing compounds likely played an important role in ice nucleation, which
53 may originate from certain microbial taxa. This study will help us understand the role of
54 microbes and organic molecules in ice formation, which has broader implications in areas like
55 preserving biological materials at low temperatures or even facilitating artificial snow
56 production.

57

58 1. Introduction

59 Ice nucleation and subsequent ice crystal formation in clouds can affect the global hydrological
60 cycle by altering cloud and precipitation formation, as well as affect the Earth's solar radiation
61 balance (DeMott et al., 2010; Lohmann & Feichter, 2005). As a result, ice nucleating particles
62 (INPs) which aid the heterogeneous ice nucleation have received a great deal of attention (Kanji
63 et al., 2017; Vali et al., 2015). To date, biological materials, mineral dust, crystalline salts,
64 volcanic ash and carbonaceous combustion products (e.g., soot particles) have been identified as
65 INPs (Hoose & Möhler, 2012; Kilchhofer et al., 2021; Möhler et al., 2008; Wise et al., 2012).
66 Among them, biological materials (e.g., bacteria, fungal spores) or biological macromolecules
67 (e.g., proteinaceous substances, saccharides, and lipids) exhibit the highest ice nucleation activity
68 (INA) (Conen et al., 2011; Pummer et al., 2015). Some bacterial INPs, notably *Pseudomonas*
69 *syringae* (Arny et al., 1976), *Ps. fluorescens* (Obata et al., 1987), *Ps. viridiflava* (Obata et al.,
70 1989), *Erwinia ananas* (Hew & Yang, 1992), *Er. uredovora* (Obata et al., 1989) and
71 *Xanthomonas campestris* (Kim et al., 1987), were particularly effective at promoting ice

72 nucleation at above -10°C . Certain strains of *Ps. syringae*, known as highly active ice nucleators,
73 even can initiate freezing at temperatures as high as -1.8°C (Maki et al., 1974). Fungi such as
74 certain strains of *Fusarium* sp. (Kunert et al., 2019), *Mortierella alpine* (Iannone et al., 2011),
75 and *Cladosporium* spores (Fröhlich-Nowoisky et al., 2015) have also demonstrated INA. The ice
76 nucleation mechanisms in these microbial INPs are associated with proteins and polysaccharides
77 (Morris et al., 2013). Additional biological materials, including pollen, lichens, and algae, are
78 capable of initiating freezing at elevated temperatures (Christner, 2010; Duan et al., 2023;
79 Karimi et al., 2019). Due to the remarkable INA of biological INPs (bio-INPs), the
80 concentrations, compositions and influencing factors of bio-INPs in various environments have
81 been widely documented (Che et al., 2019; Chen, Wu, Wu, et al., 2021; Huang et al., 2021;
82 Pereira et al., 2021).

83 Bio-INPs, as a part of organic INPs, have received limited attention regarding their INA from the
84 perspective of organic matter (OM), especially at the molecular level. Past studies have primarily
85 assessed the activities of bulk OM in environmental samples, e.g., marine aerosols (Wilson et al.,
86 2015) and agricultural soil dust (Conen et al., 2011; Hill et al., 2016; O'Sullivan et al., 2014;
87 Tobo et al., 2014), utilizing heating treatment or hydrogen peroxide digestion combined with
88 droplet freezing measurements. Diverse OM components, such as cellulose (Hiranuma et al.,
89 2019), lipids (Steinke et al., 2020), lignin (Bogler & Borduas-Dedekind, 2020), humic or fulvic
90 acids (Klumpp et al., 2022; Wang & Knopf, 2011), humic-like substances (HULIS) (Chen, Wu,
91 Zhao, et al., 2021) and proteinaceous materials (Christner, 2010; Watabe et al., 1993), were
92 found to possess significant INA. The INA of these OM components exhibits considerable
93 variation. For instance, the average freezing temperatures of fulvic acid and lignins ranged from
94 -13°C to -8°C (Borduas-Dedekind et al., 2019), while the ice nucleation temperature of
95 cellulose or polysaccharides (i.e., starches) was lower than -20°C (Hiranuma et al., 2019;
96 Hiranuma et al., 2015; Steinke et al., 2020). Some proteinaceous materials derived from
97 microorganisms exhibited notably efficient INA (Cascajo-Castresana et al., 2020; Maki et al.,
98 1974; Schnell & Vali, 1976). Specifically, iron storage proteins and ice nucleating proteins
99 initiated ice formation at -4°C and -1.8°C , respectively (Cascajo-Castresana et al., 2020; Maki
100 et al., 1974; Schnell & Vali, 1976). Nevertheless, the precise molecular-level compositions of
101 bio-INPs remain comparatively less studied, requiring further investigation to elucidate the
102 underlying ice nucleation mechanisms.

103 Ice crystal formation in clouds plays a crucial role in Earth's radiative balance and hydrological
104 cycle, prompting extensive global research on INPs in clouds and precipitation (Failor et al.,
105 2022; Petters & Wright, 2015). Previous studies have utilized the abundances of INPs in various
106 precipitation samples, such as rain, snow, fog, and hail, to estimate INPs in clouds (Beall et al.,
107 2020; Martin et al., 2019; Michaud et al., 2014). While precipitation samples may collect
108 additional INPs in the below-cloud atmosphere and undergo heterogeneous chemistry through
109 gas adsorption or absorption (Lim et al., 2010), they largely resemble cloud water, particularly in
110 clean areas (Hu et al., 2017). INPs in precipitation from clean and high-altitude regions face
111 fewer environmental interferences, offering a more direct reflection of cloud ice nucleation
112 processes. Additionally, collecting precipitation samples is more convenient than cloud water
113 collection for characterizing atmospheric water properties.

114 As a result, numerous studies have focused on INP concentrations in precipitation at
115 mountainous sites (Ahern et al., 2007; Stopelli et al., 2017). Conen et al. (2015) revealed

116 seasonal variations in INP concentrations at -8°C at Jungfraujoch, with higher levels in summer.
117 The INPs increases coincided with high wind speeds and air masses with little or no precipitation
118 (Stopelli et al., 2016). Wind direction and air mass trajectories significantly affected INP
119 concentrations and biogenic contributions at Jungfraujoch (Creamean et al., 2019) and
120 southeastern Louisiana (Joyce et al., 2019). In some instances, anthropogenic factors, like PM_{10}
121 and rain acidification, have influenced INP concentrations (Lacher et al., 2018; Pouzet et al.,
122 2017). The Sierra Nevada Mountains in California, renowned for their orographic cloud
123 formation, have undergone thorough studies regarding INP characteristics (Creamean et al., 2014;
124 Creamean et al., 2016). Long-distance transported dust and biological particles were key
125 components of INPs, significantly influencing orographic precipitation in the western United
126 States (Creamean et al., 2013). At Puy de Dôme, biological materials also constituted major
127 components of INPs, representing over 60% at -10°C (Joly et al., 2014; Testa et al., 2021).
128 Nevertheless, microorganisms with known ice nucleation ability represented only a minority of
129 bio-INPs in precipitation (Zhang et al., 2020), and many unknown bio-INPs remain to be
130 explored. Additionally, nanoscale biological fragments have been found to possess ice nucleation
131 capabilities in precipitation samples (O'Sullivan et al., 2015; Wilson et al., 2015). It is important
132 to note that organisms, as a fraction of organic matter, continuously release various organic
133 substances during their metabolic processes. It has been suggested that organic matter associated
134 with biological substances may have high ice nucleating abilities in soils (Tobo et al., 2019).
135 However, the specific connection between organic molecules and bio-INPs at the molecular level
136 in precipitation remains unexplored.

137 Mt. Lu, situated in the middle and lower reaches of Yangtze River in China, features high
138 altitudes and deep valleys, resulting in significant vertical and horizontal climatic variations. Due
139 to its unique geography, Mt. Lu frequently experiences meteorological phenomena such as rime,
140 snow rime, and rain rime (Guo et al., 2019). Therefore, Mt. Lu has been the site of
141 comprehensive observational studies focusing on the physical properties of precipitation and
142 aerosols (Huang et al., 2018; Li et al., 2015; Sun et al., 2016). Here, to assess the abundances of
143 INPs in clouds and the contribution of biological and organic fractions to INPs as well as their
144 associations, precipitation (including rain and fog water) samples were collected from a
145 relatively clean and high-altitude mountain site, Mt. Lu, in southeastern China. The abundances
146 and compositions of INPs including bio-INPs, bacterial INPs, nanoscale INPs (nano-INPs) and
147 biological nano-INPs were determined by droplet freezing assays coupled with corresponding
148 pretreatments. Microbial and organic compositions were measured with high-throughput
149 sequencing technology and ultrahigh-resolution Fourier transform ion cyclotron resonance mass
150 spectrometry (FT-ICR MS), respectively. This study aims to provide more information on the
151 components, sources and potential atmospheric processes of INPs, and especially to clarify the
152 relationships among INPs, microorganisms and organic molecules.

153 2. Materials and Methods

154 2.1. Observation site and sample collection

155 Precipitation including rainwater and fog water samples were collected at the summit of Mt. Lu
156 (29.58°N , 115.98°E , 1165 m a.s.l.) from November to December 2019. Mt. Lu is located in the
157 southeast of China and it has frequent cloudy weather and rainfall events throughout the year due
158 to the influence of the subtropical monsoon. Fog water samples were collected into Teflon

159 bottles using a Caltech Active Cloudwater Collector (CASCC2) following the operation
160 protocol. Rainwater samples were deposited directly into Teflon beakers. After collection, the
161 samples were transferred to brown polyethylene bottles and stored at -20°C . Meteorological data
162 were recorded by an automatic meteorological monitoring station (Gill MetPak, UK), and hourly
163 air quality data including air quality index (AQI), $\text{PM}_{2.5}$, PM_{10} , NO_2 , O_3 , SO_2 and CO were
164 obtained from the nearest air quality monitoring station (29.57°N , 115.98°E), which is 20 m
165 away from the sampling site. All air quality data were downloaded from China National
166 Environmental Monitoring Center (<http://www.bjmemc.com.cn/>). Detailed information about
167 sampling periods is shown in **Figure S1** and **Table S1**.

168 2.2. Droplet freezing assays

169 Total INPs, heat-sensitive INPs, lysozyme-sensitive INPs, nano-INPs ($<0.22\ \mu\text{m}$) and heat-
170 sensitive nano-INPs were determined by droplet freezing assays coupled with corresponding
171 pretreatments with a modified instrument based on Chen et al. (2018). Five aliquots of each
172 sample were treated separately as follows: (a) no treatment, (b) heated at 98°C for 15 min, (c)
173 incubated with $3\ \text{mg L}^{-1}$ lysozyme (L7773-50MG, Sigma) for 1 h at room temperature, (d)
174 filtration through a $0.22\ \mu\text{m}$ nitrous cellulose filter (Millipore, USA), (e) heated at 98°C for 15
175 min after treatment (d). In this study, the concentrations of INPs after no treatment and under the
176 latter four different treatments were considered as the concentrations of total INPs, heat-resistant
177 INPs, lysozyme-resistant INPs, nano-INPs and heat-resistant nano-INPs, respectively (Christner
178 et al., 2008; Conen et al., 2012). The concentrations of bio-INPs and bacterial INPs were
179 obtained by subtracting the concentrations of heat-resistant INPs and lysozyme-resistant INPs
180 from the total INP concentrations, respectively. The concentrations of bio-nano-INPs were
181 obtained by subtracting the concentrations of heat-resistant nano-INPs from the nano-INP
182 concentrations.

183 The INA of untreated and different treated samples was determined by droplet freezing method
184 with a modified instrument based on Chen et al. (2018). This method enables the measurement
185 of larger particles and characterizes ice nucleation activity across a broad temperature range,
186 making it popular for studying biological INPs. Substrate choice significantly influences the
187 observed freezing temperature spectrum, and hydrophobic glass is commonly used as a substrate
188 material in this method. Hydrophobic glass limits droplet spreading and contact area, and a large
189 contact angle indicates weaker liquid-substrate interaction, which potentially reduces
190 heterogeneous ice nucleation, enhancing experimental repeatability and accuracy. Nevertheless,
191 hydrophobic glass may still induce ice nucleation due to micro or nanoscale surface defects.
192 Ninety $1\text{-}\mu\text{L}$ droplets were added dropwise to a hydrophobic glass slide located on a cold stage.
193 The cold stage was initially at 0°C and then cooled until all the droplets were frozen at a rate of
194 $1^{\circ}\text{C min}^{-1}$. A charge-coupled device camera was used to record images every 6 s and the images
195 were used to determine the freezing of droplets based on the change in grayscale values during
196 the phase transition from liquid water to ice. As a result, the temperature and the number of
197 frozen droplets at that temperature could be recorded. Some samples were replicated three times
198 to ensure the accuracy of the experiment. The concentrations of INPs, bio-INPs and bacterial
199 INPs at the freezing temperature of -18°C , and nano-INPs and biological nano-INPs at -20°C
200 are calculated as described in previous studies (Vali, 1971) and used to perform related analysis.
201 The laboratory (double-distilled water) and field blanks (double-distilled water poured the clean
202 samplers before sampling) were prepared to assess the effects of contamination using the same

203 ice nucleation assay. The initial freezing temperature (T_0) for the laboratory blanks and filed
204 blanks were observed to be below -20°C and -19°C , respectively. The T_0 results were
205 comparable to the results in previous studies using a similar instrument (Chen et al., 2018) and
206 much lower than those of samples in this study, which indicated the effect of background values
207 using this ice nucleation assay was quite limited.

208 2.3. Taxonomic identification of bacteria and fungi

209 Samples with a volume of 50 mL were filtered by 0.22 μm polycarbonate filters (Φ 25 mm,
210 Whatman, UK). The filters and filtrates were used for biological analysis and OM
211 characterization, respectively. The filters were cut and placed in the lysis tubes. Subsequently,
212 the DNA was extracted by DNeasy PowerSoil Pro Kit (Qiagen, USA) according to the
213 manufacturer's protocol, and quantified by NanoDrop 2000 Spectrophotometers (Thermo Fischer
214 Scientific, Inc., USA). Then a certain amount of DNA was used for polymerase chain reaction
215 (PCR) amplification of the V3-V4 region of the 16S ribosomal RNA (rRNA) gene for bacteria
216 and the internal transcribed spacer (ITS) region of the fungal gene. Primer pair 338F (5'-
217 ACTCCTACGGGAGGCAGCAG-3') and 806R (5'-GGACTACHVGGGTWTCTAAT-3') for
218 bacteria and primer pair ITS1F (5'-CTTGGTCATTAGAGGAAGTAA-3') and ITS2 (5'-
219 GCTGCGTTCTTCATCGATGC-3') for fungi were used for the PCR. The 20- μL PCR mixture
220 contained 10 μL Master Mix, 0.5 μL 10 μM of each primer, 5 μL genome DNA and 4 μL
221 double-distilled H_2O . PCR amplification was performed as follows: initial denaturation at 95°C
222 for 5 min, followed by 35 cycles at 95°C for 30 s, 50°C for bacteria/ 55°C for fungi for 30 s and
223 72°C for 40 s, with a final elongation at 72°C for 7 min. The PCR products were purified using
224 Agencourt AMPure XP magnetic beads. After purification, the ranges of amplified fragments
225 and the DNA concentrations were examined using an Agilent 2100 Bioanalyzer. Then the
226 purified PCR products were sequenced using the Illumina Hiseq 2500 platform (2 \times 250 paired
227 ends) at Bgi Genomics Corporation, Shenzhen, China.

228 The obtained reads from the platform that could match the primers have truncated the primers
229 and connectors, and the treated reads were subjected to quality control according to the
230 requirements of previous studies (Schmieder & Edwards, 2011). The quality-controlled reads
231 were then stitched into tags by overlapping relationships between reads. The tags were clustered
232 into operational taxonomic units (OTUs) at a minimum similarity of 97%. Taxonomic annotation
233 of each OTU was performed using the Silva V138 database for bacteria and the UNITE database
234 for fungi.

235 2.4. FT-ICR MS analysis and molecular formula assignment

236 The filtrates as mentioned in Sect. 2.3 were used to perform FT-ICR MS analysis according to
237 previous studies (Chen et al., 2022). The Oasis hydrophilic-lipophilic balance (HLB) solid-phase
238 extraction (SPE)-cartridges (200 mg, 6 mL) were conditioned with 3 \times 6 mL of methanol and 2 \times 6
239 mL of Milli-Q water on a Supelco Visiprep SPE Vacuum Manifold (USA). Then, samples with a
240 volume of 50 mL were introduced at a flow rate of 1–2 mL min^{-1} onto the cartridges. A
241 subsequent rinse of the cartridges was performed using 6 mL of Milli-Q water, after which they
242 were dried under a nitrogen stream for 1 hour. The analytes loaded onto the cartridges were then
243 eluted with 2 \times 6 mL of methanol. Finally, the extracts were concentrated to 1 mL by rotary
244 evaporators (BUCHI, Switzerland) and stored at -20°C until analysis.

245 The organic fractions of extracts were subjected to analysis using a 7.0T SolariX 2xR FT-ICR
246 MS instrument (Bruker, Germany), which was equipped with an electrospray ionization source
247 (ESI) operating in the negative ion mode. The samples were introduced into the ESI source at an
248 infusion flow rate of 150 $\mu\text{L h}^{-1}$, with a capillary voltage set at 5000 V. Mass spectra were
249 acquired in the range of 150 to 1000 Da, and a total of 256 scans were collected to enhance the
250 signal-to-noise ratio (S/N) for each averaged spectrum.

251 Molecular formulas were assigned to mass peaks with a signal-to-noise ratio (S/N) greater than
252 or equal to 4 using DataAnalysis v5.0 software (Bruker, Germany). The assignment of molecular
253 formulas for the selected peaks was restricted to ^{12}C (1 – 50), ^1H (1 – 100), ^{16}O (1 – 30), ^{14}N (0 –
254 2), and ^{32}S (0 – 2). The assigned molecular formulas were categorized into four groups based on
255 their elemental compositions: CHO, CHON, CHOS, and CHONS. The van Krevelen diagrams
256 were used to classify molecules by H/C ratio and O/C ratio (Kim et al., 2004; Koch et al., 2007).
257 Seven classes of compounds are distinguished according to H/C and O/C ratios: (A) lipids ($1.5 < \text{H/C} \leq 2$,
258 $\text{O/C} \leq 0.3$), (B) aliphatic/proteins ($1.5 < \text{H/C} \leq 2.2$, $0.3 < \text{O/C} \leq 0.67$), (C)
259 lignins/carboxylic-rich alicyclic (CRAMs)-like structures ($0.67 < \text{H/C} \leq 1.5$, $0.1 \leq \text{O/C} < 0.67$),
260 (D) carbohydrates ($1.5 < \text{H/C} \leq 2.5$, $0.67 < \text{O/C} \leq 1.2$), (E) unsaturated hydrocarbons ($0.67 < \text{H/C} \leq 1.5$,
261 $\text{O/C} < 0.1$), (F) aromatic structures ($0.2 \leq \text{H/C} \leq 0.67$, $\text{O/C} < 0.67$), and (G) highly
262 oxygenated compounds (tannins) ($0.6 < \text{H/C} \leq 1.5$, $0.67 \leq \text{O/C} \leq 1.0$). More detailed information
263 about the instrument and molecular formula assignment could be obtained from previous studies
264 (Chen et al., 2022).

265 2.5. Auxiliary chemical component analyses

266 The subsamples were filtered with 0.22- μm pore PTFE filters (MillexGV, Millipore). The
267 filtrates were used to analyze dissolved organic carbon (DOC) and three-dimensional excitation-
268 emission matrix (3D-EEM). DOC was measured by using a TOC analyzer (1030 W + 1088-OI
269 Analytical, US). Fluorescence spectra were measured using 3D-EEM fluorescence spectroscopy
270 (Aqualog, Horiba, Japan) as described in previous studies (Fu et al., 2015).

271 2.6. Data analyses

272 Data analysis in this study was conducted using the R software (R-Core-Team, 2012). The
273 Spearman's correlation analysis was used to assess correlations between peak area of organic
274 molecules and INP concentrations, with a p-value less than 0.05 denoting statistical significance.
275 Selected organic molecules with significant correlations with INPs were subjected to further
276 analysis according to their elemental composition and classification as distinguished in the van
277 Krevelen diagram. Co-occurrence network was constructed based on Spearman's correlation
278 analysis. The correlation coefficients and p-values were calculated between organic molecules
279 and microbial genera. P-values were adjusted to control the False Discovery Rate (FDR) using
280 the Benjamini-Hochberg method. A network was constructed using relationships that were
281 statistically significant ($p < 0.05$) after FDR correction. Network visualization was achieved on
282 the interactive platform Gephi 0.10.1 (<https://gephi.org/>).

283

284 **3. Results**285 **3.1. Characteristics of INPs in precipitation**

286 **Figure 1** shows the cumulative spectra of total INPs per unit volume of precipitation samples
 287 collected at Mt. Lu. The concentrations of total INPs in precipitation samples were 4–4500 mL⁻¹
 288 in the temperature range from -9.6°C to -26.7°C (**Figure S1**), which were within the range of
 289 the spectra observed in rainwater over the Tibetan Plateau (Chen, Wu, Wu, et al., 2021), Quito
 290 and Mexico (Pereira et al., 2021) using the same or similar drop-freezing assays. The total INPs
 291 per unit volume of air assuming a cloud condensed water content (CWC) value of 0.4 g m⁻³
 292 varied from 0.002 to 1.8 L⁻¹ air. The T₀ for fog water and rainwater samples were from -11.2°C
 293 to -5.5°C and from -14.8°C to -9.6°C, respectively, indicating fog water had higher INA than
 294 rainwater in this study. Hartmann et al. (2021) found that fog samples collected near Canada and
 295 European Arctic contained more INPs than seawater samples, freezing onset from -15°C to
 296 -3.5°C. Previous studies observed that cloud or fog water exhibited T₀ warmer than -10°C
 297 (Gong et al., 2020; Joly et al., 2014; Schnell, 1977).

298 Compared to the untreated samples, heat treatment (98°C) decreased T₀ and the temperature at
 299 which 50% of the droplets frozen (T₅₀) by 2.1°C and 1.4°C on average for the rainwater samples,
 300 and by 4.2°C and 3.3°C for the fog water samples, respectively (**Table S1** and **Figure S2**).
 301 Heating treatment and lysozyme treatment are common pretreatment methods to detect
 302 biological and bacterial fractions of INPs, respectively (Christner et al., 2008; Conen et al.,
 303 2012). Results showed that heat-sensitive INPs at temperatures above -11°C constituted up to
 304 98% of the total INPs (**Figure 2a**). On average, 33% of the INPs in fog water were lysozyme-
 305 sensitive at -11°C (**Figure 2b**). Three fog samples were completely free of lysozyme-sensitive
 306 INPs. For rainwater, half of the rainwater samples were less sensitive to lysozyme digestion
 307 throughout the freezing process. It is noted that the samples R4, R7, R8, R9 and R10 contained
 308 11–45 heat-sensitive INPs mL⁻¹ at ≥ -12°C, and almost all these INPs were susceptible to
 309 lysozyme, which suggested that bacterial INPs dominated bio-INPs in these samples.

310 Filtration treatment with 0.22-μm pore filters significantly reduced T₀ and T₅₀ by 1.6°C and
 311 3.0°C for the precipitation samples, respectively (**Table S1** and **Figure S2**). The INPs after
 312 filtration at -18°C in the rainwater and fog water samples accounted for 28% and 35% of the
 313 total INPs, respectively (**Figure 2c**), suggesting the considerable existence of nano-INPs
 314 (defined as INPs < 0.22 μm herein). After heat treatment of the filtrate, T₀ and T₅₀ decreased by
 315 5.1°C (6.3°C for fog water and 3.9°C for rainwater) and 3.3°C (4.9°C for fog water and 1.7°C
 316 for rainwater) on average, respectively, demonstrating the biological origin of nano-INPs, such
 317 as proteinaceous materials (Christner et al., 2008) and lipids (Hill et al., 2016).

318 **3.2. Organic molecular compositions in precipitation**

319 Parallel factor analysis (PARAFAC) of EEM data identified four fluorescent components
 320 (**Figure S3a**). The components C1 (240, 310/395), C2 (242, 326/452), C3 (272/321) and C4 (249,
 321 359/445) (Ex/Em) were categorized as microbial HULIS, transitional marine HULIS, protein-
 322 like organic matter (PLOM) and terrestrial HULIS, respectively (Chen et al., 2022; Yang et al.,
 323 2019; Zhou et al., 2019). In fog water, microbial HULIS (46.1% ± 6.7%) predominated,

324 followed by PLOM ($27.1\% \pm 6.2\%$) and terrestrial HULIS ($23.9\% \pm 3.5\%$). In contrast,
 325 transitional HULIS was most prevalent in rainwater ($40.8\% \pm 25.5\%$) (**Figure S3b**), highlighting
 326 distinct OM sources in rain and fog water.

327 FT-ICR MS analyses were performed to obtain more detailed information on OM at a molecular
 328 level, and the mass spectra of assigned chemical formulas are present in **Figure S4**. In fog and
 329 rainwater samples, 8366 and 5604 formulas were identified, respectively (**Figure S5a**).
 330 Nitrogenous compounds (CHON and CHONS) in fog water predominated, accounting for 52–
 331 69%, while the CHO and CHOS compounds accounted for 15–28% and 15–30%, respectively.
 332 The formula number of CHO and CHON compounds dominated in rainwater (67–86%), and the
 333 compounds that contained sulfur (CHOS and CHONS) occupied $25 \pm 8\%$ (**Figure S5a**).
 334 According to the nominal classification of seven compound classes in van Krevelen diagrams
 335 (Koch et al., 2007), lignins predominated among the assigned formulas in the fog and rainwater
 336 samples with a number fraction of 50% and 52%, respectively, followed by proteinaceous
 337 compounds (24% and 23%) (**Figure S5b**).

338 3.3. Microbial community compositions and potential taxa with INA

339 A total of 402 bacterial OTUs, classified into nine phyla, were detected in fog water and
 340 rainwater samples. The phyla Proteobacteria (37.7% of the total detected sequences), Firmicutes
 341 (20.8%), Actinobacteria (17.1%) and Cyanobacteria (13.7%) were the predominant phyla in all
 342 samples (**Figure S6a**). At the genus level, higher percentages of *Bacillus* (10.8%), GpI
 343 (*Anabaena*, 5.6%), and *Sphingomonas* (5.5%) were identified in rainwater samples, and the
 344 genera GpI (22.0%), and *Sphingomonas* (5.2%) dominated in fog water samples (**Figure S6b**).
 345 Genera *Pseudomonas*, *Pantoea* and *Xanthomonas*, which possibly contain potential ice
 346 nucleation-active bacterial species (Ariya et al., 2009; Kim et al., 1987; Maki et al., 1974), were
 347 detected in all samples, accounting for $3.6 \pm 3.8\%$ of the detected gene sequences (**Figure S6c**).
 348 Similar proportions of known bacteria with INA were also detected in rainwater samples from
 349 Hulunber, China (Du et al., 2017) and Kumamoto, Japan (Hu et al., 2017), and snow samples
 350 from Montreal, Canada (Mortazavi et al., 2008). Genus *Xanthomonas* was present in only six fog
 351 water and four rainwater samples, while the other two genera were detected in all samples.
 352 *Pseudomonas* accounted for 1.1 ± 1.3 and $1.7 \pm 1.4\%$ of the detected gene sequences in fog
 353 water and rainwater samples, respectively (**Figure S6c**). The relative abundance of *Pantoea* was
 354 significantly higher in rainwater samples (4.4%) than in fog water samples (0.9%).

355 In the fungal community, 300 OTUs were detected and classified into seven phyla. As shown in
 356 **Figure S7a**, Ascomycota (73.1%) and Basidiomycota (26.5%) emerged as the dominant phyla in
 357 all samples. In rain samples, the majority of fungi were from the genera *Aspergillus* (18.6%),
 358 *Pseudocercospora* (7.3%), and *Cladosporium* (5.8%) (**Figure S7b**). Among fungal community,
 359 some species of *Fusarium*, *Mortierella*, *Puccinia*, *Sarocladium* (formerly named *Acremonium*)
 360 and *Isaria* with INA could trigger ice freezing above -10°C (Fröhlich-Nowoisky et al., 2015;
 361 Haga et al., 2013; Huffman et al., 2013; Kunert et al., 2019; Pummer et al., 2015; Rodríguez
 362 Zafra et al., 2016). Of these known potential ice nucleation-active fungal genera, *Isaria*,
 363 *Sarocladium*, *Fusarium* and *Mortierella* were detectable in some samples, with a fraction of
 364 0.006%, 0.38%, 0.37% and 0.007% on average, respectively. Known ice nucleation-active
 365 fungal species *Sarocladium implicatum* were detected in more than half of the samples (**Figure**
 366 **S7c**). However, the total abundances of known potential ice nucleation-active fungal genera

367 detected in this study were minor ($0.6 \pm 1.3\%$), indicating the possible existence of unknown ice
368 nucleation-active microbial taxa.

369 3.4. Relationship between INPs and OM compositions in precipitation

370 Spearman's correlation revealed that PLOM (C3) and terrestrial HULIS (C4) were significantly
371 positively related with total, heat-sensitive, and lysozyme-sensitive INPs at -18°C , and nano-
372 INPs and heat-sensitive nano-INPs at -20°C (**Figure S8**). However, transitional marine HULIS
373 (C2) exhibited a strong negative correlation with INPs, suggesting potential interference in INP
374 formation by certain substances. To deepen the understanding of the relationship between OM at
375 a molecular level and INPs, we conducted Spearman's correlation analysis of the detected
376 organic formula peak intensities and different types of INP concentrations. The analysis revealed
377 significant correlations ($p < 0.05$) for 2926, 2108 and 1382 formulas with total, heat-sensitive,
378 and lysozyme-sensitive INPs at -18°C , respectively (**Figure S9**). Among these INP-related
379 formulas, the CHONS group was predominant (34–48%), followed by the CHOS group (30–
380 36%) (**Figure S9a**). OM molecules significantly correlated with different types of INPs are
381 illustrated by van Krevelen diagrams to facilitate the classification of INP-related OM molecules
382 (**Figure 3**), and results showed that the INP-related formulas were mainly assigned to
383 proteinaceous compounds (38–43%) and lignins (28–35%).

384 After filtration treatment, more OM molecules were associated with the concentrations of nano-
385 INPs (formula number: 5380) and heat-sensitive nano-INPs at -20°C (formula number: 5379)
386 (**Figure S9**), which were dominated by CHONS compounds with a fraction of 56%. According
387 to the compound classes, these formulas were mainly assigned to lignins, accounting for 38%,
388 followed by proteinaceous compounds (37%). When correlation coefficients exceeded 0.7, the
389 majority of formulas linked to total, heat-sensitive, and lysozyme-sensitive INPs were
390 proteinaceous compounds (67%, 71% and 72%, respectively) and carbohydrates (17%, 15% and
391 17%). For nano-INPs and heat-sensitive nano-INPs, proteinaceous substances predominated,
392 accounting for 59% and 54% respectively (**Figure S10**).

393 3.5. Relations between microbial taxa and biological INPs-related organic molecules

394 To ascertain if microorganisms contribute organic molecules that potentially affect INPs, we
395 conducted a co-occurrence network analysis based on Spearman's correlations analysis between
396 microbial communities (including bacteria and fungi) and the organic molecules significantly
397 associated with heat-sensitive INPs (r values > 0.7 , formula number: 64) (**Figure S11**). In the
398 constructed network, nodes symbolize OM and microbial genera, and edges denote the
399 correlation strength between OM and microbial genera. A total of 197 and 500 pairs of
400 correlations were established between bacterial and fungal genera and these organic formulas,
401 respectively (**Figure S11**). Twelve bacterial and 13 fungal genera were linked to the organic
402 formulas. Bacterial genera GpI (*Anabaena*), *Friedmanniella* and *Roseomonas* exhibited the
403 highest degree centrality values of 26.5%, 29.1%, and 18.3%, respectively, indicating more
404 organic molecule associations. *Paenibacillus*, *Romboutsia* and *Clostridium_sensu_stricto_1*
405 showed negative associations with some carbohydrate molecules. Other bacterial genera had
406 limited associations with organic molecules. Meanwhile, strong correlations between 13 fungal
407 genera and organic molecules were noted (**Figure S11**). *Sistotrema*, *Hydnomerulius*, *Stereum*,
408 *Diatrype*, *Pseudocercospora* and *Neocucurbitaria* were closely linked together and

409 predominated in all relationships. Moreover, *Trametes* and *Hypholoma* also exhibited positive
 410 correlations with the organic molecules, while *Herpotrichia* showed negative interactions.

411 The network between microbial taxa and biological nano-INPs-related OM formulas (r values $>$
 412 0.7, formula number: 507) revealed that 18 bacterial and 17 fungal genera interacted with the
 413 OM formulas, including all genera found in the previous network (**Figure 4**). The resulting
 414 network was composed of 525 nodes and 1486 edges for bacteria, and 469 nodes and 1259 edges
 415 for fungi, respectively. Predominant bacterial genera in the network were *Microbacterium*, GpI,
 416 *Acidisphaera* and *Roseomonas*, mainly associated with proteinaceous compounds and lipids. In
 417 contrast, several Ascomycota phylum members like *Neocucurbitaria*, *Pseudocercospora*,
 418 *Cadophora*, and *Herpotrichia* showed negative associations with proteinaceous compounds and
 419 lipids, while Basidiomycota phylum members *Hydnomerulius* and *Sistotrema* displayed strong
 420 positive correlations with these compounds.

421 4. Discussion

422 4.1. Dominant contribution of biological matters to precipitation INPs

423 The INPs in all rainwater samples initiated freezing at temperatures above -15°C , and nearly all
 424 fog water samples contained INPs active at $\geq -10^{\circ}\text{C}$ (**Figure 1**). High T_0 and T_{50} values often
 425 signify bio-INPs presence, as many known bio-INPs trigger ice formation at temperatures above
 426 -15°C , while the ice nucleation-active at $< -15^{\circ}\text{C}$ is typically dominated by minerals (Hiranuma
 427 et al., 2013; Hoose & Möhler, 2012; Murray et al., 2013). Here, heat-sensitive INPs represented
 428 bio-INPs (Conen et al., 2022; Sze et al., 2023; Tang et al., 2022), as heating treatment can reduce
 429 the INA of many biological materials (e.g., bacteria, fungi, leaf debris) by inactivating ice
 430 nucleation proteinaceous matter (Failor et al., 2017; Hill et al., 2016; Pouleur et al., 1992;
 431 Schnell & Vali, 1973), though certain pollen is heat-resistant to temperature $> 95^{\circ}\text{C}$ (Duan et al.,
 432 2023; Pummer et al., 2012).

433 Results showed that the total INPs in all the rainwater samples initiated freezing at temperatures
 434 warmer than -15°C , and almost all the fog water samples contained INPs active at $\geq -10^{\circ}\text{C}$
 435 (**Figure 1**). The mechanisms of fog and rain formation differ, as well as the conditions and
 436 physical processes under which they exist in the atmosphere. The formation of fog primarily
 437 occurs through the direct condensation of water vapor onto ice nuclei or condensation nuclei,
 438 resulting in the formation of tiny water droplets (Gill et al., 1983; Gultepe et al., 2007), which
 439 may cause INPs to be more concentrated within the fog. In contrast, rain formation involves
 440 different atmospheric processes and conditions that might not always enhance the INP
 441 concentrations as much as in fog. Additionally, fog droplets are smaller, providing a larger
 442 surface area capable of more effectively absorbing and enriching particulate matter in the air.
 443 Compared to rainwater, the limited dilution of fog water and the more efficient transfer of
 444 surface emissions to fog water may contribute to higher concentrations of particulate matter in
 445 fog water (Klemm & Wrzesinsky, 2007). The total INPs at temperatures $> -15^{\circ}\text{C}$ in fog water
 446 contained a large fraction (51.2%) of heat-sensitive INPs, with proportions increasing with
 447 temperature in most samples (**Figure 2**), suggesting a dominance of biological materials in INPs.
 448 The bio-INPs to non-bio-INPs ratio was significantly higher in fog water (10.5%) than in
 449 rainwater (1.6%), indicating a greater contribution of bio-INPs in fog water. Notably, fractions of
 450 heat-sensitive INPs at -24°C to -20°C enhanced to 78% and 89% on average in R6 and R7,

451 respectively, indicating other INP types besides bio-INPs (Atkinson et al., 2013; Hiranuma et al.,
 452 2013), also found by Chen, Wu, Chen, et al. (2021). Some inorganic substances (e.g., quartz)
 453 whose INA is reduced by thermal interference, and other non-thermally labile substances may
 454 contribute to this phenomenon (Harrison et al., 2019; Wilson et al., 2015).

455 Bacteria are potentially an important type of bio-INPs considering their number in air (Failor et
 456 al., 2017; Lohmann & Feichter, 2005; Morris et al., 2004). INPs sensitive to lysozyme digestion
 457 (i.e., to dissolve the cell wall structure) have been widely regarded as bacterial INPs (Christner et
 458 al., 2008; Failor et al., 2017; Morris et al., 2013). However, lysozyme mainly acts on
 459 peptidoglycans (Masschalck & Michiels, 2003; Repaske, 1956), less effective on Gram-negative
 460 bacteria with low peptidoglycan cell wall content (Joly et al., 2014), and the INA of urediospores
 461 of rust fungi even increased after lysozyme treatment (Morris et al., 2013). Therefore, the use of
 462 lysozyme-sensitive INPs to indicate bacterial INPs may be subject to some error. Lysozyme-
 463 sensitive INPs accounted for 44.8% of bio-INPs, suggesting other types of biological particles,
 464 e.g., Gram-negative bacteria, fungi, or plant debris also contributed to bio-INPs.

465 The filtration tests showed a certain number of INPs passed through 0.2 µm filters (**Figure S2**),
 466 suggesting the existence of nano-INPs. Nanoscale INPs could promote the efficient formation of
 467 ice embryos due to the large specific surface area (Pruppacher & Klett, 2010) or attachment to
 468 the particle surfaces thereby enhancing the INA of particles (O'Sullivan et al., 2016; Pummer et
 469 al., 2015). Such findings were observed in terrestrial and marine ecosystems (Augustin et al.,
 470 2013; Du et al., 2017) and nano-INPs were highly likely contributed by smaller biological
 471 particles (Pummer et al., 2015). Known ice nucleation-active bacteria or fungi could exist as
 472 submicron fragments, and microbial fragments and ice nucleation-active proteinaceous matter
 473 were passed through the filter and retained in the filtrate (Šantl-Temkiv et al., 2015). Pollen
 474 readily released large amounts of nanoscale INPs when in contact with water (Augustin et al.,
 475 2013; Duan et al., 2023). This was consistent with the finding in this study that the proportions of
 476 nano-INPs in total INPs were almost identical to that of heat-sensitive nano-INPs (**Figure 2**).
 477 These nano-INPs were mainly biological, such as proteinaceous materials, lipid bodies,
 478 carbohydrates and cell structures (such as ribosomes), which could be degraded or inactivated by
 479 heating (Hill et al., 2016; Rederstorff et al., 2011; Urano & Douple, 2023).

480 4.2. OM molecules with high INA in precipitation

481 **Figure 3** depicts sulfur-containing compounds (CHOS, CHONS) dominating the OM formulas
 482 associated with INPs, with a positive correlation between different INP concentrations and the
 483 formula numbers of CHOS and CHONS (**Table S2**), highlighting the possible importance of
 484 sulfur-containing compounds in INPs. Previous studies have demonstrated that under deposition
 485 nucleation mode sulfate-organic particles could promote heterogeneous ice nucleation (Froyd et
 486 al., 2010; Knopf et al., 2018), and organosulfur compounds dominated ice residuals (Cziczo et
 487 al., 2013; DeMott et al., 2003). Several potential explanations were proposed in previous studies
 488 regarding OM molecules as INPs. Fukuta (1966) proposed that organics with INA are mostly
 489 crystalline solids with low solubility and high melting points, having polar or hydrogen bonding
 490 groups as active sites. Baustian et al. (2012) uncovered a complex link between ice formation
 491 and organic materials, with INA potentially influenced by chemical composition and spatial
 492 chemical arrangement in mixed particles. Here, CHOS and CHONS were predominantly
 493 assigned to proteinaceous compounds (**Figure 3**), possibly vital for ice formation (O'Sullivan et

al., 2016; Šantl-Temkiv et al., 2019), partially accounting for the possible INA of the sulfur-containing compounds in this study. Several possible structures of these proteinaceous compounds are presented in **Figure 3f**. Due to technical limitations, only the molecular formulas and the classification of the detected compounds could be identified, and the molecular structures are not yet known due to isomers. The proteinaceous compounds shown in **Figure 3f** represent merely one among several potential structures. The precise molecular structures, as yet undetermined, necessitate additional comprehensive investigation.

Proteinaceous materials are the most studied organics to nucleate ice because the presence of proteinaceous matter in the outer layer of the cell membrane enables the INA of microorganisms (Gurian-Sherman & Lindow, 1993; Lindow et al., 1989). The structure of ice nucleation-active proteinaceous matter contains a hydrophobic N-terminal domain, a hydrophilic C-terminal domain and a large central repeating domain that can act as an ice nucleation site (Huang et al., 2021; Wolber & Warren, 1989). In this study, more proteinaceous compounds had positive relationships with different types of INPs (**Figure 3**), further clarifying the important contribution of proteinaceous matter to INPs. Interestingly, some proteinaceous compounds exhibiting negative correlations with INPs were identified (**Figure 3** and **Table S3**), implying the presence of antifreeze proteinaceous materials (Davies, 2014; Dreischmeier et al., 2017; Govindarajan & Lindow, 1988). These proteinaceous materials are ice-structured materials that inhibit ice growth by maintaining the temperature in the range between the melting point and the freezing point(Baskaran et al., 2021).

In addition, lignins and carbohydrates were also significantly positively correlated with INPs (**Figure 3** and **S9**). Lignin-like substances are complex organic polymers derived from the cell wall structure of vascular plants (Miller et al., 2021). A large number of lignins existing in plant xylem had INA (Conen et al., 2016; Gute & Abbatt, 2020), and lignins have been identified as water-soluble macromolecules with INA (Pummer et al., 2015; Steinke et al., 2020). The INA of lignins may also be confirmed by the result that more lignin molecules were highly related to nano-INPs compared to total INPs (**Figure S9**). In addition, due to the stable INA under different environmental stresses (Bogler & Borduas-Dedekind, 2020), lignins have been recommended for use as an ice-nucleating standard (Miller et al., 2021).

Carbohydrates largely contributed to the INA of pollen or plants (Dreischmeier et al., 2017; Krog et al., 1979). Carbohydrate mixtures with other substances (e.g., mucilage or proteinaceous substances) serve as intrinsic ice nucleators in some plants (Brush et al., 1994; Embuscado et al., 1996). Under high humidity, pollen emits vast amounts of carbohydrate-rich submicron particles that retain the INA of the parent body (Duan et al., 2023; Hill et al., 2017; Pummer et al., 2012; Steiner et al., 2015). Notably, some polysaccharides can inhibit ice nucleation (Yamashita et al., 2002), with their effect on ice formation depending on molecular size and structure (Dreischmeier et al., 2017; Duan et al., 2023; Walters et al., 2009). Dreischmeier et al. (2017) found that the smaller polysaccharides (< 100 kDa) exhibited stronger ice-binding abilities. The molecular structure of known INPs shows that INA is primarily attributed to the presence of hydroxyl groups in the molecules (Graether & Jia, 2001), which could potentially elucidate the INA of carbohydrates with multiple hydrogen bonds, especially those with hydroxyl groups (Wolf et al., 2019).

536 4.3. Potential microbial sources of INPs-related OM molecules

537 Fundamentally, microorganisms own INA because of the presence of intrinsic substances that
538 promote ice nucleation (Delort et al., 2010). These substances are likely to be biological
539 macromolecules such as proteinaceous substances, lipids and carbohydrates (Koop & Zobrist,
540 2009). Ice-nucleating sites in bacteria and fungi were mainly contributed by proteinaceous
541 substances, while pollen is more likely to be contributed by carbohydrates and/or proteinaceous
542 substances (Dreischmeier et al., 2017; O'Sullivan et al., 2016). This is in agreement with the
543 findings in this study that proteinaceous or carbohydrate-like substances significantly correlated
544 with INPs (**Figure 3**). In the constructed co-occurrence networks, bacterial taxa with positive
545 correlations with OM mostly belong to the order Actinomycetales (**Figure S12**) which are
546 filamentous Gram-positive bacteria without a nucleus (Yanti et al., 2012). The known ice-
547 nucleating bacteria were mostly Gram-negative (Maki et al., 1974), but a few Gram-positive
548 species with INA were also detected (Failor et al., 2017). Recent studies isolated several Gram-
549 positive bacteria with INA, i.e., *Paenibacillus* sp., *Bacillus* sp. (Beall et al., 2021), and some
550 species of Actinomycetes (such as *Microbacterium esteraromaticum*, *Rhodococcus*
551 *corynebacteroides* and *Brevibacterium* sp.) (Cid et al., 2016), some of which were also occurred in
552 the constructed co-occurrence networks.

553 Compared with bacteria (edges: 772), fungi (edges: 492) were positively associated with fewer
554 nano-INPs-related OM molecules (**Figure 4**), suggesting a lesser contribution of fungi to nano-
555 INPs-related OM molecules. In previous studies, fungi with high INA were some lichen
556 mycobionts (Kieft & Ruscetti, 1990) and some species of *Fusarium*, *Penicillium* and
557 *Cladosporium* (Pouleur et al., 1992). Recently, ice nucleation abilities were detected in other
558 fungal taxa, e.g., some species of *Isaria* and *Acremonium* (Huffman et al., 2013), as well as in
559 some rust fungi (Morris et al., 2013). Among them, rust fungi had a high capacity for ice
560 nucleation, which initiated ice nucleation at $> -4^{\circ}\text{C}$ (Morris et al., 2013), but most fungi were
561 less capable of ice formation than bacteria (Maki et al., 1974; Obata et al., 1989). Meanwhile, the
562 genera positively associated with nano-INPs-related OM molecules were *Hydnomerulius* and
563 *Sistotrema*, which belong to Agaricomycetes, implying a potential ice nucleation ability of
564 Agaricomycetes. A similar finding was provided by Tang et al. (2022) that elevated INP
565 concentrations in precipitation were accompanied by increases in the relative abundances of
566 Agaricomycetes.

567 The presence of proteins, carbohydrates or lipid fractions with ice nucleation or antifreeze
568 capabilities within the microorganism enable diverse biological entities to influence ice
569 formation differently (Dreischmeier et al., 2017; O'Sullivan et al., 2015). Nonetheless, OM
570 molecules were significantly smaller than intact cells (Govindarajan & Lindow, 1988; Pummer
571 et al., 2012), and the possibility of organic compounds with varying ice nucleation abilities
572 coexisting within the same microorganism cannot be discounted (Dreischmeier et al., 2017;
573 Failor et al., 2017; Rice et al., 2015). Certain proteinaceous materials or carbohydrates possess
574 both ice-binding and ice-nucleating abilities (Xu et al., 1998). For example, the large ice-
575 nucleating polysaccharides in birch pollen may be composed of smaller clusters of ice-binding
576 polysaccharides (Dreischmeier et al., 2017). Moreover, a fraction of the ice-nucleating protein
577 within *Pseudomonas syringae* exhibited ice-binding capability (Kobashigawa et al., 2005). This
578 mechanism could help explain the occurrence of a comparable microbial taxa in both co-

579 occurrence networks of bio-INPs-related and biological nano-INPs-related OM formulas with
580 microbial communities (**Figure 4 and S11**).

581 **5. Conclusions**

582 Building upon previous research works, this study for the first time emphasizes the linkage
583 between INPs, organic molecules, and microorganisms in environmental precipitation samples. It
584 highlights the significant contribution of biological materials to INPs, advancing our
585 understanding of these interactions in real environments. OM molecules associated with INPs
586 predominantly comprised sulfur-containing compounds, as revealed by the van Krevelen
587 diagram, which allocated these molecules chiefly to biologically relevant categories such as
588 proteins, lignin, and carbohydrates. Co-occurrence networks further corroborated that specific
589 microorganisms may contribute to INP-related OM molecules, with the contribution from
590 bacteria being more substantial than that from fungi. It warrants acknowledgment, however, that
591 the genera appearing in co-occurrence networks have not been tested for INA, and the known
592 ice-nucleating taxa did not feature in the networks. The ice nucleation capacity of a considerable
593 array of microbial taxa remains uncharted owing to microbial diversity, and the studies
594 scrutinizing OM molecules possessing INA in organisms are still limited. Future investigations
595 are imperative for the identification of the ice nucleation ability across a broader spectrum of
596 microbial taxa.

597 **Acknowledgments**

598 This work was supported by the National Key R&D Plan (Grant No. 2022YFF0803000).

599 **Conflict of Interest**

600 The authors declare no conflicts of interest relevant to this study.

601 **Data Availability Statement**

602 The sequences have been deposited in the NCBI with BioProject accession number
603 PRJNA994904 (<https://www.ncbi.nlm.nih.gov/sra/PRJNA994904>). Detailed methods in the
604 study are available in the supporting information. The data are hosted at
605 <https://zenodo.org/records/10300240> (Niu et al., 2024).

606

607 **References**

- 608 Ahern, H. E., Walsh, K. A., Hill, T. C. J., & Moffett, B. F. (2007). Fluorescent pseudomonads isolated from
609 Hebridean cloud and rain water produce biosurfactants but do not cause ice nucleation. *Biogeosciences*,
610 4(1), 115-124. <https://doi.org/10.5194/bg-4-115-2007>
- 611 Ariya, P. A., Sun, J., Eltouny, N. A., Hudson, E. D., Hayes, C. T., & Kos, G. (2009). Physical and chemical
612 characterization of bioaerosols – Implications for nucleation processes. *International Reviews in Physical
613 Chemistry*, 28(1), 1-32. 10.1080/01442350802597438
- 614 Arny, D., Lindow, S., & Upper, C. (1976). Frost sensitivity of Zea mays increased by application of *Pseudomonas
615 syringae*. *Nature*, 262, 282–284. <https://doi.org/10.1038/262282a0>

- 616 Atkinson, J. D., Murray, B. J., Woodhouse, M. T., Whale, T. F., Baustian, K. J., Carslaw, K. S., et al. (2013). The
 617 importance of feldspar for ice nucleation by mineral dust in mixed-phase clouds. *Nature*, 498(7454), 355-
 618 358. <https://doi.org/10.1038/nature12278>
- 619 Augustin, S., Wex, H., Niedermeier, D., Pummer, B., Grothe, H., Hartmann, S., et al. (2013). Immersion freezing of
 620 birch pollen washing water. *Atmospheric Chemistry & Physics*, 13(21), 10989-11003.
 621 <https://doi.org/10.5194/acp-13-10989-2013>
- 622 Baskaran, A., Kaari, M., Venugopal, G., Manikkam, R., Joseph, J., & Bhaskar, P. V. (2021). Anti freeze proteins
 623 (Afp): Properties, sources and applications – A review. *International Journal of Biological
 624 Macromolecules*, 189, 292-305. <https://doi.org/10.1016/j.ijbiomac.2021.08.105>
- 625 Baustian, K. J., Cziczo, D. J., Wise, M. E., Pratt, K. A., Kulkarni, G., Hallar, A. G., & Tolbert, M. A. (2012).
 626 Importance of aerosol composition, mixing state, and morphology for heterogeneous ice nucleation: A
 627 combined field and laboratory approach. *Journal of Geophysical Research: Atmospheres*, 117(D6),
 628 D06217. <https://doi.org/10.1029/2011JD016784>
- 629 Beall, C. M., Lucero, D., Hill, T. C., DeMott, P. J., Stokes, M. D., & Prather, K. A. (2020). Best practices for
 630 precipitation sample storage for offline studies of ice nucleation in marine and coastal environments.
 631 *Atmospheric Measurement Techniques*, 13(12), 6473-6486. <https://doi.org/10.5194/amt-13-6473-2020>
- 632 Beall, C. M., Michaud, J. M., Fish, M. A., Dinasquet, J., Cornwell, G. C., Stokes, M. D., et al. (2021). Cultivable
 633 halotolerant ice-nucleating bacteria and fungi in coastal precipitation. *Atmospheric Chemistry & Physics*,
 634 21(11), 9031-9045. <https://doi.org/10.5194/acp-21-9031-2021>
- 635 Bogler, S., & Borduas-Dedekind, N. (2020). Lignin's ability to nucleate ice via immersion freezing and its stability
 636 towards physicochemical treatments and atmospheric processing. *Atmospheric Chemistry & Physics*,
 637 20(23), 14509-14522. <https://doi.org/10.5194/acp-20-14509-2020>
- 638 Borduas-Dedekind, N., Ossola, R., David, R. O., Boynton, L. S., Weichlinger, V., Kanji, Z. A., & McNeill, K.
 639 (2019). Photomineralization mechanism changes the ability of dissolved organic matter to activate cloud
 640 droplets and to nucleate ice crystals. *Atmospheric Chemistry & Physics*, 19(19), 12397-12412.
 641 <https://doi.org/10.5194/acp-19-12397-2019>
- 642 Brush, R. A., Griffith, M., & Mlynarz, A. (1994). Characterization and quantification of intrinsic ice nucleators in
 643 winter rye (*Secale cereale*) leaves. *Plant Physiology*, 104(2), 725-735. <https://doi.org/10.1104/pp.104.2.725>
- 644 Cascajo-Castresana, M., David, R. O., Iriarte-Alonso, M. A., Bittner, A. M., & Marcolli, C. (2020). Protein
 645 aggregates nucleate ice: The example of apoferritin. *Atmospheric Chemistry & Physics*, 20(6), 3291-3315.
 646 <https://doi.org/10.5194/acp-20-3291-2020>
- 647 Che, Y., Dang, J., Fang, W., Shen, X., Sun, J., Chen, Y., & Qian, Y. (2019). Measurements of natural ice nucleating
 648 particles in Beijing in the spring of 2017. *Atmospheric Environment*, 200, 170-177.
 649 <https://doi.org/10.1016/j.atmosenv.2018.12.020>
- 650 Chen, J., Pei, X. Y., Wang, H., Chen, J. C., Zhu, Y. S., Tang, M. G., & Wu, Z. J. (2018). Development,
 651 characterization, and validation of a cold stage-based ice nucleation array (PKU-INA). *Atmosphere*, 9(9),
 652 357. <https://doi.org/10.3390/atmos9090357>
- 653 Chen, J., Wu, Z., Chen, J., Reicher, N., Fang, X., Rudich, Y., & Hu, M. (2021). Size-resolved atmospheric ice-
 654 nucleating particles during East Asian dust events. *Atmospheric Chemistry & Physics*, 21(5), 3491-3506.
 655 <https://doi.org/10.5194/acp-21-3491-2021>
- 656 Chen, J., Wu, Z. J., Wu, G. M., Gong, X. D., Wang, F., Chen, J. C., et al. (2021). Ice-nucleating particle
 657 concentrations and sources in rainwater over the third pole, Tibetan Plateau. *Journal of Geophysical
 658 Research: Atmospheres*, 126(9), e2020JD033864. <https://doi.org/10.1029/2020JD033864>
- 659 Chen, J., Wu, Z. J., Zhao, X., Wang, Y. J., Chen, J. C., Qiu, Y. T., et al. (2021). Atmospheric humic-like substances
 660 (HULIS) act as ice active entities. *Geophysical Research Letters*, 48(14), e2021GL092443.
 661 <https://doi.org/10.1029/2021GL092443>
- 662 Chen, S., Xie, Q., Su, S., Wu, L., Zhong, S., Zhang, Z., et al. (2022). Source and formation process impact the
 663 chemodiversity of rainwater dissolved organic matter along the Yangtze River Basin in summer. *Water
 664 Research*, 211, 118024. <https://doi.org/10.1016/j.watres.2021.118024>
- 665 Christner, B. C. (2010). Bioprospecting for microbial products that affect ice crystal formation and growth. *Applied
 666 Microbiology and Biotechnology*, 85(3), 481-489. <https://doi.org/10.1007/s00253-009-2291-2>
- 667 Christner, B. C., Cai, R., Morris, C. E., McCarter, K. S., Foreman, C. M., Skidmore, M. L., et al. (2008).
 668 Geographic, seasonal, and precipitation chemistry influence on the abundance and activity of biological ice
 669 nucleators in rain and snow. *The Proceedings of the National Academy of Sciences*, 105(48), 18854-18859.
 670 <https://doi.org/10.1073/pnas.0809816105>

- 671 Cid, F. P., Rilling, J. I., Graether, S. P., Bravo, L. A., Mora, M. d. L. L., & Jorquera, M. A. (2016). Properties and
 672 biotechnological applications of ice-binding proteins in bacteria. *FEMS Microbiology Letters*, 363(11),
 673 fnw099. <https://doi.org/10.1093/femsle/fnw099>
- 674 Conen, F., Einbock, A., Mignani, C., & Hüglin, C. (2022). Measurement report: Ice-nucleating particles active \geq
 675 -15°C in free tropospheric air over western Europe. *Atmospheric Chemistry & Physics*, 22(5), 3433-3444.
 676 <https://doi.org/10.5194/acp-22-3433-2022>
- 677 Conen, F., Henne, S., Morris, C. E., & Alewell, C. (2012). Atmospheric ice nucleators active $\geq -12^{\circ}\text{C}$ can be
 678 quantified on PM₁₀ filters. *Atmospheric Measurement Techniques*, 5(2), 321-327.
 679 <https://doi.org/10.5194/amt-5-321-2012>
- 680 Conen, F., Morris, C. E., Leifeld, J., Yakutin, M. V., & Alewell, C. (2011). Biological residues define the ice
 681 nucleation properties of soil dust. *Atmospheric Chemistry & Physics*, 11(18), 9643-9648.
 682 <https://doi.org/10.5194/acp-11-9643-2011>
- 683 Conen, F., Rodríguez, S., Hülin, C., Henne, S., Herrmann, E., Bukowiecki, N., & Alewell, C. (2015). Atmospheric
 684 ice nuclei at the high-altitude observatory Jungfraujoch, Switzerland. *Tellus B: Chemical and Physical
 685 Meteorology*, 67(1), 25014. <https://doi.org/10.3402/tellusb.v67.25014>
- 686 Conen, F., Stopelli, E., & Zimmermann, L. (2016). Clues that decaying leaves enrich Arctic air with ice nucleating
 687 particles. *Atmospheric Environment*, 129, 91-94. <https://doi.org/10.1016/j.atmosenv.2016.01.027>
- 688 Creamean, J. M., Lee, C., Hill, T. C., Ault, A. P., DeMott, P. J., White, A. B., et al. (2014). Chemical properties of
 689 insoluble precipitation residue particles. *Journal of Aerosol Science*, 76, 13-27.
 690 <https://doi.org/10.1016/j.jaerosci.2014.05.005>
- 691 Creamean, J. M., Mignani, C., Bukowiecki, N., & Conen, F. (2019). Using freezing spectra characteristics to
 692 identify ice-nucleating particle populations during the winter in the Alps. *Atmospheric Chemistry &
 693 Physics*, 19(12), 8123-8140. <https://doi.org/10.5194/acp-19-8123-2019>
- 694 Creamean, J. M., Suski, K. J., Rosenfeld, D., Cazorla, A., DeMott, P. J., Sullivan, R. C., et al. (2013). Dust and
 695 Biological Aerosols from the Sahara and Asia Influence Precipitation in the Western U.S. *Science*,
 696 339(6127), 1572-1578. <https://doi.org/10.1126/science.1227279>
- 697 Creamean, J. M., White, A. B., Minnis, P., Palikonda, R., Spangenberg, D. A., & Prather, K. A. (2016). The
 698 relationships between insoluble precipitation residues, clouds, and precipitation over California's southern
 699 Sierra Nevada during winter storms. *Atmospheric Environment*, 140, 298-310.
 700 <https://doi.org/10.1016/j.atmosenv.2016.06.016>
- 701 Cziczo, D. J., Froyd, K. D., Hoose, C., Jensen, E. J., Diao, M., Zondlo, M. A., et al. (2013). Clarifying the dominant
 702 sources and mechanisms of cirrus cloud formation. *Science*, 340(6138), 1320-1324.
 703 <https://doi.org/10.1126/science.1234145>
- 704 Davies, P. L. (2014). Ice-binding proteins: a remarkable diversity of structures for stopping and starting ice growth.
 705 *Trends in Biochemical Sciences*, 39(11), 548-555. <https://doi.org/10.1016/j.tibs.2014.09.005>
- 706 Delort, A.-M., Vaïtilingom, M., Amato, P., Sancelme, M., Parazols, M., Mailhot, G., et al. (2010). A short overview
 707 of the microbial population in clouds: Potential roles in atmospheric chemistry and nucleation processes.
 708 *Atmospheric Research*, 98(2), 249-260. <https://doi.org/10.1016/j.atmosres.2010.07.004>
- 709 DeMott, P. J., Cziczo, D. J., Prenni, A. J., Murphy, D. M., Kreidenweis, S. M., Thomson, D. S., et al. (2003).
 710 Measurements of the concentration and composition of nuclei for cirrus formation. *The Proceedings of the
 711 National Academy of Sciences*, 100(25), 14655-14660. <https://doi.org/10.1073/pnas.2532677100>
- 712 DeMott, P. J., Prenni, A. J., Liu, X., Kreidenweis, S. M., Petters, M. D., Twohy, C. H., et al. (2010). Predicting
 713 global atmospheric ice nuclei distributions and their impacts on climate. *The Proceedings of the National
 714 Academy of Sciences*, 107(25), 11217-11222. <https://doi.org/10.1073/pnas.0910818107>
- 715 Dreischmeier, K., Budke, C., Wiedemeier, L., Kottke, T., & Koop, T. (2017). Boreal pollen contain ice-nucleating
 716 as well as ice-binding 'antifreeze' polysaccharides. *Scientific Reports*, 7(1), 41890.
 717 <https://doi.org/10.1038/srep41890>
- 718 Du, R., Du, P., Lu, Z., Ren, W., Liang, Z., Qin, S., et al. (2017). Evidence for a missing source of efficient ice
 719 nuclei. *Scientific Reports*, 7(1), 39673. <https://doi.org/10.1038/srep39673>
- 720 Duan, P., Hu, W., Wu, Z., Bi, K., Zhu, J., & Fu, P. (2023). Ice nucleation activity of airborne pollen: A short review
 721 of results from laboratory experiments. *Atmospheric Research*, 285, 106659.
 722 <https://doi.org/10.1016/j.atmosres.2023.106659>
- 723 Embuscado, M. E., BeMiller, J. N., & Knox, E. B. (1996). A survey and partial characterization of iceucleating
 724 fluids secreted by giant-rosette (*Lobelia* and *Dendrosenecio*) plants of the mountains of eastern Africa.
 725 *Carbohydrate Polymers*, 31(1), 1-9. [https://doi.org/10.1016/S0144-8617\(96\)00120-8](https://doi.org/10.1016/S0144-8617(96)00120-8)

- 726 Failor, K. C., Liu, H., Llontop, M. E. M., LeBlanc, S., Eckshtain-Levi, N., Sharma, P., et al. (2022). Ice nucleation
 727 in a Gram-positive bacterium isolated from precipitation depends on a polyketide synthase and non-
 728 ribosomal peptide synthetase. *The ISME Journal*, 16(3), 890-897. <https://doi.org/10.1038/s41396-021-01140-4>
- 729
- 730 Failor, K. C., Schmale, D. G., Vinatzer, B. A., & Monteil, C. L. (2017). Ice nucleation active bacteria in
 731 precipitation are genetically diverse and nucleate ice by employing different mechanisms. *The ISME
 732 Journal*, 11(12), 2740-2753. <https://doi.org/10.1038/ismej.2017.124>
- 733 Fröhlich-Nowoisky, J., Hill, T. C. J., Pummer, B. G., Yordanova, P., Franc, G. D., & Pöschl, U. (2015). Ice
 734 nucleation activity in the widespread soil fungus *Mortierella alpina*. *Biogeosciences*, 12(4), 1057-1071.
 735 <https://doi.org/10.5194/bg-12-1057-2015>
- 736 Froyd, K. D., Murphy, D. M., Lawson, P., Baumgardner, D., & Herman, R. L. (2010). Aerosols that form subvisible
 737 cirrus at the tropical tropopause. *Atmospheric Chemistry & Physics*, 10(1), 209-218.
 738 <https://doi.org/10.5194/acp-10-209-2010>
- 739 Fu, P., Kawamura, K., Chen, J., Qin, M., Ren, L., Sun, Y., et al. (2015). Fluorescent water-soluble organic aerosols
 740 in the High Arctic atmosphere. *Scientific Reports*, 5(1), 9845. <https://doi.org/10.1038/srep09845>
- 741 Fukuta, N. (1966). Experimental studies of organic ice nuclei. *Journal of Atmospheric Sciences*, 23(2), 191-196.
 742 [https://doi.org/10.1175/1520-0469\(1966\)023<0191:ESOON>2.0.CO;2](https://doi.org/10.1175/1520-0469(1966)023<0191:ESOON>2.0.CO;2)
- 743 Gill, P. S., Graedel, T. E., & Weschler, C. J. (1983). Organic films on atmospheric aerosol particles, fog droplets,
 744 cloud droplets, raindrops, and snowflakes. *Reviews of Geophysics*, 21(4), 903-920.
 745 <https://doi.org/10.1029/RG021i004p00903>
- 746 Gong, X., Wex, H., Voigtlander, J., Fomba, K. W., Weinhold, K., van Pinxteren, M., et al. (2020). Characterization
 747 of aerosol particles at Cabo Verde close to sea level and at the cloud level – Part 1: Particle number size
 748 distribution, cloud condensation nuclei and their origins. *Atmospheric Chemistry & Physics*, 20(3), 1431-
 749 1449. <https://doi.org/10.5194/acp-20-1431-2020>
- 750 Govindarajan, A. G., & Lindow, S. E. (1988). Size of bacterial ice-nucleation sites measured in situ by radiation
 751 inactivation analysis. *Proceedings of the National Academy of Sciences*, 85(5), 1334-1338.
 752 <https://doi.org/10.1073/pnas.85.5.1334>
- 753 Graether, S. P., & Jia, Z. (2001). Modeling *Pseudomonas syringae* ice-nucleation protein as a β -helical protein.
 754 *Biophysical Journal*, 80(3), 1169-1173. [https://doi.org/10.1016/S0006-3495\(01\)76093-6](https://doi.org/10.1016/S0006-3495(01)76093-6)
- 755 Gultepe, I., Tardif, R., Michaelides, S. C., Cermak, J., Bott, A., Bendix, J., et al. (2007). Fog Research: A Review of
 756 Past Achievements and Future Perspectives. *Pure and Applied Geophysics*, 164(6), 1121-1159.
 757 <https://doi.org/10.1007/s00024-007-0211-x>
- 758 Guo, L., Guo, X., Lou, X., Lu, G., Lü, K., Sun, H., et al. (2019). An observational study of diurnal and seasonal
 759 variations, and macroscopic and microphysical properties of clouds and precipitation over Mount Lu,
 760 Jiangxi, China. *Acta Meteorologica Sinica*, 77(5), 923-937. <https://doi.org/10.11676/qxxb2019.056>
- 761 Gurian-Sherman, D., & Lindow, S. E. (1993). Bacterial ice nucleation: Significance and molecular basis. *FASEB
 762 Journal*, 7(14), 1338-1343. <https://doi.org/10.1096/fasebj.7.14.8224607>
- 763 Gute, E., & Abbatt, J. P. D. (2020). Ice nucleating behavior of different tree pollen in the immersion mode.
 764 *Atmospheric Environment*, 231, 117488. <https://doi.org/10.1016/j.atmosenv.2020.117488>
- 765 Haga, D. I., Iannone, R., Wheeler, M. J., Mason, R., Polishchuk, E. A., Fetch Jr., T., et al. (2013). Ice nucleation
 766 properties of rust and bunt fungal spores and their transport to high altitudes, where they can cause
 767 heterogeneous freezing. *Journal of Geophysical Research: Atmospheres*, 118(13), 7260-7272.
 768 <https://doi.org/10.1002/jgrd.50556>
- 769 Harrison, A. D., Lever, K., Sanchez-Marroquin, A., Holden, M. A., Whale, T. F., Tarn, M. D., et al. (2019). The ice-
 770 nucleating ability of quartz immersed in water and its atmospheric importance compared to K-feldspar.
 771 *Atmospheric Chemistry & Physics*, 19(17), 11343-11361. <https://doi.org/10.5194/acp-19-11343-2019>
- 772 Hartmann, M., Gong, X., Kecorius, S., van Pinxteren, M., Vogl, T., Welti, A., et al. (2021). Terrestrial or marine –
 773 indications towards the origin of ice-nucleating particles during melt season in the European Arctic up to
 774 83.7°N. *Atmospheric Chemistry & Physics*, 21(15), 11613-11636. <https://doi.org/10.5194/acp-21-11613-2021>
- 775
- 776 Hew, C. L., & Yang, D. S. (1992). Protein interaction with ice. *European Journal of Biochemistry*, 203(1-2), 33-42.
 777 <https://doi.org/10.1111/j.1432-1033.1992.tb19824.x>
- 778 Hill, T. C. J., DeMott, P. J., Conen, F., & Möhler, O. (2017). Impacts of bioaerosols on atmospheric ice nucleation
 779 processes. In *Microbiology of Aerosols* (pp. 195-219).

- 780 Hill, T. C. J., DeMott, P. J., Tobo, Y., Fröhlich-Nowoisky, J., Moffett, B. F., Franc, G. D., & Kreidenweis, S. M.
 781 (2016). Sources of organic ice nucleating particles in soils. *Atmospheric Chemistry & Physics*, 16(11),
 782 7195-7211. <https://doi.org/10.5194/acp-16-7195-2016>
- 783 Hiranuma, N., Adachi, K., Bell, D. M., Belosi, F., Beydoun, H., Bhaduri, B., et al. (2019). A comprehensive
 784 characterization of ice nucleation by three different types of cellulose particles immersed in water.
 785 *Atmospheric Chemistry & Physics*, 19(7), 4823-4849. <https://doi.org/10.5194/acp-19-4823-2019>
- 786 Hiranuma, N., Möhler, O., Bingemer, H., Bundke, U., Cziczo, D. J., Danielczok, A., et al. (2013). Immersion
 787 freezing of clay minerals and bacterial ice nuclei. *AIP Conference Proceedings*, 1527(1), 914-917.
 788 <https://doi.org/10.1063/1.4803420>
- 789 Hiranuma, N., Möhler, O., Yamashita, K., Tajiri, T., Saito, A., Kiselev, A., et al. (2015). Ice nucleation by cellulose
 790 and its potential contribution to ice formation in clouds. *Nature Geoscience*, 8(4), 273-277.
 791 <https://doi.org/10.1038/ngeo2374>
- 792 Hoose, C., & Möhler, O. (2012). Heterogeneous ice nucleation on atmospheric aerosols: A review of results from
 793 laboratory experiments. *Atmospheric Chemistry & Physics*, 12(20), 9817-9854. <https://doi.org/10.5194/acp-12-9817-2012>
- 794 Hu, W., Murata, K., Toyonaga, S., & Zhang, D. (2017). Bacterial abundance and viability in rainwater associated
 795 with cyclones, stationary fronts and typhoons in southwestern Japan. *Atmospheric Environment*, 167, 104-115. <https://doi.org/10.1016/j.atmosenv.2017.08.013>
- 796 Huang, Q., Niu, S., Lü, J., Zhou, Y., & Zhang, X. (2018). Physical characteristics of freezing raindrop size
 797 distribution and terminal velocity in two ice weather cases in Lushan area. *Chinese Journal of Atmospheric
 798 Sciences*, 42(5), 1023-1037. <http://d.old.wanfangdata.com.cn/Periodical/daqikx201805006>
- 800 Huang, S., Hu, W., Chen, J., Wu, Z., Zhang, D., & Fu, P. (2021). Overview of biological ice nucleating particles in
 801 the atmosphere. *Environment International*, 146, 106197. <https://doi.org/10.1016/j.envint.2020.106197>
- 803 Huffman, J. A., Prenni, A. J., DeMott, P. J., Pöhlker, C., Mason, R. H., Robinson, N. H., et al. (2013). High
 804 concentrations of biological aerosol particles and ice nuclei during and after rain. *Atmospheric Chemistry &
 805 Physics*, 13(13), 6151-6164. <https://doi.org/10.5194/acp-13-6151-2013>
- 806 Iannone, R., Chernoff, D. I., Pringle, A., Martin, S. T., & Bertram, A. K. (2011). The ice nucleation ability of one of
 807 the most abundant types of fungal spores found in the atmosphere. *Atmospheric Chemistry & Physics*,
 808 11(3), 1191-1201. <https://doi.org/10.5194/acp-11-1191-2011>
- 809 Joly, M., Amato, P., Deguillaume, L., Monier, M., Hoose, C., & Delort, A. M. (2014). Quantification of ice nuclei
 810 active at near 0 °C temperatures in low-altitude clouds at the Puy de Dôme atmospheric station.
 811 *Atmospheric Chemistry & Physics*, 14(15), 8185-8195. <https://doi.org/10.5194/acp-14-8185-2014>
- 812 Joyce, R. E., Lavender, H., Farrar, J., Werth, J. T., Weber, C. F., D'Andrilli, J., et al. (2019). Biological Ice-
 813 Nucleating Particles Deposited Year-Round in Subtropical Precipitation. *Applied and Environmental
 814 Microbiology*, 85(23), e01567-01519. <https://doi.org/10.1128/aem.01567-19>
- 815 Kanji, Z. A., Ladino, L. A., Wex, H., Boose, Y., Burkert-Kohn, M., Cziczo, D. J., & Krämer, M. (2017). Overview
 816 of ice nucleating particles. *Meteorological Monographs*, 58, 1.1–1.33.
 817 <https://doi.org/10.1175/AMSMONOGRAPH-D-16-0006.1>
- 818 Karimi, B., Nosrati, R., Fazly, B., Mirpour, M., Malboobi, M. A., & Owlia, P. (2019). A comparative evaluation of
 819 freezing criteria and molecular characterization of epiphytic ice-nucleating (Ice +) and non-ice-nucleating
 820 (Ice -) *Pseudomonas syringae* and *Pseudomonas fluorescens*. *Journal of Plant Pathology*, 102, 1-10.
 821 <https://doi.org/10.1007/s42161-019-00402-7>
- 822 Kieft, T. L., & Ruscetti, T. (1990). Characterization of biological ice nuclei from a lichen. *Journal of Bacteriology*,
 823 172(6), 3519-3523. <https://doi.org/10.1128/jb.172.6.3519-3523.1990>
- 824 Kilchhofer, K., Mahrt, F., & Kanji, Z. A. (2021). The role of cloud processing for the ice nucleating ability of
 825 organic aerosol and coal fly ash particles. *Journal of Geophysical Research: Atmospheres*, 126(10),
 826 e2020JD033338. <https://doi.org/10.1029/2020JD033338>
- 827 Kim, H. K., Orser, C., Lindow, S. E., & Sands, D. C. (1987). *Xanthomonas campestris* pv. *translucens* strains active
 828 in ice nucleation. *Plant Disease*, 71(11), 994-997. <https://doi.org/10.1094/pd-71-0994>
- 829 Kim, S., Kaplan, L. A., Benner, R., & Hatcher, P. G. (2004). Hydrogen-deficient molecules in natural riverine water
 830 samples—Evidence for the existence of black carbon in DOM. *Marine Chemistry*, 92(1), 225-234.
 831 <https://doi.org/10.1016/j.marchem.2004.06.042>
- 832 Klemm, O., & Wrzesinsky, T. (2007). Fog deposition fluxes of water and ions to a mountainous site in Central
 833 Europe. *Tellus B: Chemical and Physical Meteorology*, 59(4), 705-714. <https://doi.org/10.1111/j.1600-0889.2007.00287.x>

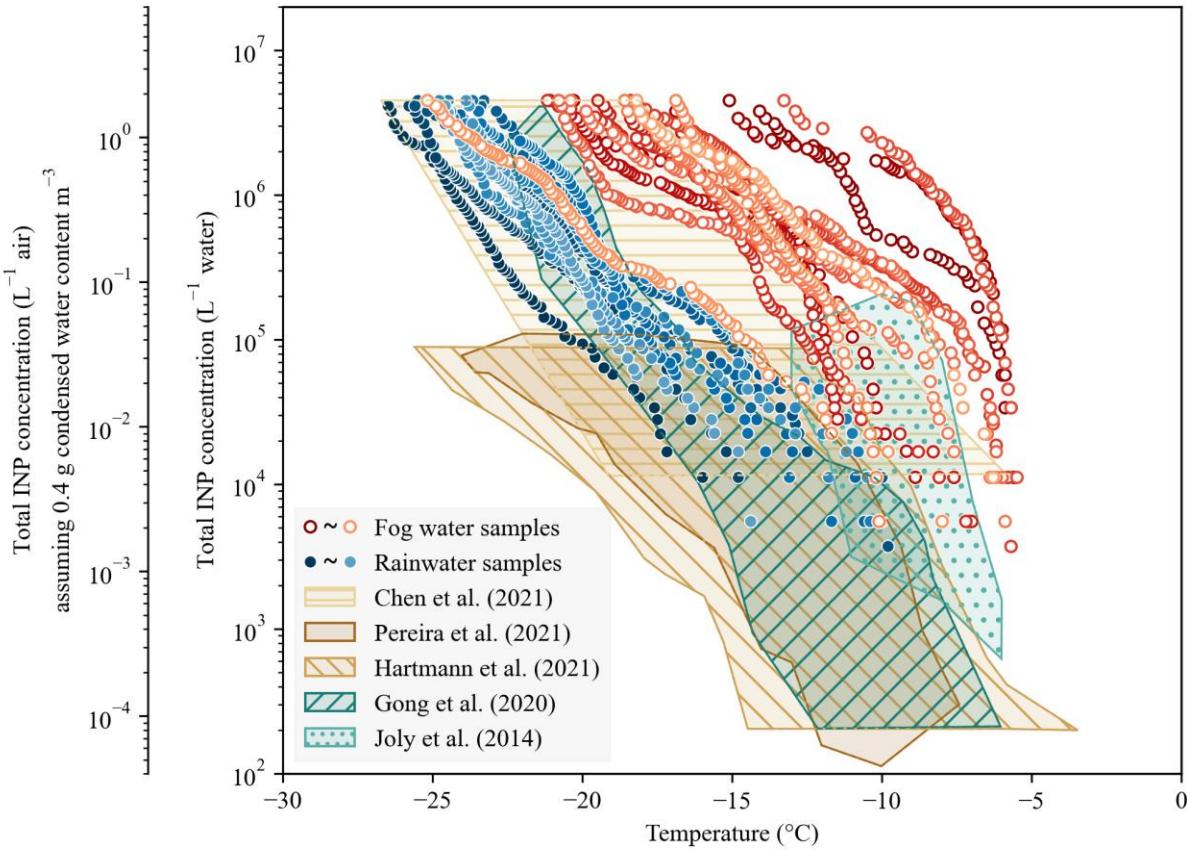
- 835 Klumpp, K., Marcolli, C., & Peter, T. (2022). The impact of (bio-)organic substances on the ice nucleation activity
 836 of the K-feldspar microcline in aqueous solutions. *Atmospheric Chemistry & Physics*, 22(5), 3655-3673.
 837 <https://doi.org/10.5194/acp-22-3655-2022>
- 838 Knopf, D. A., Alpert, P. A., & Wang, B. (2018). The role of organic aerosol in atmospheric ice nucleation: A
 839 review. *ACS Earth and Space Chemistry*, 2(3), 168-202.
 840 <https://doi.org/10.1021/acsearthspacechem.7b00120>
- 841 Kobashigawa, Y., Nishimiya, Y., Miura, K., Ohgiya, S., Miura, A., & Tsuda, S. (2005). A part of ice nucleation
 842 protein exhibits the ice-binding ability. *FEBS Letters*, 579(6), 1493-1497.
 843 <https://doi.org/10.1016/j.febslet.2005.01.056>
- 844 Koch, B. P., Dittmar, T., Witt, M., & Kattner, G. (2007). Fundamentals of molecular formula assignment to
 845 ultrahigh resolution mass data of natural organic matter. *Analytical Chemistry*, 79(4), 1758-1763.
 846 <https://doi.org/10.1021/ac061949s>
- 847 Koop, T., & Zobrist, B. (2009). Parameterizations for ice nucleation in biological and atmospheric systems. *Physical
 848 Chemistry Chemical Physics*, 11(46), 10839-10850. <https://doi.org/10.1039/B914289D>
- 849 Krog, J. O., Zachariassen, K. E., Larsen, B., & Smidsrød, O. (1979). Thermal buffering in Afro-alpine plants due to
 850 nucleating agent-induced water freezing. *Nature*, 282, 300-301. <https://doi.org/10.1038/282300a0>
- 851 Kunert, A. T., Pöhlker, M. L., Tang, K., Krevert, C. S., Wieder, C., Speth, K. R., et al. (2019). Macromolecular
 852 fungal ice nuclei in *Fusarium*: Effects of physical and chemical processing. *Biogeosciences*, 16(23), 4647-
 853 4659. <https://doi.org/10.5194/bg-16-4647-2019>
- 854 Lacher, L., Steinbacher, M., Bukowiecki, N., Herrmann, E., Zipori, A., & Kanji, Z. A. (2018). Impact of Air Mass
 855 Conditions and Aerosol Properties on Ice Nucleating Particle Concentrations at the High Altitude Research
 856 Station Jungfraujoch. *Atmosphere*, 9(9), 363. <https://doi.org/10.3390/atmos9090363>
- 857 Li, P., Wang, Y., Li, T., Sun, L., Yi, X., Guo, L., & Su, R. (2015). Characterization of carbonaceous aerosols at
 858 Mount Lu in South China: implication for secondary organic carbon formation and long-range transport.
 859 *Environmental Science and Pollution Research*, 22(18), 14189-14199. <https://doi.org/10.1007/s11356-015-4654-9>
- 860 Lim, Y. B., Tan, Y., Perri, M. J., Seitzinger, S. P., & Turpin, B. J. (2010). Aqueous chemistry and its role in
 861 secondary organic aerosol (SOA) formation. *Atmospheric Chemistry & Physics*, 10(21), 10521-10539.
 862 <https://doi.org/10.5194/acp-10-10521-2010>
- 863 Lindow, S., Lahue, E., Govindarajan, A., Panopoulos, N., & Gies, D. (1989). Localization of ice nucleation activity
 864 and the iceC gene product in *Pseudomonas syringae* and *Escherichia coli*. *Molecular Plant-Microbe
 865 Interact*, 2(5), 262-272. <https://doi.org/10.1094/MPMI-2-262>
- 866 Lohmann, U., & Feichter, J. (2005). Global indirect aerosol effects: A review. *Atmospheric Chemistry & Physics*,
 867 5(3), 715-737. <https://doi.org/10.5194/acp-5-715-2005>
- 868 Maki, L. R., Galyan, E. L., Chang-Chien, M.-M., & Caldwell, D. R. (1974). Ice nucleation induced by *Pseudomonas
 869 syringae*. *Applied Microbiology*, 28(3), 456-459. <https://doi.org/10.1128/am.28.3.456-459.1974>
- 870 Martin, A. C., Cornwell, G., Beall, C. M., Cannon, F., Reilly, S., Schaap, B., et al. (2019). Contrasting local and
 871 long-range-transported warm ice-nucleating particles during an atmospheric river in coastal California,
 872 USA. *Atmospheric Chemistry & Physics*, 19(7), 4193-4210. <https://doi.org/10.5194/acp-19-4193-2019>
- 873 Masschalck, B., & Michiels, C. W. (2003). Antimicrobial properties of lysozyme in relation to foodborne vegetative
 874 bacteria. *Critical Reviews in Microbiology*, 29(3), 191-214. <https://doi.org/10.1080/713610448>
- 875 Michaud, A. B., Dore, J. E., Leslie, D., Lyons, W. B., Sands, D. C., & Priscu, J. C. (2014). Biological ice nucleation
 876 initiates hailstone formation. *Journal of Geophysical Research: Atmospheres*, 119(21), 12186-12197.
 877 <https://doi.org/10.1002/2014JD022004>
- 878 Miller, A. J., Brennan, K. P., Mignani, C., Wieder, J., David, R. O., & Borduas-Dedekind, N. (2021). Development
 879 of the drop Freezing Ice Nuclei Counter (FINC), intercomparison of droplet freezing techniques, and use of
 880 soluble lignin as an atmospheric ice nucleation standard. *Atmospheric Measurement Techniques*, 14(4),
 881 3131-3151. <https://doi.org/10.5194/amt-14-3131-2021>
- 882 Möhler, O., Benz, S., Saathoff, H., Schnaiter, M., Wagner, R., Schneider, J., et al. (2008). The effect of organic
 883 coating on the heterogeneous ice nucleation efficiency of mineral dust aerosols. *Environmental Research
 884 Letters*, 3(2), 025007. <https://doi.org/10.1088/1748-9326/3/2/025007>
- 885 Morris, C. E., Georgakopoulos, D. G., & Sands, D. C. (2004). *Ice nucleation active bacteria and their potential role
 886 in precipitation*. Paper presented at the Journal de Physique IV France.
- 887 Morris, C. E., Sands, D. C., Glaux, C., Samsatly, J., Asaad, S., Moukahel, A. R., et al. (2013). Urediospores of rust
 888 fungi are ice nucleation active at $> -10^{\circ}\text{C}$ and harbor ice nucleation active bacteria. *Atmospheric
 889 Chemistry & Physics*, 13(8), 4223-4233. <https://doi.org/10.5194/acp-13-4223-2013>

- 891 Mortazavi, R., Hayes, C. T., & Ariya, P. A. (2008). Ice nucleation activity of bacteria isolated from snow compared
 892 with organic and inorganic substrates. *Environmental Chemistry*, 5(6), 373-381.
 893 <https://doi.org/10.1071/EN08055>
- 894 Murray, E., Murray, B., & Sivakumar, V. (2013). Ice nucleation by particles immersed in supercooled cloud
 895 droplets. *Chemical Society Reviews*, 42, 9571-9572. <https://doi.org/10.1039/C2CS35200A>
- 896 Niu, M., Hu, W., & Fu, P. (2024). Deciphering the significant role of biological ice nucleators in precipitation at the
 897 organic molecular level [Dataset]. *Zenodo*. <https://doi.org/10.5281/zenodo.10300240>
- 898 O'Sullivan, D., Murray, B. J., Malkin, T. L., Whale, T. F., Umo, N. S., Atkinson, J. D., et al. (2014). Ice nucleation
 899 by fertile soil dusts: Relative importance of mineral and biogenic components. *Atmospheric Chemistry &*
 900 *Physics*, 14(4), 1853-1867. <https://doi.org/10.5194/acp-14-1853-2014>
- 901 O'Sullivan, D., Murray, B. J., Ross, J. F., & Webb, M. E. (2016). The adsorption of fungal ice-nucleating proteins on
 902 mineral dusts: A terrestrial reservoir of atmospheric ice-nucleating particles. *Atmospheric Chemistry &*
 903 *Physics*, 16(12), 7879-7887. <https://doi.org/10.5194/acp-16-7879-2016>
- 904 O'Sullivan, D., Murray, B. J., Ross, J. F., Whale, T. F., Price, H. C., Atkinson, J. D., et al. (2015). The relevance of
 905 nanoscale biological fragments for ice nucleation in clouds. *Scientific Reports*, 5(1), 8082.
 906 <https://doi.org/10.1038/srep08082>
- 907 Obata, H., Nakai, T., Tanishita, J., & Tokuyama, T. (1989). Identification of an ice-nucleating bacterium and its ice
 908 nucleation properties. *Journal of Fermentation and Bioengineering*, 67, 143-147.
 909 [https://doi.org/10.1016/0922-338X\(89\)90111-6](https://doi.org/10.1016/0922-338X(89)90111-6)
- 910 Obata, H., Saeki, Y., Tanishita, J., Tokuyama, T., Hori, H., & Higashi, Y. (1987). Studies on ice-nucleating
 911 microorganisms. Part I. Identification of an ice-nucleating bacterium KUIN-1 as *Pseudomonas fluorescens*
 912 and its ice nucleation properties. *Agricultural and Biological Chemistry*, 51, 1761-1766.
 913 [https://doi.org/10.1016/0922-338X\(89\)90111-6](https://doi.org/10.1016/0922-338X(89)90111-6)
- 914 Pereira, D. L., Silva, M. M., García, R., Raga, G. B., Alvarez-Ospina, H., Carabali, G., et al. (2021).
 915 Characterization of ice nucleating particles in rainwater, cloud water, and aerosol samples at two different
 916 tropical latitudes. *Atmospheric Research*, 250, 105356. <https://doi.org/10.1016/j.atmosres.2020.105356>
- 917 Petters, M. D., & Wright, T. P. (2015). Revisiting ice nucleation from precipitation samples. *Geophysical Research
 918 Letters*, 42(20), 8758-8766. <https://doi.org/10.1002/2015GL065733>
- 919 Pouleur, S., Richard, C., Martin, J. G., & Antoun, H. (1992). Ice Nucleation Activity in *Fusarium acuminatum* and
 920 *Fusarium avenaceum*. *Applied and Environmental Microbiology*, 58(9), 2960-2964.
 921 <https://doi.org/10.1128/aem.58.9.2960-2964.1992>
- 922 Pouzet, G., Peghaire, E., Aguès, M., Baray, J.-L., Conen, F., & Amato, P. (2017). Atmospheric processing and
 923 variability of biological ice nucleating particles in precipitation at Opme, France. *Atmosphere*, 8(11), 229.
 924 <https://doi.org/10.3390/atmos8110229>
- 925 Pruppacher, H. R., & Klett, J. D. (2010). *Micrometeorology of clouds and precipitation* (Vol. 18): Springer Dordrecht.
- 926 Pummer, B. G., Bauer, H., Bernardi, J., Bleicher, S., & Grothe, H. (2012). Suspended macromolecules are
 927 responsible for ice nucleation activity of birch and conifer pollen. *Atmospheric Chemistry & Physics*, 12(5),
 928 2541-2550. <https://doi.org/10.5194/acp-12-2541-2012>
- 929 Pummer, B. G., Budke, C., Augustin-Bauditz, S., Niedermeier, D., Felgitsch, L., Kampf, C. J., et al. (2015). Ice
 930 nucleation by water-soluble macromolecules. *Atmospheric Chemistry & Physics*, 15(8), 4077-4091.
 931 <https://doi.org/10.5194/acp-15-4077-2015>
- 932 R-Core-Team. (2012). R: A Language and Environment for Statistical Computing [Software]. *R Foundation for
 933 Statistical Computing*. <http://www.r-project.org/>
- 934 Rederstorff, E., Fatimi, A., Sinquin, C., Ratiskol, J., Merceron, C., Vinatier, C., et al. (2011). Sterilization of
 935 exopolysaccharides produced by deep-sea bacteria: Impact on their stability and degradation. *Marine
 936 Drugs*, 9(2), 224-241. <https://doi.org/10.3390/md9020224>
- 937 Repaske, R. (1956). Lysis of gram-negative bacteria by lysozyme. *Biochimica et Biophysica Acta*, 22(1), 189-191.
 938 [https://doi.org/10.1016/0006-3002\(56\)90240-2](https://doi.org/10.1016/0006-3002(56)90240-2)
- 939 Rice, C. V., Middaugh, A., Wickham, J. R., Friedline, A., Thomas, K. J., Scull, E., et al. (2015). Bacterial
 940 lipoteichoic acid enhances cryosurvival. *Extremophiles*, 19(2), 297-305. [https://doi.org/10.1007/s00792-014-0714-1](https://doi.org/10.1007/s00792-

 941 014-0714-1)
- 942 Rodríguez Zafra, J. M., de Cara García, M., Tello Marquina, J., & Palmero Llamas, D. (2016). Dispersal of
 943 *Fusarium* spp. by rainwater and pathogenicity on four plant species. *Aerobiologia*, 32(3), 431-439.
 944 <https://doi.org/10.1007/s10453-015-9416-0>

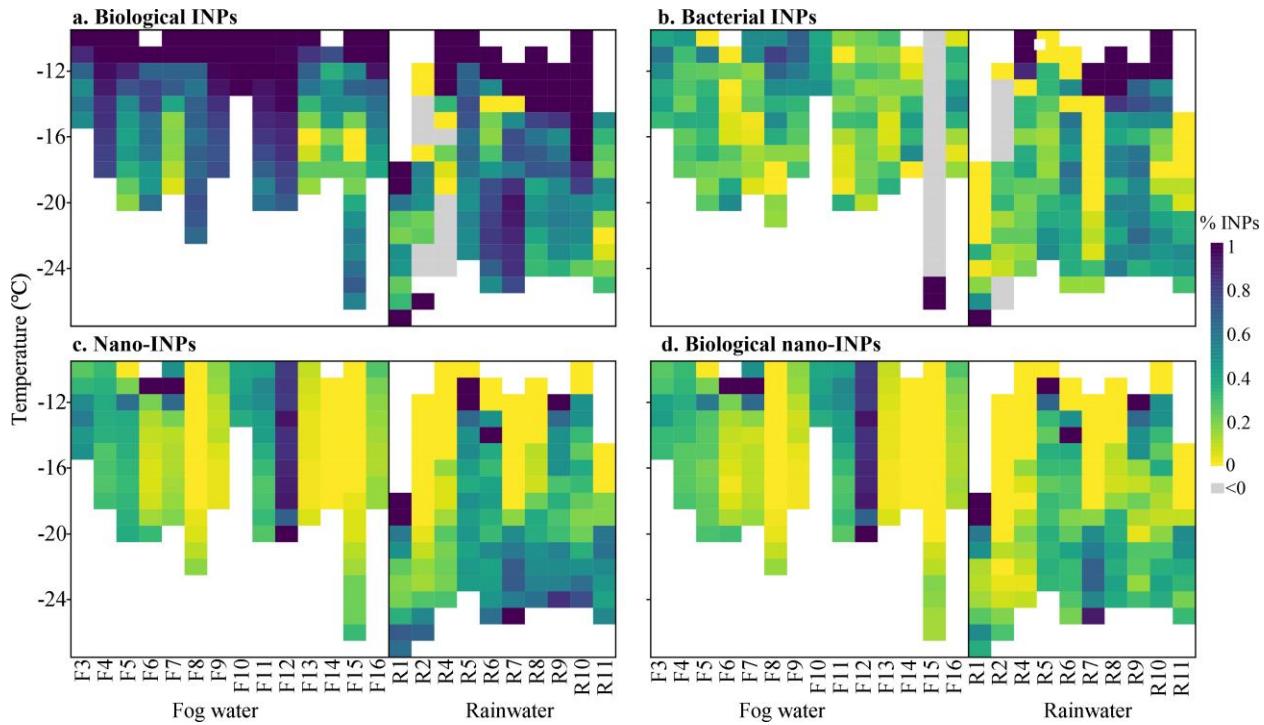
- 945 Šantl-Temkiv, T., Lange, R., Beddows, D., Rauter, U., Pilgaard, S., Dall’Osto, M., et al. (2019). Biogenic sources of
 946 ice nucleating particles at the High Arctic site villum research station. *Environmental Science &*
 947 *Technology*, 53(18), 10580-10590. <https://doi.org/10.1021/acs.est.9b00991>
- 948 Šantl-Temkiv, T., Sahyoun, M., Finster, K., Hartmann, S., Augustin-Bauditz, S., Stratmann, F., et al. (2015).
 949 Characterization of airborne ice-nucleation-active bacteria and bacterial fragments. *Atmospheric*
 950 *Environment*, 109, 105-117. <https://doi.org/10.1016/j.atmosenv.2015.02.060>
- 951 Schmieder, R., & Edwards, R. (2011). Quality control and preprocessing of metagenomic datasets. *Bioinformatics*,
 952 27(6), 863-864. <https://doi.org/10.1093/bioinformatics/btr026>
- 953 Schnell, R. C. (1977). Ice nuclei in seawater, fog water and marine air off the coast of Nova Scotia: Summer 1975.
 954 *Journal of Atmospheric Sciences*, 34(8), 1299-1305. [https://doi.org/10.1175/1520-0469\(1977\)034<1299:INISFW>2.0.CO;2](https://doi.org/10.1175/1520-0469(1977)034<1299:INISFW>2.0.CO;2)
- 955 Schnell, R. C., & Vali, G. (1973). World-wide source of leaf-derived freezing nuclei. *Nature*, 246(5430), 212-213.
 956 <https://doi.org/10.1038/246212a0>
- 957 Schnell, R. C., & Vali, G. (1976). Biogenic ice nuclei: Part I. Terrestrial and marine sources. *Journal of Atmospheric*
 958 *Sciences*, 33(8), 1554-1564. [https://doi.org/10.1175/1520-0469\(1976\)033<1554:Binpit>2.0.Co;2](https://doi.org/10.1175/1520-0469(1976)033<1554:Binpit>2.0.Co;2)
- 959 Steiner, A. L., Brooks, S. D., Deng, C., Thornton, D. C. O., Pendleton, M. W., & Bryant, V. (2015). Pollen as
 960 atmospheric cloud condensation nuclei. *Geophysical Research Letters*, 42(9), 3596-3602.
 961 <https://doi.org/10.1002/2015GL064060>
- 962 Steinke, I., Hiranuma, N., Funk, R., Höhler, K., Tüllmann, N., Umo, N. S., et al. (2020). Complex plant-derived
 963 organic aerosol as ice-nucleating particles – more than the sums of their parts? *Atmospheric Chemistry &*
 964 *Physics*, 20(19), 11387-11397. <https://doi.org/10.5194/acp-20-11387-2020>
- 965 Stopelli, E., Conen, F., Guilbaud, C., Zopfi, J., Alewell, C., & Morris, C. E. (2017). Ice nucleators, bacterial cells
 966 and *Pseudomonas syringae* in precipitation at Jungfraujoch. *Biogeosciences*, 14(5), 1189-1196.
 967 <https://doi.org/10.5194/bg-14-1189-2017>
- 968 Stopelli, E., Conen, F., Morris, C. E., Herrmann, E., Henne, S., Steinbacher, M., & Alewell, C. (2016). Predicting
 969 abundance and variability of ice nucleating particles in precipitation at the high-altitude observatory
 970 Jungfraujoch. *Atmospheric Chemistry & Physics*, 16(13), 8341-8351. <https://doi.org/10.5194/acp-16-8341-2016>
- 971 Sun, X., Wang, Y., Li, H., Yang, X., Sun, L., Wang, X., et al. (2016). Organic acids in cloud water and rainwater at
 972 a mountain site in acid rain areas of South China. *Environmental Science and Pollution Research*, 23(10),
 973 9529-9539. <https://doi.org/10.1007/s11356-016-6038-1>
- 974 Sze, K. C. H., Wex, H., Hartmann, M., Skov, H., Massling, A., Villanueva, D., & Stratmann, F. (2023). Ice-
 975 nucleating particles in northern Greenland: Annual cycles, biological contribution and parameterizations.
 976 *Atmospheric Chemistry & Physics*, 23(8), 4741-4761. <https://doi.org/10.5194/acp-23-4741-2023>
- 977 Tang, K., Sanchez-Parra, B., Yordanova, P., Wehking, J., Backes, A. T., Pickersgill, D. A., et al. (2022). Bioaerosols
 978 and atmospheric ice nuclei in a Mediterranean dryland: Community changes related to rainfall.
 979 *Biogeosciences*, 19(1), 71-91. <https://doi.org/10.5194/bg-19-71-2022>
- 980 Testa, B., Hill, T. C. J., Marsden, N. A., Barry, K. R., Hume, C. C., Bian, Q., et al. (2021). Ice nucleating particle
 981 connections to regional argentinian land surface emissions and weather during the cloud, aerosol, and
 982 complex terrain interactions experiment. *Journal of Geophysical Research: Atmospheres*, 126(23),
 983 e2021JD035186. <https://doi.org/10.1029/2021JD035186>
- 984 Tobo, Y., Adachi, K., DeMott, P. J., Hill, T. C. J., Hamilton, D. S., Mahowald, N. M., et al. (2019). Glacially
 985 sourced dust as a potentially significant source of ice nucleating particles. *Nature Geoscience*, 12(4), 253-
 986 258. <https://doi.org/10.1038/s41561-019-0314-x>
- 987 Tobo, Y., DeMott, P. J., Hill, T. C. J., Prenni, A. J., Swoboda-Colberg, N. G., Franc, G. D., & Kreidenweis, S. M.
 988 (2014). Organic matter matters for ice nuclei of agricultural soil origin. *Atmospheric Chemistry & Physics*,
 989 14(16), 8521-8531. <https://doi.org/10.5194/acp-14-8521-2014>
- 990 Urano, M., & Douple, E. B. (2023). *Thermal Effects on Cells and Tissues* (Vol. 1). London: CRC Press.
- 991 Vali, G. (1971). Quantitative evaluation of experimental results on the heterogeneous freezing nucleation of
 992 supercooled liquids. *Journal of Atmospheric Sciences*, 28(3), 402-409. [https://doi.org/10.1175/1520-0469\(1971\)028<0402:QEOERA>2.0.CO;2](https://doi.org/10.1175/1520-0469(1971)028<0402:QEOERA>2.0.CO;2)
- 993 Vali, G., DeMott, P. J., Möhler, O., & Whale, T. F. (2015). Technical Note: A proposal for ice nucleation
 994 terminology. *Atmospheric Chemistry & Physics*, 15(18), 10263-10270. <https://doi.org/10.5194/acp-15-10263-2015>
- 995

- 999 Walters, K. R., Serianni, A. S., Sformo, T., Barnes, B. M., & Duman, J. G. (2009). A nonprotein thermal hysteresis-
1000 producing xylomannan antifreeze in the freeze-tolerant Alaskan beetle *Upis ceramboides*. *Proceedings of
1001 the National Academy of Sciences*, 106(48), 20210-20215. <https://doi.org/10.1073/pnas.0909872106>
- 1002 Wang, B., & Knopf, D. A. (2011). Heterogeneous ice nucleation on particles composed of humic-like substances
1003 impacted by O₃. *Journal of Geophysical Research: Atmospheres*, 116(D3), D03205.
1004 <https://doi.org/10.1029/2010JD014964>
- 1005 Watabe, S., Abe, K., Hirata, A., Emori, Y., Watanabe, M., & Arai, S. (1993). Large-scale Production and
1006 Purification of an *Erwinia ananas* Ice Nucleation Protein and Evaluation of Its Ice Nucleation Activity.
1007 *Bioscience, Biotechnology, and Biochemistry*, 57(4), 603-606. <https://doi.org/10.1271/bbb.57.603>
- 1008 Wilson, T. W., Ladino, L. A., Alpert, P. A., Breckels, M. N., Brooks, I. M., Browne, J., et al. (2015). A marine
1009 biogenic source of atmospheric ice-nucleating particles. *Nature*, 525(7568), 234-238.
1010 <https://doi.org/10.1038/nature14986>
- 1011 Wise, M. E., Baustian, K. J., Koop, T., Freedman, M. A., Jensen, E. J., & Tolbert, M. A. (2012). Depositional ice
1012 nucleation onto crystalline hydrated NaCl particles: A new mechanism for ice formation in the troposphere.
1013 *Atmospheric Chemistry & Physics*, 12(2), 1121-1134. <https://doi.org/10.5194/acp-12-1121-2012>
- 1014 Wolber, P., & Warren, G. (1989). Bacterial ice-nucleation proteins. *Trends in Biochemical Sciences*, 14(5), 179-182.
1015 [https://doi.org/10.1016/0968-0004\(89\)90270-3](https://doi.org/10.1016/0968-0004(89)90270-3)
- 1016 Xu, H., Griffith, M., Patten, C. L., & Glick, B. R. (1998). Isolation and characterization of an antifreeze protein with
1017 ice nucleation activity from the plant growth promoting rhizobacterium *Pseudomonas putida* GR12-2.
1018 *Canadian Journal of Microbiology*, 44(1), 64-73. <https://doi.org/10.1139/w97-126>
- 1019 Yamashita, Y., Kawahara, H., & Obata, H. (2002). Identification of a novel anti-ice-nucleating polysaccharide from
1020 *Bacillus thuringiensis* YY529. *Bioscience, Biotechnology, and Biochemistry*, 66(5), 948-954.
1021 <https://doi.org/10.1271/bbb.66.948>
- 1022 Yang, L., Chen, W., Zhuang, W.-E., Cheng, Q., Li, W., Wang, H., et al. (2019). Characterization and bioavailability
1023 of rainwater dissolved organic matter at the southeast coast of China using absorption spectroscopy and
1024 fluorescence EEM-PARAFAC. *Estuarine, Coastal and Shelf Science*, 217, 45-55.
1025 <https://doi.org/10.1016/j.ecss.2018.11.002>
- 1026 Yanti, Y., Zainon, M. N., & Marshida, A. H. U. (2012, 23-26 Sept. 2012). *Antagonistic activity of three
1027 Actinomycetes, MG01, MG02 And KT2F towards Phellinus noxius*. Paper presented at the 2012 IEEE
1028 Symposium on Business, Engineering and Industrial Applications.
- 1029 Zhang, S., Du, R., Chen, H., Zhang, Y., Du, P., & Ren, W. (2020). Characteristics and Distribution of efficient ice
1030 nucleating particles in rainwater and soil. *Atmospheric Research*, 246, 105129.
1031 <https://doi.org/10.1016/j.atmosres.2020.105129>
- 1032 Zhou, L., Zhou, Y., Hu, Y., Cai, J., Liu, X., Bai, C., et al. (2019). Microbial production and consumption of
1033 dissolved organic matter in glacial ecosystems on the Tibetan Plateau. *Water Research*, 160, 18-28.
1034 <https://doi.org/10.1016/j.watres.2019.05.048>
- 1035



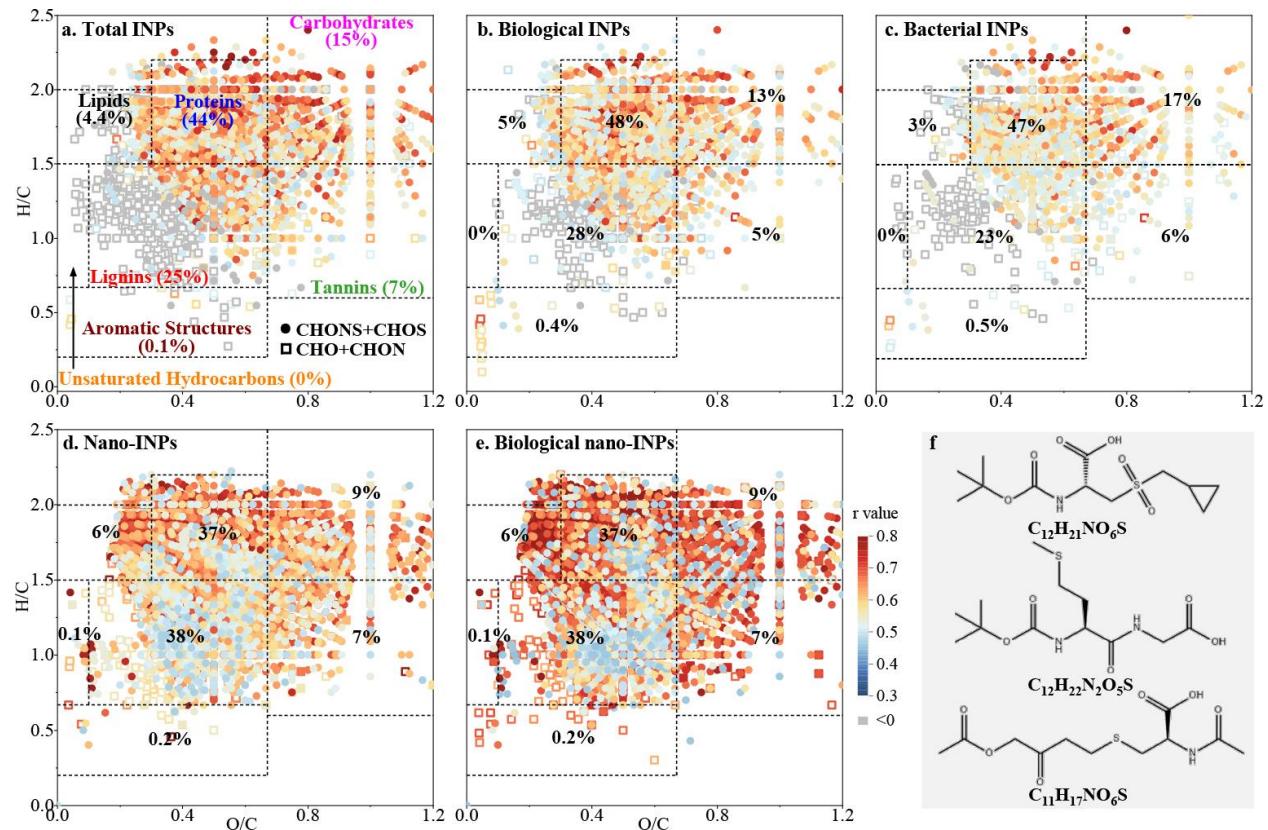
1036

1037 **Figure 1. Total concentrations of ice nucleating particles in precipitation samples.** The
 1038 cumulative ice nucleating particle (INP) spectra per unit volume of rainwater samples (blue dots)
 1039 and fog water samples (red dots) were performed by droplet freezing assays. The INP spectra per
 1040 volume of air were calculated by assuming a cloud-condensed water content of 0.4 g m^{-3}
 1041 according to Chen, Wu, Wu, et al. (2021). The shaded area represents the total INP
 1042 concentrations in precipitation samples measured by other studies (Chen, Wu, Wu, et al., 2021;
 1043 Gong et al., 2020; Hartmann et al., 2021; Joly et al., 2014; Pereira et al., 2021).

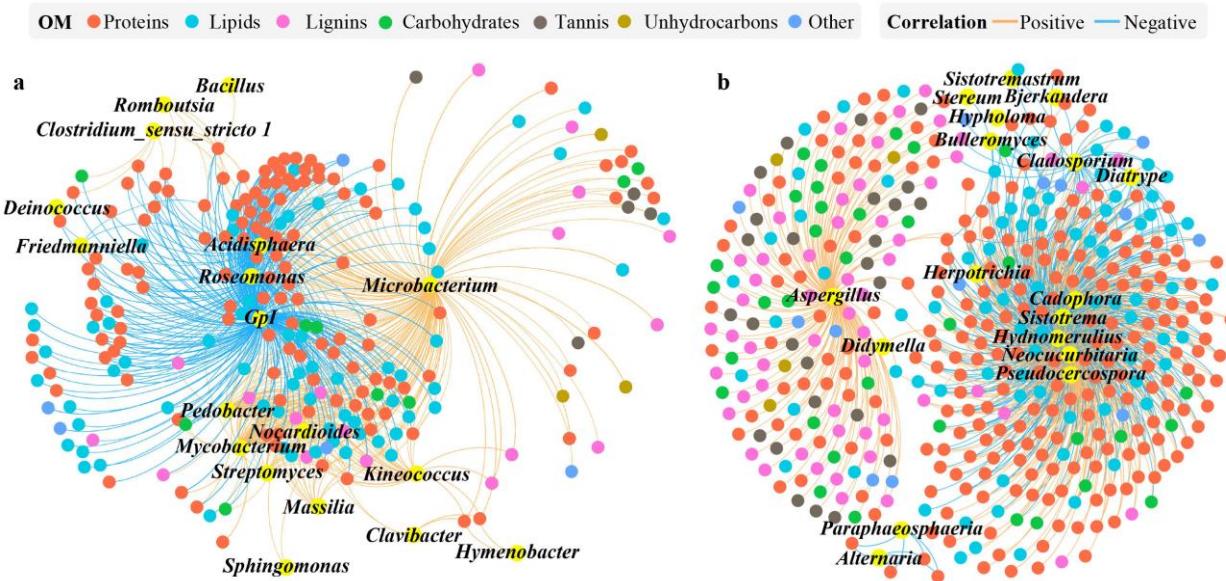


1044

1045 **Figure 2. Contribution of biological materials to total ice nucleating particles.** Variations in
 1046 the percentages of heat-sensitive ice nucleating particles (INPs) (a, regarded as biological INPs
 1047 in this study) and lysozyme-sensitive INPs (b, regarded as bacterial INPs) to total INPs were
 1048 observed at different temperatures. (c) and (d) represent the proportions of nanoscale INPs
 1049 (nano-INPs, INPs smaller than $0.22 \mu\text{m}$ herein) and heat-sensitive nano-INPs (biological nano-
 1050 INPs) to total INPs, respectively.



1051
Figure 3. Organic molecular composition associated with different types of ice nucleating
1052 **particles.** The organic molecules significantly correlated with the concentrations of total (a),
1053 biological (b) and bacterial (c) ice nucleating particles (INPs) at -18°C , and nanoscale INPs
1054 (nano-INPs) (d) and biological nano-INPs (e) at -20°C based on Spearman correlation analysis
1055 ($p < 0.05$) and their classification are presented by van Krevelen diagrams. The fractional
1056 contributions to the organic molecules are shown for each molecular class. The contributions of
1057 sulfur-containing compounds (CHOS and CHONS, circle) and non-sulfur-containing compounds
1058 (CHO and CHON, square) to total organic molecules correlated with different types of INPs are
1059 shown. **f** shows the possible structures of sulfur-containing compounds associated with INPs
1060 based on molecular formulas.
1061



1062

Figure 4. Association between microbial taxa and nanoscale ice nucleating particles-related organic matter. Co-occurrence networks of biological nanoscale ice nucleating particles-related organic matter (OM) were constructed based on Spearman's correlation analysis with bacterial (a) and fungal (b) genera, respectively. Only OM molecules significantly correlated with biological nanoscale ice nucleating particles (Spearman's correlation coefficient > 0.7) were applied for network construction. Microbial genera are labeled in yellow in networks.

Figure 1.

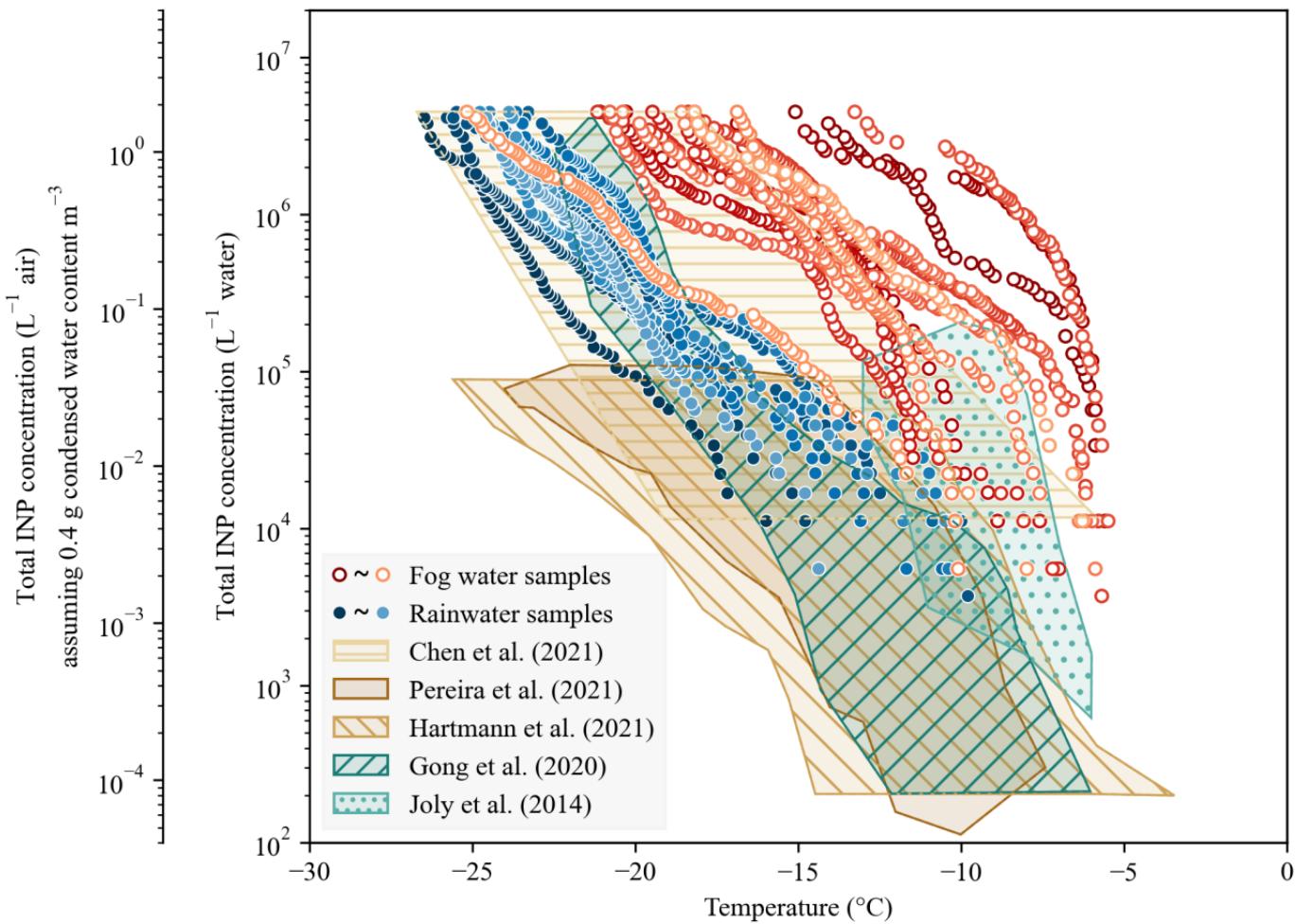


Figure 2.

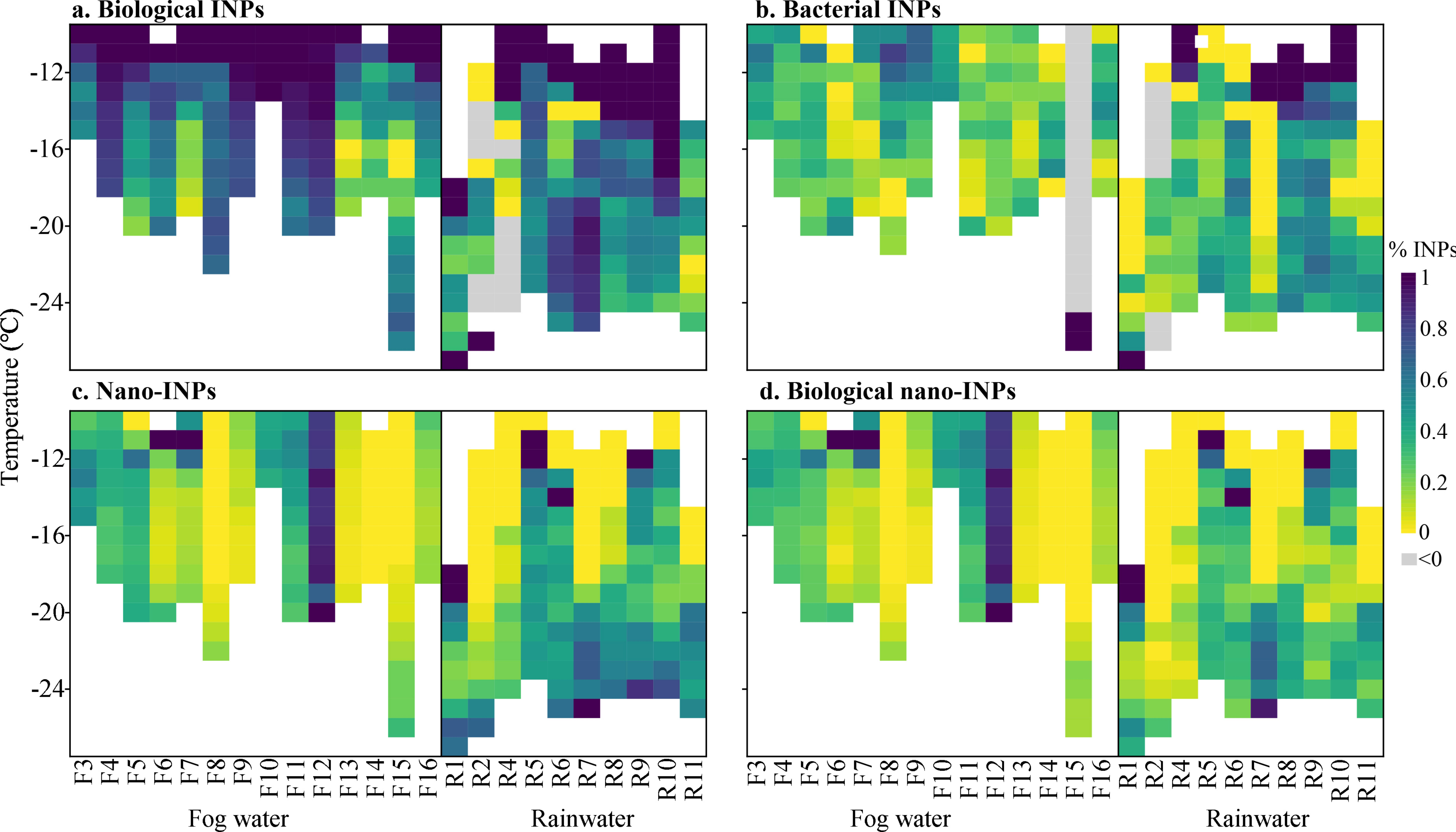


Figure 3.

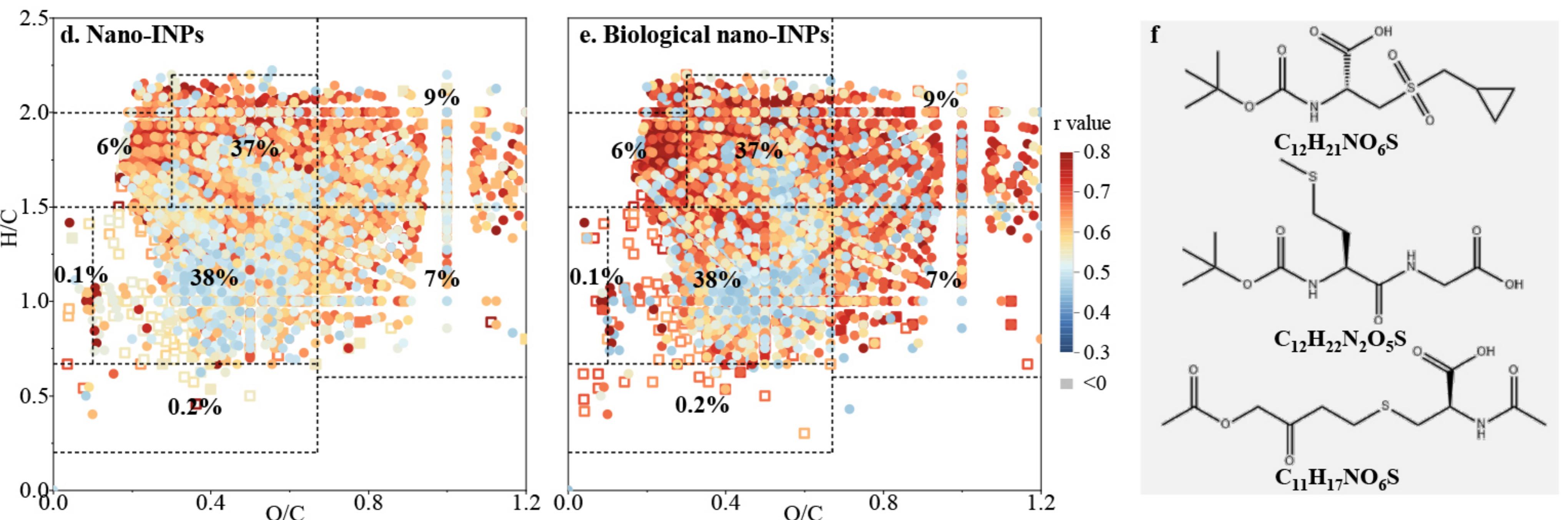
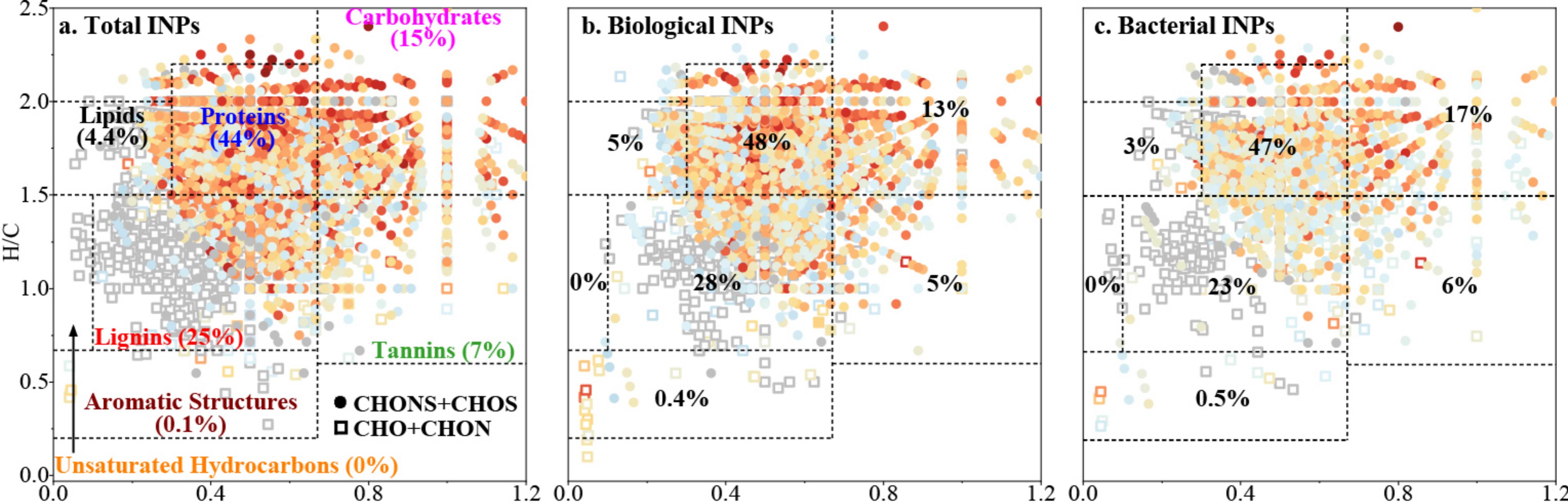


Figure 4.

OM ● Proteins ● Lipids ● Lignins ● Carbohydrates ● Tannins ● Unhydrocarbons ● Other Correlation — Positive — Negative

